Curriculum

Certificate/Diploma Level

in

Health Sciences

(General Medicine, Medical Laboratory Technology, Diagnostic Radiography, Homeopathy, Ayurveda, Amchi Science, Dental Science, Ophthalmic Science, Pharmacy, Physiotherapy and Acupuncture, Acupressure & Moxibustion)

(First year)



Council for Technical Education and Vocational Training
Curriculum Development Division

Sanothimi, Bhaktapur

Revision on July 2016

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1. Program Description

This first year curriculum is designed for all health science programme of Diploma/Certificate level (General Medicine, Medical Laboratory Technology, Radiography, Homeopathy, Ayurveda, Amchi Science, Dental Science, Opthalmic Science, Pharmacy and Ocupuncture, ocupressure & Moxisbuston) except PCL Nursing. In this curriculum foundational subjects such as Mathematics, English, Nepali, Social Study and Basic Sciences (Chemistry, Physics, Botany & Zoology) are offered to built the base. The disciplinary subject 'Anatomy and physiology' is included in order to lure them to their respective field. Most of the subjects offered in first year are of theoretical nature with some lab practices in basic science, computer and anatomy & physiology. It has the computer part in mathematics that demands practice in computer lab. Anatomy and physiology introduces the sector and guides the students to their specified medical field. The curriculum structure and the content reflects the details of all first year subjects. Academic requirements to enter bachelor in health sciences is considered while designed this first year course.

2. Target Location:

The target location will be all over Nepal.

3. Entry criteria

- SLC Pass upto 2071 SLC or SLC with GPA 2.00 plus minimum C grade in Compulsory Mathematics, English & Science after letter grading.
- TSLC in relevant decipline with minimum 66.68%.
- Should pass entrance examination as administered by CTEVT.

4. Selection:

Applicants fulfilling the entry criteria will be selected for admission on the basis of merit.

5. Medium of Instruction:

The medium of instruction will be in English and/or Nepali.

6. Pattern of Attendance:

Minimum of 90% attendance in each subject is required to appear in the respective final examination.

7. Teacher and Student Ratio

- Overall ratio of teacher and student must be 1:10 (at the institution level)
- For theory: As per the nature of the course
- For practical/lab/demonstration: 1:10

8. Program Coordinator, Teachers and Demonstrators:

- The foundational subject related teacher should be master degree holder in the related area.
- The disciplinary subject related teacher should be a bachelor's degree holder in the related area.
- The demonstrators should be bachelor's degree holder in the related area with two years experiences in training activities

9. Instructional Media and Materials:

The following instructional media and materials are suggested for the effective instruction and demonstration.

- **Printed Media Materials (**assignment sheets, handouts, information sheets, individual training packets, performance checklists, textbooks etc.).
- Non-projected Media Materials (display, models, flip chart, poster, writing board etc.).
- Projected Media Materials (opaque projections, overhead transparencies, slides etc.).
- Audio-Visual Materials (audiotapes, films, slide-tape programmes, videodiscs, videotapes etc.).
- Computer-Based Instructional Materials (computer-based training, interactive video etc.).

10. Teaching Learning Methodologies:

The methods of teachings for this curricular programme will be a combination of several approaches (not limited to as mentioned here) such as illustrated lecture, tutorial, group discussion, demonstration, simulation, guided practice, practical experiences, report writing, term paper presentation, project work and other independent learning.

Theory: Lecture, discussion, interaction, assignment, group work. **Practical:** Demonstration, observation, guided practice, self-practice, project work, etc.

11. Mode of Education:

There will be inductive and deductive mode of education.

12. Examination and Marking Scheme:

- The subject teacher will internally assess the students' achievement in each subject during the course followed by a final examination at the end of the year.
- A weightage of internal assessment and annual examination are allocated in the course structer of this curriculum.
- The final examinations of all theory part will be administered through written tests.
- For theory exam, short and long questions will be asked covering all units of subjects as far as possible.
- The method of continuous assessment will be adopted for practical components. Final practicum evaluation will be based on:
 - a. Institutional practicum attendance 10%
 - b. Logbook/Practicum book maintenance 10%
 - c. Spot performance (assigned task/practicum performance/identification/arrangement preparation/measurement) 40%
 - d. Viva voce : Internal examiner 20%

External examiner - 20%

• Student who fails in the internal assessment of any subject will not be allowed to sit in the final examination of that subject.

13. Provision of Back Paper:

There will be the provision of back paper but a student must pass all the subjects within six years from the enrollment date.

14. Disciplinary and Ethical Requirements:

- Intoxication, insubordination or rudeness to peers will result in immediate suspension followed by the review of the disciplinary review committee of the institute.
- Dishonesty in academic or practical activities will result in immediate suspension followed by administrative review, with possible expulsion.
- Illicit drug use, bearing arms in institute, threats or assaults to peers, faculty or staff will result in immediate suspension, followed by administrative review with possible expulsion.

15. Pass Marks:

The students must secure minimum of 40% marks in theory and 50% marks in practical. Moreover, the students must secure minimum pass marks in the internal assessment and final examination of each subject of theory and practical separately to pass all subjects offered

Course Structure

First year

				Maa				Distribut	ion of Marl	s			
		Mode Theory		Practical									
SN	Subject	Activity	т	Р	Total	Internal	Final	Exam Hour	Internal	Final	Minimum Exam Hour	Total Marks	Remarks
1	English	Т	3	0	3	20	80	3	-	-	-	100	
2	Nepali	Т	3	0	3	20	80	3	-	-	-	100	
3	Social Studies	Т	2	0	2	10	40	1.5	-	-	-	50	
4	Anatomy & Physiology	T+P	4	1	5	20	60	3	10	10	3	100	
5	Physics	T+P	4	2	6	20	60	3	10	10	3	100	
6	Chemistry	T+P	4	2	6	20	60	3	10	10	3	100	
7	Zoology	T+P	3	2	5	20	60	3	10	10	3	100	
8	Botany	T+P	3	2	5	20	60	3	10	10	3	100	
9	Mathematics & Statistics	T+P	4	1	5	20	60	3	10	10	3	100	
	Total		30	10	40	170	560		60	60		850	

English

3

2 8

Program	Health Science	Total Hours:
Year	First	Weekly Hrs:
Level	Certificate	Theory:
		Internal Assessment:
		Final Assessment:

Course Description

This is an integrated general English course, which treats English as a medium for communication a means to knowledge and skill related to health. It provides a remedial refresher course includin English grammar and structures and use of a dictionary, tools for receiving and imparting infor effectively, and exposure to poems, essays and stories which are interesting and informative topics of interest. This course provides a bridge between secondary and university English.

Course Objectives

On completion of the course student will be enabled to:

- Use English for academic and communicative purposes.
- Demonstrate functional, notional and grammatical skill in English language usage.
- UseEnglish structures in informal communication.
- Analyze the prescribed texts related to different literary genres.
- Answer the questions based on the reading texts.
- Produce different types of free compositions

Contents:

Part 1: Grammar

Unit 1: Link English	Theory Time Hrs. 10
Objectives:	Contents
 Use English dictionary appropriately Differentiate American and British English spelling Enrich English vocabulary Form English sentences correctly 	 Dictionary Skills: Alphabetic order, dictionary quarter system, guide words, head words etc. British and American English: spelling differences Word formation process through affixes (prefix and su vocabulary Sentence formation
Unit 2: Comparison	Theory Time Hrs. 5
Objectives	Contents
• Apply the structures for making comparisons using adjectives and adverbs	 Comparatives and superlatives forms of Adjectives Comparative and superlatives and there uses Other ways of comparing things
Unit 3: Prepositions	Theory Time Hrs. 5
Objectives	Contents
• Apply the prepositions 'in', 'on' and 'at' in different contexts.	 Prepositions of Place: on, in, at Prepositions of Time: on, in, at Prepositions with forms of transport
Unit 4: Tenses	Theory Time Hrs. 8
Objectives	Contents
 Use present tenses, past tenses and perfect tenses in different situations. Talk about the future using 'will' and 'going to' 	 Auxiliary verbs: be, have, do The Present Tenses The past tenses The perfect tense

• Talk about the future using present	• Talking about the present tense
tense	• Talking about the past
	• Reporting the past
	• Talking about the future using 'will' and 'going to'
	Talking about the future using present tense
Unit 5: Mood	Theory Time Hrs. 7
Objectives	Contents
• Apply the structures for making	Questions
yes/no questions beginning with	• Wh – words
auxiliary or modal.	Question tags- forms
Use Question tags	• Question tags – uses
• Use indirect questions to ask for	Indirect and reported questions
information or help.	• Negative sentence with "not"
• Use negative sentence with "not"	
Unit 6: Modals	Theory Time Hrs. 7
Objectives	Contents
Introduce modals	Instructions to modals
• Use modals for probability, certainty,	Modal negative and questions
permission, instructions, request and	 Can/ could, may/ might- possibility
suggestions.	 Cannot, can't, must, ought, should, will- probability and
	certainty.
	 Can, could, may – permission
	 Can/could, will/ would – Instructions and requests
	 Can/could, with would instructions and requests Can/could, might, shall - suggestions
Unit 7: The Passive	Theory Time Hrs. 4
Objectives	Contents
Transform the active voice into	The passive voice
passive.	"It" as impersonal subject
• Use 'it' and 'there' as impersonal	 There' as impersonal subject
subjects	• There as impersonal subject
Unit 8: Reporting	Theory Time Hrs. 4
Objectives	Contents
• Make reporting structures using 'that	Reporting structures- 'that clause'
clause'.	 Other report structures
• Perform reporting and order, request	
and advise.	
Unit 9: Sentence Structures	Theory Time Hrs. 10
Objectives	Contents
• Use time clauses in various situations.	Time clauses
	I lille clauses
• Make conditional clauses with 'if'	Conditional clauses using 'If'
	 Conditional clauses using 'If' Conditional clauses using modals and 'unless'
 Make conditional clauses with 'if' and 'unless'. 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses
 Make conditional clauses with 'if' and 'unless'. 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining relative clauses. 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence Cohesion: Making connection in speech and writing
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining relative clauses. Unit 10: Free Writing 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence Cohesion: Making connection in speech and writing Theory Time Hrs. 07
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining relative clauses. Unit 10: Free Writing Objectives	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence Cohesion: Making connection in speech and writing Theory Time Hrs. 07 Contents
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining relative clauses. Unit 10: Free Writing Objectives Write free paragraphs 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence Cohesion: Making connection in speech and writing Theory Time Hrs. 07 Contents Paragraph Writing
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining relative clauses. Unit 10: Free Writing Objectives Write free paragraphs Write free and guided essays 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence Cohesion: Making connection in speech and writing Theory Time Hrs. 07 Contents Paragraph Writing Essay Writing
 Make conditional clauses with 'if' and 'unless'. Use defining and non-defining relative clauses. Unit 10: Free Writing Objectives Write free paragraphs 	 Conditional clauses using 'If' Conditional clauses using modals and 'unless' Defining Relative clauses Non- Defining Relative clauses Changing the focus of a sentence Cohesion: Making connection in speech and writing Theory Time Hrs. 07 Contents Paragraph Writing

Compose Dialogues	Making Dialogues
Unit 11: Comprehension Passage and Terminologies	Theory Time Hrs. 03
Objectives	Contents
• Answer the short questions based on	Passages related to Medical Issues
the passage.	Common Medical Terminologies
Define Common Medical	
Terminologies	

Part : 2 Extensive reading (Literature)

The Magic of Words (collection of	Theory Time Hrs. 50
poetry, essays, prose)	
Objectives	Contents
Unit 1: Poems	Theory hrs. (4*3 = 12)
	My Heart Leaps Up When I Behold, William Wordsworth
	The Poplar Field, William Cowper
	Keeping Things Whole, Mark Strand
	On the Vanity of Earthly Greatness, Arthur Guiterman
Unit 2: Supernatural Stories	Theory hrs. (4*3 = 12)
	The Recurring Dream
	The Lost Doll
	The House Call
	The Loving Mother
Unit 3: Stories	Theory hrs. (2*3 = 06)
	A Worn Path, Eudora Welty
	The Gardener
Unit 4: Essays	Theory hrs. (4*4 = 16)
	Speaking of Children, Barbara Holland
	The Nightmare Life Without Fuel, Isaac Asimov
	Ooops! Hows' That Again, Roger Rosen Blatt
	The Six Million Dollar Man, Harold J. Morowitz
Unit 5: Drama/Play	Theory hrs. 4
	Malini, Rabindra Nath Tagore,

Recommended texts

1. Link English, Sajhaprakashan,

2. <u>The Magic of Words</u> (collection of poetry, essays, prose)

3. W. Dave (2011), Students Grammar, the University of Brigham, London; Harper Collins Publishers.

Evaluation Scheme:

This paper carries 100 (20 internal + 80 final) marks. The final assessment 80 marks will be divided as follows;

- Magic of Words: 25 Marks
- Link English: 10 Marks
- Student Grammar: 20 Marks
- ➢ Free Writing : 15 Marks
- > Passage : 10 Marks

वर्षः प्रथम तहः प्रमाणपत्र/डिप्लोमा

पाठघण्टा: १२० मूल्याङकन अंक: १०० आन्तरिक मूल्याङ्कन: २० अन्तिम मूल्याङ्कन: ८०

यो पाठयांश प्रवीणता प्रमाणपत्र तहमा अध्ययन गर्ने विद्यार्थीहरूका लागि नेपाली भाषाको व्याकरणात्मक ज्ञान र सुफको विकासका साथै पठनबोध र अभिव्यक्ति क्षमताको विकास गर्ने दृष्टिले राखिएको हो । यसलाई मुख्यतः दुई खण्डमा बांडिएको छः व्याकरण खण्ड र बोध (अभिव्यक्ति) खण्ड । व्याकरण अन्तर्गत वर्ण, वर्णविन्यास, शब्दवर्ग, रूपायन, शब्द निर्माण र वाक्यसम्बन्धी पाठ्यवस्तुहरू राखिएका छन् भने बोध/अभिव्यक्ति अन्तर्गत सामान्य बोध र प्रयोजनपरक बोधका साथै अभिव्यक्ति रचनाका लागि अपेक्षित सीपहरू र समीक्षाका लागि साहित्यिक विधाका पाठहरू समाविष्ट छन् ।

पाठचांशको उद्देश्यः

यो पाठ्यांश पूरा गरेपछि विद्यार्थीहरू निम्नलिखित कुरामा सक्षम हुनेछन् :

- 9. कथ्यभाषा र लेख्यभाषाका बीचको भिन्नता पहिल्याउन ।
- २. अभिव्यक्तिमा प्रयोग हुने शब्दहरूको उपयुक्त वर्णविन्यास लेख्न ।
- ३. शब्दहरूका स्रोत, बनोट र वर्ग-पहिचान गर्न, रूपायन गर्न र निर्माण गर्न ।
- ४. वाक्यतत्व र वाक्यान्तरणका कडीहरू बुभेर आफुना अभिव्यक्तिमा तिनको उपयुक्त प्रयोग गर्न ।
- ४. खास वाक्यतत्वसंग सम्बद्ध ढांचा र सर्न्दभका आधारमा अनुच्छेद रचना गर्न ो
- ६. स्तर अनुरूप पाठ्यसामग्रीमा प्रयुक्त शब्दहरूका आधारमा शब्दभण्डारको विस्तार गर्न ।
- ७. बोध र संक्षेपीकरणका पाठ्यसामग्रीमा प्रयुक्त शब्दहरूका आधारमा शब्दभण्डारको विस्तार गर्न ।
- प. ज्ञान-विज्ञानका विभिन्न शीर्षकहरूमा स्वतन्त्र रूपमा अनुच्छेद र निबन्ध रचना गर्न ।
- ९. तोकिएका आधारमा साहित्यिक कृतिहरूको समीक्षा गर्ने ।

खण्ड १: नेपाली व्याकरण

पर्णाङ्च : ५० पाळघण्टा :६० एकाइ पाठय विषयको विवरण पाठघण्टा अङ्क ٩. वर्ण र वर्णविन्यास : ς (क) उच्चार्य वर्णहरूको परिचय : ς • स्वर र व्यञ्जन वर्णहरू • देवनागरी लिपि र उच्चार्य नेपाली वर्णहरू • नेपाली अक्षरहरूको संरचना, अक्षरीकरण र अक्षरसंख्या निर्धारण (ख) वर्ण विन्यास : γ • कथ्य र लेख्य नेपाली भाषामा भिन्नता • ह्रस्व-दीर्घ (इ, उ), स/श/ष, ब/व, व/ओ, य/ए, ऋ/रि, क्ष/छे, क्ष्य/छ्य, शिरविन्दु र चन्द्रविन्दु, हलन्त, पदयोग र पदवियोग तथा लेख्य चिन्ह सम्बन्धी अशुद्धि संशोधन अभ्यास ર. शब्दभण्डारः 99 9२ शब्दवर्ग, शब्दरूपायन र शब्दनिर्माण • स्रोतका आधारमा शब्दहरूको परिचय, पहिचान र प्रकार • ब्युत्पादनका आधारमा शब्दहरूको परिचय, पहिचान र प्रकार • शब्दवर्ग-नाम, सर्वनाम विशेषण, कियापद, नामयोगी, कियायोगी, संयोजक, विस्मयादिवोधक र निपातहरूको पहिचान- अभ्यास

	 शब्दरूपायन-नाम, सर्वनाम र विशेषणको लिङ्ग, वचन, आदर, कारकका आधारमा तथा क्रियापदको लिङ्ग, वचन, पुरूष, आदर, काल, पक्ष, भाव, वाच्य र अकरणका आधारमा शब्द रूपायनको अभ्यास । शब्द निर्माण अभ्यास निम्नलिखित उपसर्गहरूद्वारा शब्दनिर्माणको अभ्यास प्र, अप, सम्, अनु, वि, अधि, उत्, प्रति, परि, उप, सु, नि, निर, दुर्, अ, अन, कु । निम्नलिखित कृत् प्रत्ययद्वारा शब्दनिर्माणको अभ्यास : 		
	आइ, ओट, ओ, आउ, आहा, अक्कड, उवा, इलो । अक, अन ई इत, य, तव्य ।		
	 निम्नलिखित तद्धित प्रत्ययहरूद्वारा शब्दनिर्माणको अभ्यास : आइ, आली, इया, इलो, ई, ए, एली, ली, ले । 		
	इक, ई, ईय, इत, ता, त्व, मान, वान, आलु । • समस्त शब्दहरूको पहिचान र तत्पुरूष, कर्मधारण, द्विगू, द्वन्द्व, अव्ययीभाव,		
ર.	र वहूब्रीहिको प्रक्रियाबाट समस्त शब्दहरूको निर्माण गेर्ने अभ्यास	१९	२०
	वाक्यतत्त्व, वाक्यान्तरण		·
	(क) वाक्यत्तत्व: उद्देश्य र विधेयको पहिचान		
	• किया र यसका प्रकार		
	• वाक्यका प्रकार: सरल र जटिल वाक्यको पहिचान		
	 वाक्य संश्लेषण र विश्लेषण लिङ्ग, वचन, पुरूष र आदरका आधारमा कर्ता र क्रियापदका बीचको सङ्गति सम्बन्धी अभ्यास 		
	• विशेष्य र विशेषण र नाम र सर्वनामको बीचको सङ्गति सम्बन्धी अभ्यास		
	• विभक्तिनियम तथा ले, लाई, देखि, बाट, द्वारा, को, का, की, रो, रा, री, नो,		
	ना, नी, मा आदि विभक्ति प्रयोगको अभ्यास • सरल र तिर्यक विभक्ति नियमको अभ्यास	90	90
	• सरल र ।तथक् ।वभाक्त ।नथमका अम्थास	૧૦	૧૦
	 (ख) वाक्यान्तरण : विभिन्न काल, पक्ष, भाव, अकरण, वाच्य, प्रेरणार्थक, उक्ति आदिमा 		
	वाक्यान्तरण गर्ने अभ्यास		

खण्ड खः बोध तथा अभिव्यक्ति

पाठघण्टा : ६०

पूर्णाङ्च : ५०

		ू एक	· · · · ·
एकाइ	पाठ्य विषयको विवरण	पाठघण्टा	अङ्क
٩.	बोध र शव्दभण्डार	१२	१२
	चिकित्सा विज्ञानसम्बन्धि गद्यांशहरूको बोध र शब्दभण्डारको अभ्यास		
	चिकित्सा, शल्य चिकित्सा तथा स्वास्थ्य विज्ञानसम्बन्धी प्राविधिक		
	शब्दहरूको ज्ञान अभ्यास (अर्थ लेख्ने र अर्थ खूल्ने गरी वाक्यमा प्रयोग ————————————————————————————————————		
	गर्ने)		
	इन्द्रलुप्त, उत्क्लेस, ज्वर, पाण्डुरोग, प्रमेह, मधुमेह, पित्तदोस, प्रदर (१.		
	रक्तपदर, २. स्वेतप्रदर) क्षयरोग, नशच्छे्दन, रक्तचाप, उच्चू रक्तचाप,		
	न्यून रक्तचाप, गर्भपात⁄पतन, हृदयरोग, पाचनकिया, पित्तविकार,		
	रक्तविकार, चिकित्सा, निदान, परिचारिका, प्रसववेदना, प्रसुति, औषधालय,		

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	चिरफार, बहिरंग, हिक्का, हरिताल पार्नु, हरिनाश, हियो उठ्नु, वातज्वर,		
	सिफर पल्टिनु, सिङ्गारू, सेपाउनु, सप्कों गर्नु, सन्निपात, सभिपात, शल्य		
	चिकित्सा, शूल, शल्योपचार, माँसु फरफराउँनु, माथा विग्रनु, माटे, माई,		
	भूंग्रेज्वरो, बेर्नु निस्कुन, बालतोड, बाथ, बान लाग्नू, बाउंडिनु, बहलाग्नु,		
	बेमन, विरेचन, फुसिनु, फुलो पर्नु, फांकफुक, पौंठों बस्नु, पेंट बटारिनु,		
	पेट काट्नु, पेट पोर्ल्नु, पिनाश, पाछ्नु, निसलोठ, धर्मकी, दोख, दमें		
	खटिरो, दम, डकार्नु, ठेउला, भुसिलो डकार आउनु, भिजो मान्नु, भाडा,		
	जिरिङ्ग गर्नु, जल गडा, जनै खटिरा, जगाउनु, छोप्नु, छेर्नु, चिलचिलाउनु,		
	चिप्रा बस्तु, चस्का पर्तु, घमौरा, गला लाग्तु, गलगण्ड, गलफुलो (हांडे),		
	विश्वा असुं, यस्का पुं, वमारा, गेला लागुं, गेलगण्ड, गेलपुंखा (हाड),		
	गोला चल्नु, गानो चल्नु, खरापानी लाग्नु, कण्डु, कुण्ठ, कोर, कोख,		
	कैंठ्रिनु, कुंजो, कांसो लाग्नु, कास, काम्नु, कामज्वर, कांडो, कांध लाग्नु,		
	काई लाग्नु, क्ब्जियत, औसनी लाग्नु, औडाहा च्ल्नु, ओछ्यान पर्नु,		
	ओखत मुलो, ओइलाउनु, ऐंठन, उभर्को लाग्नु, उदररोग, उपर्तली, उकुँच		
	पल्टनु, अर्बुद, अजीर्ण, अपस्मार, आन्द्रा बटार्नु, आंत, आंठी गांठी, आंठे,		
	आङ् चल्नु, आक्तो, आउं, अरूची, अम्मल, अमल पित्त, अमन हुनु,		
	अग्नी जाग्नु, अतिसार, अंधो खटिरा, स्वेदन ।		
२	संक्षेपीकरणः :	8	8
	 बुंदा टिपोट तथा संक्षेपीकरण गर्ने अभ्यास 		
	 अनुच्छेद, पत्र, निवेदन, विज्ञापन र प्रतिवेदन लेखन 		
	 ज्ञॉन विज्ञान र प्रविधिसंग सम्बन्धित विभिन्न विषय शीर्षकहरूमा 		
	अनुच्छेद लेख्ने अभ्यास		
३	निवन्ध लेखन :	90	ፍ
	• निबन्ध योजना र सोसंग सम्बन्धित बुंदा अनुरूप अनुच्छेद गठनको		
	अभ्यास		
	 वस्तुपरक र भावपरक निवन्ध लेखनको अभ्यास 		
8	कृति समीक्षा :	३४	રદ્
	विषयवस्तु, कथानक, पात्र, परिवेश, सन्देश, शीर्षक र भाषा	۳.	17
	शैलीका आधारमा निम्नलिखित रचनाहरूको समीक्षात्मक अभ्यास:		
	कथा :		
	 गुरू प्रसाद मैनाली छिमेकी 		
	 विश्वेश्वरप्रसाद कोइराला सिपाही 		
	• इन्द्रबहादुर राई रातभरि हुरी चल्यो		
	• रमेश विकल मधुमालतीको कथा		
	निवन्ध :		
	• लक्ष्मी प्रसाद देवकोटा पहाडी जीवन		
	 शंकर लामिछाने एक पत्र सम्पादकलाई 		
	 भैरव अर्याल महापुरूषको संगत 		
	कविताः		
	 लेखनाथ पौडेल नैतिक दृष्टान्त 		

	٠	पारिजात	मानूषी
	•	गोपाल प्रसाद रिमाल	आमाको सपना
	•	माधव प्रसाद घिमिरे	नेपालै नरहे
न	गटक	:	
	٠	विजय मल्ल	बहुला काजीको सपना

द्रष्टव्य :

२० प्रतिशत अङ्क आन्तरिक मूल्याङ्कनका लागि छुटचाइएको छ भने८० प्रतिशत अङ्क अन्तिम मूल्याङ्कनका लागि छुटचाइएको छ।

सहायक पुस्तकहरू (सम्बद्ध अंश मात्र):

- मोहनराज शर्मा, शब्द रचना र वर्ण-विन्यास वाक्यत्तत्व र अभिव्यक्ति, काठमाण्डौ बुक सेन्टर, काठमाण्डौ
- चित्र कुमार गुरूङ्ग एम्.एस्सी.र केदार न्यौपाने एम्.ए., प्राविधिक शब्दार्थावली (चिकित्सा तथा विज्ञान खण्ड), त्रिभुवन विश्वविद्यालय, चिकित्सा शाश्त्र अध्ययन संस्थान, अनुसन्धान शाखा, महाराजगंज, काठमाण्डौ ।
- त्रि.वि. पाठ्यक्रम विकास केन्द्र, अनिवार्य नेपाली शिक्षण निर्देशन, काठमाण्डौ
- सागरमणि पाण्डेय, ईश्वरी पाण्डेय, अनिवार्य नेपाली, रत्नसागर प्रा.लि., काठमाण्डौ
- टीकाहरि बराल र अन्य, सीटीइभीटी अनिवार्य नेपाली ।

Social Studies

Year First

Level Certificate

Total Hours:80Full Marks:50

Course Description

This course offers an introduction to Nepal in general. It provides basic information about the geography, natural resources, history, society, culture, politics, economy, and foreign policy of Nepal. Analysis of current social and national problems are discussed relating to these country's features.

Course Objectives

On completion of this course the student will be able to:

- Identify the climate, geography, natural resources and administrative units of Nepal.
- Summarize the history of Nepal.
- Describe the arts and cultural achievements of Nepal.
- Explore the social problems challenging Nepal at present.
- Analyze the salient features of Nepalese economic development.
- Distinguish between democratic and non-democratic forms of government.
- Examine the features of the constitution of the Federal democratic Republic Nepal, 2047(1990) and 2072 (2015).
- Mention the chief characteristics of Nepal's foreign policy.
- Describe Nepal's role in the peace-keeping efforts of the world
- Summarize the political development in Nepal.

Evaluation : written exams

Course: Social Studies	Hrs. theory 80	Marks: 50
Unit: 1 Introduction		
Sub-unit 1.1: The land of Nepal	Hrs. theory 10	6
Objectives:	Content:	
 Describe the geographical divisions of Nepal. Identify the administrative units of Nepal. Compare the ecological, climatic, and regional diversities in Nepal. Describe the natural resources of Nepal. 	 Geographical locations, diversities, and unique characteristics of Nepal. Geographical divisions of Nepal: Ecologic Climatic Rivers Vegetation Administrative units Natural resources of Nepal (general introduction). Patterns of land use in Nepal. 	
Unit 2: Political History of Nepal	Hrs. theory 16	10
Sub-unit 2.1: Ancient and medieval Nepal	Hrs. theory 6	
Objectives:	Content:	
 Discuss the historical events of the ancient period. Explain why the period of Lichhavi rule is known as the golden period. Summarize the brief history of Doya, Khash and Malla kingdoms. 	 Ancient Nepal: Origin of the word "Nepal" Ancient dynasties: Gopal, Mahispal, Kirat Licchavi period (Licchavi civilization). 	

	 Medieval Nepal: Doya Rajya or Karnatac Kasha kingdom of Karnali region Malla kingdom of Kathmandu valley 	
Sub-unit 2.2: Unification of Nepal	Hrs. theory 5	
 Objectives: Describe the geographical fragmentation of Nepal in the later medieval period. Identify the causes of geographical fragmentation. Explain the political, social, economic and geographical situation of Nepal before the enthronement of Prithvi Narayan Shah. Analyse the policies adopted by Prithvi Narayan Shah and his successors during the time of unification. Identify the factors which influenced the rise of the Ranas. 	 Content: Petty states of Nepal (Baisi, Chaubisi), states in Kathmandu valley, three Sena states of eastern Nepal. Political, social, economic and geographical conditions of Nepal before Prithvi Narayan Shah. Unification of Nepal: role of Prithvi Narayan Shah, Rajendra Laxmi, Bahadur Shah, and Bhim Sen Thapa. Political instability and the factors which influenced the rise of Jang Bahadur: Conspiracies, Assassinations, Kot Massacre, Bhandarkhal Parva, 	
Sub-unit 2.3: Peoples' Movements and Rise	Alau Parva.Hrs. theory 5	
of democracy		
Objectives:	Content:	
 Assess the improvement works of the first elected government of Nepal. Examine the people's movement of 2046 B.S. and its impacts. Examine the characteristics of the constitution of Nepal , 2047BS & 2072BS. Discuss the impact of people's second movement on the social conditions of Nepal. 	 The first elected government of Nepal 2015. People's movement of 2046 BS (1990 A.D). Comparative study of the characteristics of the constitution of 2047 and the Federal Democratic Republic Nepal (2072) B.S. Second people's second movement 2062/063 	
Unit 3: Society & Culture	Hrs. theory 14	10
Sub-unit 3.1: Development of Nepalese culture and society	Hrs. theory 7	
 Objectives: Analyze the population growth of Nepal: Describe the contributing factors of population growth. Describe the origin of the caste system in Nepal Analyze the current laws about cast practice. 	 Content: Population growth in Nepal Contributing factors of population growth: fertility, mortality, and migration. Caste beliefs and constitutional provision. Establishment of national languages 	

 Discuss the establishment national languages. Identify different ethnic languages and culture. Identify the social problems of Nepal: 	 Ethnic languages and culture: Nepali Newari Sanskrit Maithili Social problems: Poverty Gender issues Unemployment Drug addictions HIV/AIDS Prostitution Child labor Trafficking Other 	
Sub-unit 3.2: Arts and religion	Hrs. theory 7	
Objectives:	Content:	
 Analyze the cultural heritage of Nepal. Discuss the development of arts in Nepal. Explain the history of religious harmony in Nepal. 	Cultural heritages in Himalayan, Hilly and Terai regions: Food habits Dress and ornaments Festivals and temples Music, songs and dances Occupations Art in Nepal: Paintings, sculpture and architecture in ancient, medieval and modern times. Religions in Nepal: Hinduism Buddhism Muslim Kirat Christian	
Unit 4: Nepalese Economy	Hrs. theory 14	8
Sub-unit 4.1: Resources and development	Hrs. theory 7	
Objectives:	Content:	
 Analyse the affecting factors of Nepalese economic development. Explain the various features of Nepal's economic system. 	Affecting factors for the Nepalese economy: • Poverty, • Inequality, • Population growth • Unemployment, • Regional disparities • Land tenures. Features of the Nepalese economic system: • Agriculture and land reform system • Cottage and large scale	

	 Internal and external trade Tourism Cooperation Planned economy Mixed economy (capitalism and socialism) 	
Sub-unit 4.2: Natural resources	Hrs. theory 7	
Objectives:	Content:	
Explain the resources for the economic	Resources of national development:	
development.	Human resources	
	• Forests	
	• Land	
	• Water	
Unit 5: Politics and Government	Minerals	•
	Hrs. theory 14	8
Sub-unit 5.1: Democratic constitution	Hrs. theory 2	
Objectives:	Content:	
Define democracy	 Meaning and definition of democracy; Characteristics of democratic 	
• Distinguish between a democratic and		
non-democratic form of government.Explain the salient features of the	government;Features of the constitution of	
constitution of Nepal 2047 and 2072	2047 and 2072BS.	
B.S.	2047 and 2072D5.	
Sub-unit 5.2: Federalism	Hrs. theory 7	
• Explain the structure of the state and	Structure of the state	
distribution of power.	• Federal	
	Provincial	
• Explain the Civic duties and	• Local	
responsibilities for the successful	Distribution of state power	
implementation of the constitution of	• Federation	
Nepal.	Province	
	• Local	
	• Legislative:	
	Federal Parliament(House of Representatives and National	
	Assembly)	
	Composition, power and functions	
	Executive: Federal Executive(Council of	
	Ministers)	
	Composition, power and functions	
	• Judiciary: Courts	
	• Supreme court	
	Appeal court District court	
	• District court Composition, power and functions of	
	Judiciary	
	President and vice president:	
	 Functions, duties and Authorities 	
	• Fundamental rights and duties of	
	the citizen/people	

Sub unit 5.3: Provincial Legislature and Provincial Executive	Hrs. theory 5	
 Explain the structure of local excecutive. Explain the Interrelationship between the Federation provinces and local level. 	 Local Legislature and local executive Local legislature Village Assembly, Municipal Assembly Local Executive Village executive and municipality District Assembly and District coordination committee Interrelationship between the Federation, provinces and local level 	
Unit 6: Foreign Policy	Hrs. theory 12	8
 Objectives: Explain the characteristics of Nepal's foreign policy. Explain Nepal's foreign policy with special reference to her relations with India and China. Describe Nepal's role in the peacekeeping process of UNO. Assess the importance of regional cooperation: SAARC. 	Content:Nepal's foreign policy:• Geographical• Historical• Cultural• Economic• International.Features of Nepal's foreign policy:• Non-aligned• Panchasila• Acceptance of UNO charter• Regional cooperation• Peace movement• Disarmament• OthersNepal's relations with its neighbors:• China• India• UNOFoundation of SAARC;Nepal's role for the development of theSAARC countries.	

References

<u>Faces of Nepal</u>, Jagadamba Press. Bista, Dor Bahadur<u>, People of Nepal</u> Bista Dur Bahadur, <u>Sabai Jatko Fulbari</u>

Year : First Level : Certificate Assessment Marks: 100

Credit Hours: Theory 160 Practical: 40

Course description

This course provides basic knowledge of the normal structure and function of the systems of the human body. The content prepares the student to understand the pathology and clinical features of medical and surgical conditions, diseases and disorders, as well as the rationale for treatments and management.

Objectives

On completion of this course the student will be able to:

- Identify the classifications of the systems of the human body.
- Locate and describe the structure and function of the components of each body system.
- Explain the interrelationship of the body systems.
- Transfer knowledge of anatomy and physiology of the body to medical and surgical circumstances.
- Explain the mechanisms of body repair and resistance to disease.
- Describe the physical changes that occur during normal growth and development, from conception to senescence.

Recommended Text

- 1. Amatya Dr. Mrigendra, A Text Book of Anatomy and Physiology, 2nd edition
- 2. *Pal, G.K. & Pal, Pravati,* Text Book of Practical Physiology, Publised by Orirnt Longman Private Limited, Chennai, India
- 3. A Text Book of Anatomy and Physiology, 2nd edition, Dr. Mrigendra Amatya,
- 4. Text Book of Practical Physiology, *By G.K. Pal & Pravati Pal*, Publised by Orirnt Longman Private Limited, Chennai, India

Reference Texts

- 1. Shier, D., Butler, J. & Lewis, R., Hole's Human Anatomy and Physiology. Wm. C. Brown Publishers, London. 1996 or current edition.
- 2. Chauarasia, Handbook of Human Anatomy. CBS Publication. Current edition.

Course: Anatomy	Hrs. theory 80 Hrs. tutorial 20
Unit 1: Introduction of Anatomy	Hrs. theory 5 Hrs. tutorial 1
Objectives:	Content:
1. Define anatomy and relation between	1. Concepts of anatomy: the homeostatic,
anatomy and physiology2. Describe the locations of each main body cavity and list the organs within each cavity.	integrated, self-healing nature of body cells and tissues.Organization of the human body.
3. Name the organ systems, tell the function of each system, and list the organs associated with each system.	 Review of cellular and tissue characteristics, functions. Anatomical terms:
 Define the terms that describe body positions, body sections, and body regions. 	CardiovascularDigestion
6. Review the general characteristics of cell structure, function, and reproduction.	ExcretionOrgan
 Describe the general characteristics and functions of the body tissues: epithelial, 	PeritonealPericardial
	Physiology

Part I: Anatomy

cartilage, connective, bone, nerve, adipose and three kinds of muscle tissue.	ThoracicVisceral
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub unit 2.1: Skin and integumentry	Hrs. theory 4 Hrs. tutorial 1
system	
Objectives:	Content:
 Describe the four chief types of membranes. Describe the structure of the various layers of the skin with diagram. Describe the location and function of the accessory organs located within the layers of skin. Summarize the factors that determine skin 	 Types of membranes. Layers of the skin. Accessory organs and glands of the skin.
color.	
Evauation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.2: Skeletal system	Hrs. theory 6 Hrs. Practical 2
Objectives:	Content:
 Classify bones according to their shape and give an example from each group. Describe the structure of a bone and tell the function of each parts Differentiate between intramembranous and endochondral bones and tell how each type of bone grows and develops. Differentiate between axial and appendicular skeletons and name the major bones of each system. Identify and label long, short and flat bones. Locate and identify the bones that comprise the skull, vertebral column, thoracic cage, pectoral girdle, upper limb, pelvic girdle, and lower limb. Locate and identify the features of these bones. 	 Bone growth and development. Skeletal organization. Identification and labeling of long, short and flat bones. Terms related to the skeletal system: axial appendicular articular cartigage diaphysis fontanel hematopoiesis marrow periosteum
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources:
Unit 2: Systems of the Body	classroom instruction, models, charts.Hrs. theoryHrs. tutorial
Sub-unit 2.3: Skeletal joints	Hrs. theory 3 Hrs. Practical 2
Obectives:	Content:
 Describe how joints can be classified according to the type of tissue that holds them together. Describe how bones are held together in fibrous joints and cartilaginous joints. 	 Classifications of joints: fibrous, cartilaginous, synovial. Types of joint movement. Structure and types of synovial joints. Terms related to joints:

 Describe the structure of a synovial joint. List six types of synovial joints and give an example of each type. Describe these joints and explain how the articulation parts are held together: Shoulder elbow 	 articulation bursa ligament suture symphysis
 hip knee ankle wrist 	
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.4: Muscular system	Hrs. theory 4 Hrs. Practical 1
Objectives:	Content:
 Name the chief parts of a skeletal muscle fiber. Distinguish between the structures and functions of skeletal, cardiac and smooth muscles. Identify and describe the locations of the chief skeletal muscles and describe the action of each muscles of facial expression and mastication muscles that move the: head pectoral girdle arm forearm hand abdominal wall pelvic outlet thigh leg foot 	 Structure of a skeletal muscle: connective tissue coverings skeletal muscle fibers
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.5: Nervous system	Hrs. theory 3 Hrs. Practical 1
Objectives:	Content:
 Describe the structure of a neuron. Explain how neurons are classified. Describe a reflex arc. 	 Classifications of neurons and neuralgia. Nerve pathways. Reflex arc. Terms related to the nervous system: axon central nervous system dendrite
	effectormyelin

systemObjectives:Content:1. Describe the coverings of the brain and spinal cord.1. Divisions of the central nervous system:2. Describe the structure of the spinal cord.1. Divisions of the central nervous system:3. Locate the chief parts of the brain .• meninges4. Locate the motor, sensory, and association areas of the cerebral cortex.2. Structure and function of the cerebrum.5. Describe the formation and storage of cerebrospinal fluid2. Structure and function of the cerebral nervous system:6. Locate the chief components of the peripheral nervous system.5. Terms related to the nervous system:7. Describe the structure of a peripheral nerves.5. Terms related to the nervous system:6. the cranial and spinal nerves.5. Terms related to the nervous system:7. Describe the location and function of each of the cranial and spinal nerves.5. Terms related to the nervous system:6. the cranial and spinal nerves.6. Terebral hemisphere7. define the formation and spinal nerves.6. Terebral hemisphere8. Describe the location and function of each of the cranial and spinal nerves.6. Terebral hemisphere9. cerebral hemisphere1. Divisions of the peripheral nervous system:9. define the total nerves.1. Divisions of the peripheral nervous system:9. define the total nerves.1. Divisions of the peripheral nervous system:9. define the total nerves.1. Divisions of the peripheral nerve.9. define the total nerves.1. Divisions of the nervous system:9. define the total nerves.1. Divisions of the peripheral	Evaluation methods: written and viva exams. Unit 2: Systems of the Body Sub-unit 2.6: Components of the nervous	 neuroglia neuron neuroratansmitter receptor reflex synapse Teaching / Learning activities and resources: classroom instruction, models, charts. Hrs. theory Hrs. tutorial Hrs. theory 3 Hrs. Practical 1
 Describe the coverings of the brain and spinal cord. Describe the structure of the spinal cord. Locate the chief parts of the brain . Locate the motor, sensory, and association areas of the cerebral cortex. Describe the formation and storage of cerebrospinal fluid Locate the chief components of the peripheral nervous system. Describe the structure of a peripheral nerve. Describe the location and function of each of the cranial and spinal nerves. Describe the location and function of each of the cranial and spinal nerves. Terms related to the nervous system: cerebral cortex cerebral cortex denergic brain stem cerebral memous system: adrenergic brain stem cerebrum cholinergic hypothalamus medulla oblongata meninges midbrain parasympathetic reticular formation sympathetic reticular formation sympathetic thalamaus ventricle Reflex-Deep tendon reflex and superficial reflex Teaching / Learning activities and resources: classroom instruction, models, charts. 		Content:
classroom instruction, models, charts.Unit 2: Systems of the BodyHrs. theoryHrs. tutorial	 spinal cord. Describe the structure of the spinal cord. Locate the chief parts of the brain . Locate the motor, sensory, and association areas of the cerebral cortex. Describe the formation and storage of cerebrospinal fluid Locate the chief components of the peripheral nervous system. Describe the structure of a peripheral nerve. Describe the location and function of each of the cranial and spinal nerves. 	 meninges spinal cord brain Structure and function of the cerebrum. Ventricles and cerebrospinal fluid Divisions of the peripheral nervous system: cranial nerves spinal nerves Terms related to the nervous system: adrenergic brain stem cerebellum cerebral cortex cerebral hemisphere cerebrum cholinergic hypothalamus medulla oblongata meninges midbrain parasympathetic reticular formation sympathetic thalamaus ventricle Reflex-Deep tendon reflex and superficial reflex
	Evaluation methods: written and viva exams.	
Sub-unit 2.7: Somatic and special sensesHrs. theory3Hrs. PracticalObjectives:Content:		

 Name five kinds of sensory receptors and explain the function of each Locate and name the parts of the ear and explain the function of each parts. Name the parts of the eye and explain the function of each parts. 	 Functions of receptors and sensations. Olfactory organs: location and functions. Auditory organs: location, function of parts the middle, inner and external ear. Terms related to senses: accommodation ampulla chemoreceptor cochlea cornea dynamic/static equilibrium labyrinth macula optic photoreceptor referred pain retina sclera thermoreceptor
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.8: Endocrine system	Hrs. theory 6 Hrs. Practical 1
Objectives:	Content:
 Differentiate between endocrine and exocrine glands. Name and locate the chief endocrine glands and tell the hormones they secrete. 	 Characteristics f the endocrine system. Structures, functions and locations of endocrine glands: pituitary thyroid parathyroid adrenal pancreas thymus ovary/testes/placenta pineal Terms related to endocrine system: adrenal medulla aldosterone anterior pituitary epinepherine catacholamine glucagon luteinizing hormone metabolic rate norepinepherine prolactin prostaglandin steroid thyroxine

Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.9: Blood	Hrs. theory 3 Hrs. Practical 1
Objectives:	Content:
 Describe the characteristics of the blood and tell the functions of blood Differentiate between the different types of blood cells 	1. Components and function of the blood.
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.10: Cardiovascular system	Hrs. theory 6 Hrs. Practical 1
Objectives: 1. Name the organs of the cardiovascular system and describe their functions	Content:1. Structures and functions of the heart.2. Locations, functions and characteristics of
 Locate and name the major parts of the heart and describe the functions of each. 	 arterioles and arterioles. Capillaries and their actions.
3. Describe the pathway of the blood through the heart and the vessels of the coronary circulation.	 Locations, functions and characteristics of veins and venules. Names, functions and locations of the
 Compare the structures of the chief blood vessels of the body. Compare the pulmonary and systemic 	vascular components of the pulmonary and systemic circulatory systems.6. Terms related to circulation:
 Compare the pulmonary and systemic pathways of the cardiovascular system Identify and locate the chief arteries and veins 	 arteriole atrium
of the pulmonary and systemic systems.	cardiac cyclecardiac output
	diastolic pressureelectrocardiogram
	myocardiumpericardium
	 peripheral resistance sphygmomanometer pacemaker
	 systolic pressure vasoconstriction
	vasodilationventricle
	venuleviscosity
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.11: Lymphatic system and immunity	Hrs. theory 3 Hrs. Practical 1
Objectives:	Content:

 Describe the functions of the lymphatic system and locate the chief lymphatic pathways. Locate the chief lymph nodes and describe their functions. 	 Patterns of Lymphatic movement. Lymph node location, function and structure. Terms related to lymphatics and immune system: allergen antibody antigen interferon lymphocyte macrophage pathogen vaccine
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.12: Digestive system	Hrs. theory 7 Hrs. Practical 1
Objectives:	Content:
 Locate and describe the functions chief organs of the digestive system. Name the parts of the stomach, liver and gall bladder, large and small intestine. Describe the structure of the wall of the alimentary canal. List the enzymes of secreted by various digestive organs and glands . 	 Structures and functions of the alimentary canal. Movement and enervation f the alimentary canal. Mouth structures and functions. Pharynx and esophagus structure and function. Structure and functions of the pancreas and liver in regard to digestion. Structure and function of the small and large intestines. Terms related to the digestive system: absorption anal canal bile chyme deciduous duodenum emulsification feces jejunum ilium mesentery mucous membrane panacreatic juice peristalsis pyloric sphinctor rectum sphincter muscle vermiform appendix villi/villus

Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial
Sub-unit 2.13: Respiratory system	Hrs. theory 6 Hrs. Practical 1
 Describe the functions of the respiratory system Locate the organs of the respiratory system Explain how inspiration and expiration are achieved. Locate the respiratory center and explain how it controls normal breathing. Describe the functions of the respiratory membrane. 	 Organs of the respiratory system. Terminology related to respiration: alveolus bronchial tree diaphragm glottis intercostal muscles hilus hyperventilation oxyhemoglobin parietal pleura partial pressure pleural cavity respiratory membrane respiratory volume surface tension
Evaluation methods: written and viva exams. Unit 2: Systems of the Body Sub-unit 2.14: Urinary system	 visceral pleura Teaching / Learning activities and resources: classroom instruction, models, charts. Hrs. theory Hrs. theory 5 Hrs. Practical 1
Objectives:	Content:
 Locate the organs of the urinary system and describe their general function 	1. Location, structure and function of the
 Describe the structure and functions of the kidneys. Describe the pathway of blood through the kidneys. Describe a nephron and explain the function of each parts . Describe the structure of the ureters, urinary bladder, and urethra. 	 organs of the urinary system. 2. Renal circulation. 3. Terms related to the urinary system: renal cortex renal medulla glomerulus afferent arteriole efferent arteriole juxtaglomerular apparatus nephron loop pertibular capillary renal corpuscle renal tubule renal plasma threshold retroperitoneal autoregulation destusor muscle
 Describe the structure and functions of the kidneys. Describe the pathway of blood through the kidneys. Describe a nephron and explain the function of each parts . Describe the structure of the ureters, urinary bladder, and urethra. 	 2. Renal circulation. 3. Terms related to the urinary system: renal cortex renal medulla glomerulus afferent arteriole efferent arteriole juxtaglomerular apparatus nephron loop pertibular capillary renal corpuscle renal tubule renal plasma threshold retroperitoneal autoregulation
 Describe the structure and functions of the kidneys. Describe the pathway of blood through the kidneys. Describe a nephron and explain the function of each parts . Describe the structure of the ureters, urinary 	 Renal circulation. Terms related to the urinary system: renal cortex renal medulla glomerulus afferent arteriole efferent arteriole juxtaglomerular apparatus nephron loop pertibular capillary renal tubule renal plasma threshold retroperitoneal autoregulation destusor muscle

Objectives:	Content:		
1. Name the parts of the male reproductive	1. Structure and function of the external		
system and describe the general functions of	organs: penis, testes, scrotum.		
each.	2. Internal accessory organs: epididymis, vas		
2 Describe the path of sperm sells from their	deferens, seminal vesicle, prostate gland,		
origin to their exit from the body	bulbourethral glands,		
3. Describe the structure of penis and explain the	3. Terms related to male reproductive system:		
mechanism of erection and ejaculation.	• glans penis		
4. Describe how a vasectomy is performed, and	prepuce		
discuss the relative simplicity of this	 corpora cavernosa 		
procedure.	 corpus cavernosa 		
1	 spermatogenesis 		
	• spermatogenesis		
	• inguinal		
	• gonadotropin		
	• testosterone		
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources:		
	classroom instruction, models, charts.		
Unit 2: Systems of the Body	Hrs. theory Hrs. tutorial		
Sub-unit 2.16: Female reproductive system	Hrs. theory 4 Hrs. tutorial 1		
Objectives:	Content:		
1. Name the parts of the female reproductive	1. Structure and function of the vagina,		
system and describe the general functions of	clitoris, labia, ovaries, fallopian tubes,		
each.	uterus, breasts and mammary glands.		
	2. Fertilization and embryonic		
	development.		
	3. Terms related to the female reproductive		
	system:		
	• follicle		
	• estrogen		
	• progesterone		
	fertilization		
	• meiosis		
	 oogenesis 		
	• zygote		
	implantation		
	 infundibulum 		
	• orgasm		
	ovulation		
	menstrual cycle		
	• puberty and menarche		
	 placenta 		
	menopause		
	-		
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.		
Unit 3: Human Growth & Development	Hrs. theory 5 Hrs. Practical 1		
<u>Sinter Human Growth &</u> Development	Content:		
Objectives:			
Objectives: 1. Describe the process of development from	1. Embryionic and foetal development.		
Objectives:			
Objectives: 1. Describe the process of development from	1. Embryionic and foetal development.		

 Describe the formation and function of the placenta. Define the term foetus and describe the foetal stage of development. Describe the path of blood through the foetal circulatory system. 	 chorion zygote embryo foetus placenta umbilical cord prenatal neonatal postnatal
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.

Part II: Physiology

Course: Physiology	Hrs. theory 80 Hrs. tutorial 20
Unit 1: Introduction of Physiology	Hist theory00Hist theorian20Hrs. theory2Hrs. tutorian1
Objectives:	Content:
 Define physiology and relation between anatomy and physiology. List and describe the chief characteristics of life and the chief requirements of living organisms. Define homeostasis and discuss its importance for survival. Name the organ systems, tell the function of each system, and list the organs associated with each system. Review the general characteristics of cell structure, function, and reproduction. Describe the general characteristics and functions of the body tissues: epithelial, cartilage, connective, bone, nerve, adipose and three kinds of muscle tissue. 	 Content: Concepts of physiology: the homeostatic, integrated, self-healing nature of body cells and tissues. Organization of the human body. Review of cellular and tissue characteristics, functions. Physiological terms: Cardiovascular Digestion Excretion Organ Peritoneal Physiology Thoracic Visceral Homeostasis Yoga
Evaluation methods: written and viva exams. Unit 2: Systems of the Body	Aging Teaching / Learning activities and resources: classroom instruction, models, charts. Hrs. theory Hrs. Practical
Sub unit 2.1: Skin and integumentry	Hrs. theory 3 Hrs. Practical 1
system Objectives:	Content:
 Describe the function of each layer of skin. Describe the location and function of the accessory organs located within the layers of skin. Explain how the skin regulates body temperature. Summarize the factors that determine skin color 	 Thermoregulatory function of the skin. Pigmentation. Healing of wounds, burns; skin disorders.
Evauation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.2: Skeletal system	Hrs. theory 4 Hrs. Practical 1
Objectives:	Content:
 Describe the chief functions of bones. Describe the structure of a bone and tell the function of each parts Discuss the effects of hormones , sunlight, and exercise on bone development. 	 Functions of bone: support/protection body movement blood cell formation Bone growth and development. Factors affecting growth and repair. Skeletal organization.

Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.		
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical		
Sub-unit 2.3: Skeletal joints	Hrs. theory 3 Hrs. Practical 1		
Obectives:	Content:		
 Describe the functions of different types of Joint. Shoulder elbow hip knee ankle wrist Explain how skeletal muscles produce movements at joints and give examples of different kinds of movement. Describe the function of the fontanels? 	 Explanation how skeletal muscles produce movements at joints and give examples of different kinds of movement. Terms related to joints: Shoulder elbow hip knee ankle wrist Function of the fontanels 		
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.		
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical		
Sub-unit 2.4: Muscular system	Hrs. theory3Hrs. Practical 1		
Objectives:	Content:		
 Name the chief parts of a skeletal muscle fiber. Differentiate between fast and slow muscles and between twitch and sustained contraction. Name the chief parts of a skeletal muscle fiber and describe the function of each part. Describe the process of muscle contraction. Explain how muscle contractions produce body movement and maintain posture. Distinguish between the structures and functions of skeletal, cardiac and smooth muscles. 	Content: 1. Structure of a skeletal muscle: • skeletal muscle fibers • neuromuscular junction • motor units 2. Skeletal muscle contraction: • role of myosi and actin • stimulus for contraction • muscular responses 3. Smooth muscle fibers and contraction. 4. Cardiac muscle fibers and contraction. 5. Actions of the muscles of the skeletal system: 6. Terms related to the muscular system: • antagonist • fascia • insertion • muscle impulse • neurotransmitter • origin • synergist		
	Teaching / Learning activities and resources: classroom instruction, models, charts.		
Evaluation methods: written and viva exams.	classroom instruction, models, charts.		
Evaluation methods: written and viva exams. Unit 2: Systems of the Body Sub-unit 2.5: Nervous system			

 Describe the structure of a neuron. Explain how neurons are classified. Describe a reflex arc. Explain the general functions of the nervous system Describe the events that lead to the conduction of a nerve impulse. Explain how a nerve impulse is transmitted from one neuron to another. Explain what is meant by reflex behavior. 	 Classifications of neurons and neuralgia. cell membrane function. Synapse function. Neurotransmitters and neuropeptides. Impulse processing. Nerve pathways. Reflex function. Terms related to the nervous system: axon central nervous system dendrite effector myelin neuron neuroratansmitter receptor reflex synapse
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.6: Components of the nervous	Hrs. theory 3 Hrs. Practical 1
system	
Objectives:	Content:
 Locate the chief components of the peripheral nervous system. Describe the structure of a peripheral nerve. Describe the structure of the spinal cord. Locate the chief parts of the brain Explain the meaning of hemisphere dominance. Mention the chief functions of Spinal cord. Describe the functions of each part of brain. Describe the stages of memory storage. Describe the formation and storage of cerebrospinal fluid Describe the composition and function of Cerebrospinal fluid. Describe the functions of the limbic system and reticular formation. Locate the chief components of the peripheral nervous system. Describe the location and function of each of the cranial and spinal nerves. Compare the functions of the sympathetic and parasympathetic divisions of the autonomic nervous system. 	 Structure and function of the cerebrum. Effects of cerebral injury. Divisions of the peripheral nervous system: Cranial nerves and its test Spinal nerves Functions of the autonomic nervous system. Terms related to the nervous system: adrenergic brain stem cerebral cortex cerebral hemisphere cerebral hemisphere cerebral mus medulla oblongata meninges midbrain parasympathetic reticular formation sympathetic thalamaus ventricle

Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.7: Somatic and special senses	Hrs. theory 3 Hrs. tutorial 1
 Objectives: Name five kinds of sensory receptors. explain the function of each Locate and name the parts of the earand explain the function of each parts. Name the parts of the eye and explain the function of each parts. Differentiate between static and dynamic equilibrium. Explain the function of each sensory receptors. Explain how the receptors stimulate sensory impulses. Describe how sensation is produced. Describe how the sense of pain is produced. Explain the functions of each parts of the ear. Explain the functions of each parts of the ear. Explain the visual nerve pathway. 	Institutorial 1 Content: 1. Functions of receptors and sensations. 2. Function of the somatic senses: touch/pressure, temperature, stretch, pain. 3. Olfactory organs: location and functions. 4. Taste perception. 5. Auditory organs: location, function of parts the middle, inner and external ear. 6. Processes of equilibrium. 7. Function of visual organs: • visual accessory organs • structure of the eye light refraction • visual nerve pathways 8. Terms related to senses: • accommodation • ampulla • chemoreceptor • cochlea • cornea • dynamic/static equilibrium • labyrinth • macula • optic • photoreceptor • ceptor • proprioreceptor • proprioreceptor • proprioreceptor • proprioreceptor • sclera • thermoreceptor • Test of cranial nerves
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.8: Endocrine system Objectives:	Hrs. theory8Hrs. Practical 1Content:
 Differentiate between endocrine and exocrine glands. Explain how steroid and nonsteroid hormones produce effects on target cells. Discuss how negative feedback mechanisms regulate hormonal secretions. Explain how the nervous system controls hormonal secretions Describe the functions of the hormones secreted by the endocrine glands. 	 Characteristics f the endocrine system. Functions of hormones. Control of hormone secretion. Structures, functions and locations of endocrine glands: pituitary thyroid parathyroid adrenal

	Hrs. theory 6 Hrs. Practical 1			
Unit 2: Systems of the Body Sub-unit 2.10: Cardiovascular system	Hrs. theory Hrs. Practical			
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.			
 Objectives: Describe the characteristics of the blood and tell the functions of blood Differentiate between the different types of blood cells. Explain the interpretation of blood cell counts. Describe the production and control of red blood cells. Tell the components of plasma and the function of each . Define homeostasis and describe how it is maintained. Describe the steps in blood coagulation Discuss factors which increase or interfere with blood coagulation Explain the purpose and process of blood typing. Describe how blood reactions may occur between the fetal and maternal tissues. 	Content:1. Components and function of the blood.2. Differential interpretation of blood counts.3. Role of blood in maintaining homeostasis.4. Production and regulation of blood cells.5. Coagulation factors.6. Blood types and blood reactions.7. terms related to the study of blood:albuminantibodyleukocyteantigenlymphocytebasophilmacrophageembolusmonocyteeosinophilneutrophilerythrocyteglobulin8. WBC Count9. RBC Count10. DL Count11. BT Count12. CT Count13. Blood grouping			
Sub-unit 2.9: Blood	Hrs. theory 4 Hrs. Practical 2			
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical			
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.			
 Explain how the secretion of each hormone is regulated. Distinguish between physical and psychological stress. Describe the general stress response and its effects on the body when stress is continuous 	 thymus ovary/testes/placentapineal Terms related to endocrine system: adrenal cortex adrenal medulla aldosterone anterior pituitary epinepherine catacholamine glucagon luteinizing hormone metabolic rate norepinepherine prolactin prostaglandin steroid thyroxine 			

Objectives:	Content:				
1. Name the organs of the cardiovascular	1. Structures and functions of the heart.				
system and describe their functions	2. Interpretation of heart sounds.				
2. Locate and name the major parts of the heart	3. Locations, functions and characteristics of				
and describe the functions of each.	arteries and arterioles.				
3. Describe the pathway of the blood through	4. Capillaries and their actions.				
the heart and the vessels of the coronary	5. Locations, functions and characteristics of				
circulation.	veins and venules.				
4. Describe the cardiac cycle and tell how it is	6. Regulation of blood pressure.				
controlled.	7. Mechanisms of venous flow.				
5. Mention the functions of the chief blood	8. Names, functions and locations of the				
vessels of the body	vascular components of the pulmonary and				
6. Compare the structures of the chief blood	systemic circulatory systems.				
vessels of the body	9. Terms related to circulation:				
7. Explain the mechanisms that aid in	• arteriole				
returning venous blood to the heart.	• atrium				
8. Explain how blood pressure is produced and	• cardiac cycle				
controlled.	• cardiac output				
9. Compare the pulmonary and systemic	• diastolic pressure				
pathways of the cardiovascular system	• electrocardiogram				
	• myocardium				
	• pericardium				
	 peripheral resistance 				
	 sphygmomanometer pacemaker 				
	 systolic pressure 				
	 vasoconstriction 				
	 vasocinsuletion vasodilation 				
	vasounation ventricle				
	• venule				
	viscosity				
	Precordium examination normal				
	sound				
	Murmur				
	Measurement of pulse and blood pressure				
	pressure				
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources:				
	classroom instruction, models, charts.				
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical				
Sub-unit 2.11: Lymphatic system and	Hrs. theory 3 Hrs. Practical 1				
immunity					
Objectives:	Content:				
1. Describe the functions of the lymphatic	1. Patterns of Lymphatic movement.				
system and locate the chief lymphatic	2. Lymph node location, function and				
pathways.	structure.				
2. Explain how lymphatic circulation is	3. Functions of the thymus and spleen.				
maintained.	4. Specific and nonspecific defenses against				
3. Locate the chief lymph nodes and describe	infection.				
their functions.	5. Immunity and allergic reactions.				
4. Differentiate between specific and nonspecific	6. Disorders with autoimmune origins:				
immunity and provide examples of each	7. Type 1 Diabetes				
	 Lupus erythematosus 				

5.	Describe the function of lymphocytes and		• R1	neumatoi	d arthritis	
	immunoglobulins.			leroderm		
6.	Differentiate between active and passive		• M	ultiple sc	lerosis	
	immunity.			hizophre		
7.	Distinguish between primary and secondary	8.	Terms rela	ated to ly	mphatics and imm	nune
	immune responses.		system:			
8.	Explain how allergic reactions, tissue			lergen		
	rejection reactions, and autoimmunity are related to immune mechanisms.			ntibody		
9.	Describe the disorders believed to be caused			ntigen		
<i>.</i>	by an autoimmune reaction			terferon		
	5		•	mphocyte		
				acrophag	e	
			-	athogen		
E	1	T .		accine	·····	
Eva	aluation methods: written and viva exams.				tivities and resour nodels, charts.	ces:
				ruction, i		
	it 2: Systems of the Body	-	s. theory	(Hrs. Practical	1
	b-unit 2.12: Digestive system	-	rs. theory ntent:	6	Hrs. tutorial	1
	jectives: Locate and describe the functions chief	1.		and fund	tions of the alima	ntom
1.	organs of the digestive system.	1.	canal.	and func	tions of the alime	mai y
2.	Describe how the contents of the alimentary	2.	•••••••	t and ene	rvation f the alime	entarv
	canal are moves and mixed.	2.	canal.	e una ente		Jiitui j
3.	Describe the functions of enzymes secreted	3.	Mouth str	uctures ai	nd functions.	
	by various digestive organs and glands .	4.	Pharynx a	nd esoph	agus structure and	l
	describe the function of each.		function.			
4.	List the enzymes of secreted by various	5.			and absorption.	_
	digestive organs and glands .	6.			ions of the pancre	as and
5.	Describe how directive secretions are	7.	liver in reg		ion of the small a	ad larga
5.	Describe how digestive secretions are controlled.	/.	intestines.			lu laige
6.	Discuss how digestive reflexes control	8.			e digestive system	:
0.	movement of material through the alimentary	0.		sorption		
	canal.			al canal		
7.	Describe the mechanisms of swallowing,		• bi	le		
	vomiting and defecating.		• ch	nyme		
8.	Explain how the products of digestion are		• de	eciduous		
	absorbed.			ıodenum		
				nulsificat	ion	
				ces		
				junum		
				um		
				esentery	1	
				ucous me		
			-	nacreatic	Juice	
			-	eristalsis	inctor	
				/loric sph ctum	metor	
				ohincter n	nuscle	
					appendix	
				lli/villus	uppendix	
L			• VI	m viilus		

	• Vomiting, Diarrhoea, Constipation
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.13: Respiratory system	Hrs. theory 7 Hrs. Practical 1
Objectives:	Content:
 Describe the functions of the respiratory system Locate the organs of the respiratory system Explain how inspiration and expiration are achieved. Describe the respiratory cycle and define the related terms: tidal volume, inspiratory reserve, expiratory reserve, residual volume, vital capacity, inspiratory capacity, functional residual capacity, total lung capacity. Locate the respiratory center and explain how it controls normal breathing. Describe the various factors which affect the respiratory center. Describe the functions of the respiratory membrane. Explain how oxygen and carbon dioxide are transported in the blood. Describe the process of cellular respiration. Explain how cells use oxygen 	 Organs of the respiratory system. Mechanisms of breathing and control of breathing. Alveolar gas exchanges. Gas transport. Terminology related to respiration: alveolus bronchial tree diaphragm glottis intercostal muscles hilus hyperventilation oxyhemoglobin parietal pleura partial pressure pleural cavity respiratory wolume surface tension surfactant visceral pleura Breath sound (Bronchial and vesicular Measurement of Respiratory rate
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.14: Urinary system	Hrs. theory 4 Hrs. Practical 1
Objectives:	Content:
 Locate the organs of the urinary system and describe their general function Describe the structure and functions of the 	 Location, structure and function of the organs of the urinary system. Percel simulation
2. Describe the structure and functions of the kidneys.	 Renal circulation. Processes and regulation of urine formation:
 Describe the pathway of blood through the kidneys. Describe a nephron and explain the function 	 Frocesses and regulation of unite formation. glomerular filtration tubular reabsorption concentration and volume
of each parts .	7. Formation of urea, ureic acid.

reproductive organs.	organs: penis, testes, scrotum.
Objectives: 1. Describe the structure of the male	Content: 1. Structure and function of the external
Sub-unit 2.16: Male reproductive system	Hrs. theory4Hrs. tutorial1Content:
Unit 2: Systems of the Body	Hrs. theoryHrs. PracticalHrs. theory4Hrs. tutorial1
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
 Explain the meaning of acid-base balance within the body. Discuss the regulation of how electrolytes enters and leaves the body. Dercribe where hydrogen ions come from within the body. Describe the action of the body's chemical buffer systems, respiratory center, and the kidneys in regulating acid-base balance. 	 phosphate buffer system protein buffer system extracellular intracellular transcellular osmoreceptor electrolyte balance
 Describe how water intake and output are regulated by the body systems. List the important electrolytes of the body. List the ways electrolytes enter and leave the body Explain the meaning of acid-base balance 	 6. Terms related to water and electrolyte balance: acidosis alkalosis bicarbonate buffer system
differs between compartments, now fluid composition differs between compartments, and how fluids move from one compartment to another.3. List the routes by which water leaves and enters the body.	 Mechanisms of water balance and regulation. Mechanisms of electrolyte balance and regulation. Regulation of hydrogen ion concentration.
 Discuss the importance of water and electrolyte balance within the human body. Describe how body fluids are distributed within compartments, how fluid composition 	 Composition of body fluids. Distribution and movement of fluids between compartments. Mechanisms of water balance and
balance Objectives:	Content:
Sub-unit 2.15: Water, electrolyte & pH	Hrs. theory6Hrs. Practical1
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
	autoregulationdestusor muscle
	 renal corpuscle renal tubule renal plasma threshold retroperitoneal
 Describe the process of micturation and tell how it is controlled 	 efferent arteriole juxtaglomerular apparatus nephron loop pertibular capillary
secretion, in the production of urine8. Describe the functions of the ureters, urinary bladder, and urethra.	glomerulusafferent arteriole
 Discuss the regulation of glomerular filtration and factors that may affect this. Describe tubular reabsorption and tubular 	 10. Terms related to the urinary system: renal cortex renal medulla
5. Explain how glomerular filtrate is produced and state it's components.	 8. Tubular secretion and urine composition. 9. Elimination of urine.

 State the general functions of the male reproductive system. Name the parts of the male reproductive system and describe the general functions of each. Describe the process of spermatogensesis. Describe the path of sperm sells from their origin to their exit from the body Describe the structure of penis and explain the mechanism of erection and ejaculation. Explain how hormones control the activities of the m male reproductive organs and the development of male secondary sexual characteristics. Describe how a vasectomy is performed, and discuss the relative simplicity of this procedure. 	 2. Formation and release of sperm cells. 3. Actions of male sex hormones. 4. Terms related to male reproductive system: glans penis prepuce corpora cavernosa corpus cavernosa spermatogenesis semen inguinal gonadotropin testosterone Male contraceptive devices.
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.
Unit 2: Systems of the Body	Hrs. theory Hrs. Practical
Sub-unit 2.17: Female reproductive system	Hrs. theory 4 Hrs. tutorial 1
Objectives:	Content:
 Name the parts of the female reproductive system and describe the general functions of each. Describe the process of oogenesis. Describe how the hormones control the activities of the female reproductive system and the development of female secondary sexual characteristics. Describe the process of the menstrual cycle. Describe the hormonal changes that occur in the maternal body during pregnancy. Describe the birth process and explain the role of hormones in this process. Explain why females are more easily infected by sexually transmitted diseases than men, given equal exposure. 	 Structure and function of the vagina, clitoris, labia, ovaries, fallopian tubes, uterus, breasts and mammary glands. Ova development and ovulation. Hormonal control of the reproductive system. Fertilization and embryonic development. Pregnancy changes. Process of childbirth and physiological recovery. Structure and function of the mammary glands. Terms related to the female reproductive system: follicle estrogen progesterone fertilization meiosis oogenesis zygote implantation infundibulum orgasm ovulation menstrual cycle puberty and menarche placenta menopause Female contraceptive devices.

Evaluation methods: written and viva exams.	Teaching / Learning activities and resources:			
	classroom instruction, models, charts.			
Unit 3: Human Growth & Development	Hrs. theory 5 Hrs. Practical 1			
 Objectives: Distinguish between growth and development. Describe the formation and function of the placenta. Describe the path of blood through the foetal circulatory system. Describe the chief circulatory and physiological adjustments that occur in the newborn. List the stages of development that occur between the neonatal period and death, and tell the general characteristics of each stage. 	Content: Foetal circulation and neonatal changes. 1. Characteristics of the stages of life development:			
Evaluation methods: written and viva exams.	Teaching / Learning activities and resources: classroom instruction, models, charts.			

1. Textbook of practical Physiology, GK PAL, PRAVATI PAL, Orient Longman.

Physics

Year: First Level: Certificate

Credit Hours: Theory 160 Practical: 80 Assessment Marks: 100

Course Description

This course in physics is designed to provide students with an understanding of the scientific laws of our physical world, and how physics contributes to life's activities in modern society. The course emphasizes both quantitative and qualitative aspects of physics, involving mathematical models and equations. The application of physics to social and environmental situations is well illustrated.

The practical component of this course is designed to supplement learning through the application of learned theory. The students will handle simple apparatus to do simple measurements, demonstrate simple electrical circuits, and apply their knowledge of physics to real life examples.

Course objectives

On completion of the course the student will be able to:

- Correlate physics and its applications related to everyday experiences of their life.
- Identify the social, economic, environmental and other implications of physics.
- Describe physics as a coherent and developing framework of knowledge based on fundamental theories of the structures and processes of the physical world.
- Demonstrate the skills of experimenting, observing, interpreting data and evaluating evidence to formulate generalizations and models.
- Apply knowledge of physical principles to familiar and unfamiliar situations.
- Apply facts, vocabulary and conventions to unit measurements and common measuring instruments.
- Explain the definitions, laws, concepts, theories and models presented in this course.
- Describe the applications and implications of physical facts and principles.

Evaluation methods: written and viva exams, performance observation.

Teaching / Learning activities and resources: classroom instruction and demonstration, return demonstration, models, solving related problems.

Recommended Texts

- 1. Brij Lal and Subramanyan, Principles of Physics.
- 2. Nelkon and Parker, Advanced Level Physics (5th ed.)
- 3. Physics Practical Manual, Basanta Raj Rosyra (second edition)

Reference Texts

- 1. Pradhan, J.M. & Gupta, S.K., A Textbook of Physics (part I & II)
- 2. Verma, H.C., Concepts of Physics I & II
- 3. Sears, Zemansky & Young, University Physics
- 4. Halliday, D & Resnick, R., Physics Part I & II

Co	urse: Physics	Hrs. theory 160 Hrs. lab 80
Un	it 1: Mechanics	Hrs. theory 40 Hrs. lab
Su	b-unit 1.1: Units and Measurement	Hrs. Theory 4 Hrs. lab
Ob	jectives:	Content:
1. 2. 3. 4. 5. 6. 7.	Define fundamental and derived units. Explain, MKS, CGS and SI system of units. Convert one system of units into another system of units. Express derived units in terms of fundamental units. Define precise and accurate measurement Use of dimensions to derive simple physical quantities and equations (time period of simple pendulum) Convert one system of units into another using dimensional formula	 Physical concept of mass, length and time. Various systems of units and their conversion Express derived units in terms of fundamenta units. Precise and accurate measurement Dimensional formula for various physical quantities. Conversion of system of units using dimensions Solve simple numerical problems
	it 1: Mechanics	··· ·· ·· ·· ·· ·· ·· ·· ·· ·· ·· ·· ··
	b-unit 1.2: Scalar and Vectors	Hrs. theory 5 Hrs. lab
	jectives:	Content:
1. 2.	Differentiate between scalars and vectors Identify whether a physical quantity is scalar or vector.	 Scalar and vectors with examples. Vector addition by parallelogram and triangle method.
3.	Resolve vectors into two rectangular components.	 Resolve a vector into two components. Triangle and parallelogram law of vectors
4.	State and explain triangle and parallelogram law of vectors	5. The product of two vectors either results in a scalar quantity or a vector quantity.
5. 6.	Point out the resultant of two or more vectors by graphical method. Write the values of scalar product and vector	6. Simple numerical problems
I.I.	product, for selected problems. it 1: Mechanics	
		Hardbarr 5 Harlah
	b-unit 1.3: Kinematics	Hrs. theory 5 Hrs. lab
	jectives: Define displacement, velocity, instantaneous velocity, average velocity, uniform velocity and acceleration retardation. Differentiate between distance and	 Content: Displacement, velocity, instantaneous velocity, average and uniform velocity and acceleration (retardation). Distance and displacement, speed and
3.	displacement, speed and velocity. Write down the relation of kinematics equation of motion (linear and gravitational).	velocity.3. The concept of projectile motion (Show that path of the projectile is parabolic)
4.	Calculate the time of flight, maximum height and horizontal range of a projectile (Both cases)	 Solve simple numerical problems
5.	Solve simple problems related to the projectile.	
Ev	aluation methods: written and viva exams, formance observation.	Teaching / Learning activities and resources: classroom instruction and demonstration, return
T -		demonstration, models, solving related problems
	it 1: Mechanics	
	b-unit 1.4: Force	Hrs. theory 9 Hrs. lab
C1	jectives:	Content:

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1.	State Newton's laws of motion.	1. Linear momentum and significance of
2.	Give the concept of inertia of rest, motion and	Newton's laws of motion in various
3.	direction.	concepts.
4.	Define force in terms of rate of change of	2. Interpret the meaning of inertia of rest and
	momentum and give their directions.	inertia of motion.
5.	Derive $F = ma$ and used it to solve simple	3. Illustrate the applications of inertia and
	problems.	impulse.
6.	Recognize the impulse is a force act in very	4. Angular displacement, velocity and
	short interval of time.	acceleration
7.	State and prove principle of conservation of	5. Derive the relation $v = \omega r$.
/ .	linear momentum with examples.	 Berryc the relation v = on. Recall vector nature of velocity and change
8.	Define angular displacement, angular velocity	
0.		the direction of velocity in circular motion.
	& angular acceleration.	7. Know the magnitude of centripetal force and
9.	Distinguish between angular velocity and	8. centrifugal force, $F = mv^2/r = mr\omega^2$ (With
	linear velocity and derive relation between	derivation)
	them.	9. Friction, limiting friction, angle of friction
10.	Define circular motion, centripetal force, and	and
	centrifugal force.	coefficient of friction.
11.	State the magnitude and direction of centripetal	10. State law of limiting friction.
	and centrifugal force and their applications to	11. Derive the relation between angle of friction
	centrifuge and satellite (not derivation).	and coefficient of friction.
12.	Differentiate between elastic and inelastic	12. Simple numerical problems
	collision.	
13.	Define friction, laws of limiting friction, angle	
	of friction, angle of repose e of repose and	
	coefficient of friction.	
1		
Un	it 1: Mechanics	
-		Hrs. theory 4 Hrs. lab
Sub	it 1: Mechanics	Hrs. theory 4 Hrs. lab Content:
Sub Ob	it 1: Mechanics p-unit 1.5: Work, Energy and power jectives:	Content:
Sub	it 1: Mechanics unit 1.5: Work, Energy and power jectives: Define work energy and power and give their	Content: 1. The distinction between the common uses of
Sub Obj 1.	it 1: Mechanics 	Content:1. The distinction between the common uses of the term work, energy and power and its
Sub Obj 1. 2.	it 1: Mechanics 	 Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics.
Sub Obj 1.	it 1: Mechanics 	 Content: The distinction between the common uses of the term work, energy and power and its meaning in Physics. Conservation of energy i.e. change of KE into
Sub Obj 1. 2. 3.	it 1: Mechanics 	 Content: The distinction between the common uses of the term work, energy and power and its meaning in Physics. Conservation of energy i.e. change of KE into PE giving example of falling body.
Sub Obj 1. 2. 3.	it 1: Mechanics 	 Content: The distinction between the common uses of the term work, energy and power and its meaning in Physics. Conservation of energy i.e. change of KE into PE giving example of falling body. Give the transformation of different forms of
Sub Obj 1. 2. 3.	it 1: Mechanics 	 Content: The distinction between the common uses of the term work, energy and power and its meaning in Physics. Conservation of energy i.e. change of KE into PE giving example of falling body. Give the transformation of different forms of energies i.e. PE into KE etc.
Sut Obj 1. 2. 3. 4.	it 1: Mechanics 	 Content: The distinction between the common uses of the term work, energy and power and its meaning in Physics. Conservation of energy i.e. change of KE into PE giving example of falling body. Give the transformation of different forms of
Sub Obj 1. 2. 3. 4. Un	it 1: Mechanics 	 Content: The distinction between the common uses of the term work, energy and power and its meaning in Physics. Conservation of energy i.e. change of KE into PE giving example of falling body. Give the transformation of different forms of energies i.e. PE into KE etc. Simple numerical problems
Sut Obj 1. 2. 3. 4. Un Sut	it 1: Mechanics 	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab
Sub Obj 1. 2. 3. 4. Un Sub Obj	it 1: Mechanics 	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. theory 5 Hrs. lab Content:
Sut Obj 1. 2. 3. 4. Un Sut Obj 1.	it 1: Mechanics 	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² .
Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2.	it 1: Mechanics 	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight
Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2.	it 1: Mechanics unit 1.5: Work, Energy and power jectives: Define work energy and power and give their units in various systems. Define KE and PE also give their magnitude. State and verify the principle of conservation of energy. Give examples to demonstrate the uses of the transfer of energy. it 1: Mechanics unit 1.6: Gravity and Gravitation jectives: State Newton's law of gravitation. Deduce unit and dimension of G. Define acceleration due to gravity and	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and
Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3.	it 1: Mechanics 	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and acceleration due to gravity.
Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. 4.	it 1: Mechanics 	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and acceleration due to gravity. 4. The variation of g due to height and depth.
Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. 5.	it 1: Mechanics unit 1.5: Work, Energy and power jectives: Define work energy and power and give their units in various systems. Define KE and PE also give their magnitude. State and verify the principle of conservation of energy. Give examples to demonstrate the uses of the transfer of energy. it 1: Mechanics unit 1.6: Gravity and Gravitation jectives: State Newton's law of gravitation. Deduce unit and dimension of G. Define acceleration due to gravity and variation of g due to height and depth Differentiate between mass and weight. Explain weightlessness condition in lift	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and acceleration due to gravity. 4. The variation of g due to height and depth. 5. Center of mass and center of gravity.
Sut Ob 1. 2. 3. 4. Un Sut Ob 1. 2. 3. 4. Un Sut Ob 1. 2. 3. 4. 5. 6.	it 1: Mechanics unit 1.5: Work, Energy and power jectives: Define work energy and power and give their units in various systems. Define KE and PE also give their magnitude. State and verify the principle of conservation of energy. Give examples to demonstrate the uses of the transfer of energy. it 1: Mechanics unit 1.6: Gravity and Gravitation jectives: State Newton's law of gravitation. Deduce unit and dimension of G. Define acceleration due to gravity and variation of g due to height and depth Differentiate between mass and weight. Explain weightlessness condition in lift	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and acceleration due to gravity. 4. The variation of g due to height and depth. 5. Center of mass and center of gravity. 6. Conditions of equilibrium of a body with
Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. Un Sut Obj 1. 2. 3. 4. 5.	it 1: Mechanics unit 1.5: Work, Energy and power jectives: Define work energy and power and give their units in various systems. Define KE and PE also give their magnitude. State and verify the principle of conservation of energy. Give examples to demonstrate the uses of the transfer of energy. it 1: Mechanics unit 1.6: Gravity and Gravitation jectives: State Newton's law of gravitation. Deduce unit and dimension of G. Define acceleration due to gravity and variation of g due to height and depth Differentiate between mass and weight. Explain weightlessness condition in lift State the condition of equilibrium of a body Differentiate between center of gravity and	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and acceleration due to gravity. 4. The variation of g due to height and depth. 5. Center of mass and center of gravity. 6. Conditions of equilibrium of a body with examples.
Sut Ob 1. 2. 3. 4. Un Sut Ob 1. 2. 3. 4. Un Sut Ob 1. 2. 3. 4. 5. 6.	it 1: Mechanics unit 1.5: Work, Energy and power jectives: Define work energy and power and give their units in various systems. Define KE and PE also give their magnitude. State and verify the principle of conservation of energy. Give examples to demonstrate the uses of the transfer of energy. it 1: Mechanics unit 1.6: Gravity and Gravitation jectives: State Newton's law of gravitation. Deduce unit and dimension of G. Define acceleration due to gravity and variation of g due to height and depth Differentiate between mass and weight. Explain weightlessness condition in lift	Content: 1. The distinction between the common uses of the term work, energy and power and its meaning in Physics. 2. Conservation of energy i.e. change of KE into PE giving example of falling body. 3. Give the transformation of different forms of energies i.e. PE into KE etc. 4. Simple numerical problems Hrs. lab Content: 1. Laws of gravitation F = GMm/R ² . 2. Acceleration due to gravity, mass and weight 3. The relation between gravitation constant and acceleration due to gravity. 4. The variation of g due to height and depth. 5. Center of mass and center of gravity. 6. Conditions of equilibrium of a body with

Unit 1: Mechanics			
Sub-unit: 1.7 Properties of Matter	Hrs. theory 5		Hrs. lab
Objectives:	Content:		
 Define elasticity, stress, strain and elastic limit on the basis of Hook's law Write relation for energy stored in a stretched wire and energy density Define surface tension. Differentiate adhesive and cohesive force. Define viscosity of liquid. Describe how the height of liquid rises in a capillary tube of sufficient and insufficient length. 	strain and ela2. Elastic potenti a stretched win	asticity of so ial energy as re(without of of surface te cohesive fo action. fluid movem	nd energy density in lerivation) ension of liquid. rces. ment
Unit 1: Mechanics			
Sub-unit 1.8: Hydrostatics	Hrs. theory 3		Hrs. lab
Objectives:	Content:		
 Demonstrate that fluid pressure acts in all directions Explain that liquid pressure is proportional to the depth of the liquid and independent of the shape of the vessel. Define density, relative density and specific gravity of solids and liquids. Upthrut, Archimedes's principle. Apply Archimedes's principle to determine the specific gravity of various solids and liquids. State the principle of flotation & condition of equilibrium of floating bodies. Explain how barometers works Describe how atmospheric pressure affects human body functions. 	 Difference bet gravity. Archimedes's p Design equipm principle. The principle of equilibrium fo Atmospheric p Introduction of The effect of a Simple numerical 	gh. ve density a tween densi principle ar nent to veri of floatation or floatation floating b pressure wit of Mercury b air pressure ical probler	and specific gravity. ty and specific od its uses. fy Archimedes's n and condition of odies. h examples. barometer on human body. ns
Unit 2: Heat	Hrs. theory	22	Hrs. lab
Sub-unit 2.1: Thermometry Objectives:	Hrs. theory Content:	2	Hrs. lab
 Define heat and temperature. Distinguish between heat and temperature. Explain sensitivity of liquid thermometers Explain the operation and use of a thermometer. Determine the lower and upper fixed points of the thermometer. Define different temperature scales (Celsius, Fahrenheit and Kelvin) Convert one temperature scale into another. Use the temperature conversion formula to convert and solve numerical problems related to it 	 Concept of hea Factors on white Demonstrate value and explain the Derivation of the C/5 = [F - 32] 	ich sensitiv various type neir uses. the formula /9 = [K - 2] een differen	ity depends s of thermometers 73] tt temperature scales.
Unit 2: Heat Sub-unit 2.2: Expansion	Hrs. theory	6	Hrs. lab

Ob	jectives:	Content:
	Describe linear, superficial and cubical	1. Linear, superficial and cubical expansion of
	expansion of solids and their expansivity.	solids.
2.	Derive the relation between linear, superficial	2. The relations $l_2 = l_1[1 + \alpha (\theta_2 - \theta_1)]$,
	and cubical expansivity of solids	$A_{2} = A_{1}[1 + \beta(\theta_{2} - \theta_{1})], V_{2} = V_{1}[1 + \gamma(\theta_{2} - \theta_{1})]$
3.	Define real and apparent expansion of liquid.	θ_1
4.	Explain the change in density of a substance	/ -
	with the variation temperature.	3. Derivation of $\gamma = 3\alpha$ and $\beta = 2\alpha$.
5.	Discuss the density variation of water with	4. Apparent and real expansion of a liquid and its relation
	temperature (anomalous properties of water).	
6.	Discuss the concept of water therapy due to	5. Change in density of an object due to change
	high specific heat capacity of water.	in temperature.
	ingli specific from cupacity of materia	6. Anomalous expansion of water and its
		importance to marine life.7. Why water is used for cooling and heating
TT	··· • • • • • • • • • • • • • • • • • •	purposes.
	it 2: Heat p-unit 2.3: Calorimetry	Hrs. theory 6 Hrs. lab
	jectives:	Hrs. theory6Hrs. labContent:
	Define heat capacity, specific heat capacity.	1. Heat capacity, specific heat capacity.
1. 2.	Distinguish between joule and calorie as heat	 Preat capacity, specific field capacity. Give the relation between joule and calorie.
2.	unit.	 3. Melting point, boiling point and freezing point
2	Understand the quantity of heat content of a	of a substance.
5.	body $Q = ms\theta$.	4. The effect of pressure on melting and boiling
4.	Explain the energy required to cause a phase $\sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{i=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{j=1}^{n} \sum_{i=1}^{n} \sum_{i=1}$	point of substance
4.	change at constant temperature.	5. Determination of latent heat of fusion of ice
5		and latent heat of steam by the method of
5.	Define freezing, melting and boiling point of a substance	mixture.
6		6. Simple numerical problems
6.	Explain latent heat of fusion and latent heat of vaporisation.	0. Shiple humerical problems
7	Discuss the effect of pressure on melting and	
7.	boiling point of the substance.	
Un	it 2: Heat	
	p-unit 2.4: Hygrometry	Hrs. theory 3 Hrs. lab
	jectives:	Hrs. theory 3 Hrs. lab Content: 3
1. 2	Define saturated and unsaturated vapours. Differentiate between SVP and USVP.	 Learner will become knowledgeable about: Saturated and unsaturated vapours.
2.		 Saturated and unsaturated vapours. Saturated VP and unsaturated VP.
3.	Draw P-V and P-T diagrams and explain the	
1	behaviours of vapours.	8
4.	Discuss the effect of pressure and altitude on the heiling point of a liquid	behaviours of vapours.
5	the boiling point of a liquid.	5. The effect of pressure and altitude on the boiling point of a liquid.
5.	Explain the term due point, absolute humidity and relative humidity.	
6	5	
6.	Demonstrate the wet and dry bulb hygrometer and describe its use to determine the relative	$\frac{\text{Partial vapour pressure of water}}{\times 100}$
		vapour pressure of water %
	humidity	7. Wet and dry bulb hygrometer and relative
		humidity.
Un	it 2: Heat	
	b-unit 2.5: Transfer of heat	Hrs. theory 5 Hrs. lab
Sur		

 Differentiate between conduction, convection and radiation. Define thermal conductivity with its unit and dimension. Distinguish between good and bad conductors of heat. Define black body and black body radiation. Explain transmission of heat by conduction, convection and radiation with appropriate application to medical field and daily use. Define black body. State and explain Stefan Boltzmann's law and give an example of its application. Describe medical uses of thermal radiation. 	 The transfer of heat by conduction, convection and radiation. Thermal conductivity giving their dimension and units. Laws of black body radiation. Medical uses of heat radiation(thermal therapy) Solve simple numerical problems
Unit 3: Light	Hrs. theory 18 Hrs. lab
Sub-unit 3.1: Reflection of light	Hrs. theory 6 Hrs. lab
Objectives:	Content:
 Explain the laws of reflection of light. Find the deviation of light by plane mirror as rotating mirror. Distinguish between real and virtual image. Show that in plane mirror object distance = image distance. Define the terms pole, center of curvature, radius of curvature, principal focus, principal axis, focal length. Show that r = 2f for spherical mirrors. Draw ray diagrams to solve problems involving spherical mirrors. Derive the formula 1/u + 1/v = 1/f 	 The phenomenon of reflection and hence state thelaws of reflection of light. Principles of reflection of light – The rotation of mirror through angle θ the reflected ray is rotated through 20. Object distance is just equal to image distance i.e. u = v but the image is virtual. Real and virtual image. Image formation of spherical mirror. How to correct sign for the focal length, object distance and image distance. The relation, r = 2f, 1/u + 1/v = 1/f and I/O = v/u = m for mirrors. Nature, size and position of the image formed by spherical mirrors at various positions of the object distance on the principal axis. Simple numerical problems
Unit 3: Light	· · · · · · · · · · · · · · · · · · ·
Sub-unit 3.2: Refraction	Hrs. theory 7 Hrs. lab
Objectives:	Content:
 State and explain the laws of refraction of light. Verify the laws of refraction of light and define refractive index in different media. 	 Phenomenon of refraction. Refractive index in terms of the speed of light in vacuum to the speed of light in medium. The relations _aµ^g×_gµ^w×_wµ^a = 1.
 Derive the expression for apparent depth and lateral shift in a glass slab. 	 The relations aµ^{2×}gµ ×_wµ - 1. Refractive index in terms of real depth and apparent depth.

4.	Define critical angle and total internal reflection.	1	The relation $t[sin(i - r)]$	· · ·	I/μ) and lateral shift p =
5.	Explain the phenomenon of total internal				ula $\mu = 1/sin(C)$
5.	reflection.				ditions for total internal
6.	Explain the passage of light rays through a		reflection.		
0.	prism.			f total inter	mal phenomenon,
7.	Derive the formula $i + e = A + \delta$ and		mirage, lig		nai phenomenon,
1.	$(A = r_1 + r_2)$				te the deviation of light
8.	Define minimum deviation and derive the	1	ray in prisn		
	formula $\mu = \sin[(A + \delta_m)/2]/\sin(A/2)$			$a A + \delta = i$	+ e and
9.	Define the terms convex lens, image in lens,	10.			$(2)/\sin(A/2)$
1	optical center, and thin lens.	11		ferent types	
10.	Draw a ray diagram to locate positions of				convex lens and
	image in thin lenses (concave and convex).			spect of con	
11.	Derive lens formula and lens maker's formula.				e image formation by
				well as conc	
					maker's formula.
				nerical prob	
Un	it 3: Light		•		
	b-unit 3.3: Optical Instrument	Hrs	theory	5	Hrs. lab
-	jectives:		tent:		
1.	Draw a labeled diagram of human eye.	1.	Structure o	f human ey	e with diagram.
2.	Explain the eye as an instrument, which forms		The "mode		-
	as sharp image on the retina.	3.	Ray diagra	m to explain	n the correction of
3.	Explain the terms far point, near point, and		defect of vi		
	least distance of distinct vision.				pound microscopes.
4.	Define the terms visual angle and angular				nifying power of
	magnification.	1		.	microscopes.
5.	Explain the technique of removing the defect			of light by	
	of vision.				ation of refractive index
6.	Trace the path of rays through simple and		with colour		1
7	compound microscopes.	8.	Simple nur	nerical prob	biems
7.	Explain how white light is a combination of				
	seven different colours, easily decomposed into				
8.	its components. Understand that refractive index varies with				
0.	colours.				
9.	Demonstrate the dispersion of light by prism.				
	it 4: Waves and Sound	Hre	theory	10	Hrs. lab
	b-unit 4.1: Waves		theory	5	Hrs. lab
-	jectives:	1	tent:	0	111 5, 1417
1.	Define transverse, longitudinal, progressive	-		of progres	sive and stationary
1.	and stationary waves with examples.		waves	or progres	sive and stationally
2	Define amplitude, wavelength, frequency,			w in a way	
2.					
2	period andvelocity of the wave.			tion of way	
3.	Describe how a wave carries only energy			-	n, diffraction, and
	from onepoint to another and no material			ce of wave	
	particle is transmitted in the wave motion.	5.	Simple nu	merical pr	oblems
	Show that a wave undergoes reflection	1			

refraction, interference and diffraction	
phenomena	
Unit 4: Waves and Sound	
Sub-unit 4.2: Characteristics of Sound Waves	Hrs. theory 5 Hrs. lab
Objectives:	Content:
 Differentiate between noise and music. Explain the characteristic of musical sound. Define the terms sonic (audible), infrasonic, ultrasonic and super sound. Define beats and write down beats formula using superposition of waves. Describe how intensity of sound is proportion to the square of amplitude. Define intensity level, bel and decibel. Explain the threshold of hearing and threshol of pain. Explain ultrasonic waves and its medical use Explain the evidence that sound waves in an are longitudinal waves. Explain how air undergoes compression and rarefaction as sound waves travels through tair. 	 2. Qualitative relations of pitch with frequency, intensity with loudness and overtones with quality of sound. 3. Beat and beat frequency. 4. Intensity level in terms of decibel. 5. Threshold of hearing and threshold of pain. 6. Ultrasonic wave and its medical uses. 7. V=√E/p 8. Simple numerical problems
11. $V=\sqrt{E/p}, V\alpha \sqrt{T}, V\alpha \sqrt{1/M}$	
Unit 5: Electrostatics	Hrs. theory 10 Hrs. lab
Sub-unit 5.1: Fundamentals of electrostatics	Hrs. theory 3 Hrs. lab
Objectives:	Content:
 Explain the properties of electrical charges. Distinguish between conductor, insulator, ar semiconductor. Explain the phenomenon of charging by friction, conduction and induction. Describe the surface charge density on vario conductors 	induction on the basis of modern theory.3. Surface charge density4. Simple numerical problems
Evaluation methods: written and viva exams, performance observation.	Teaching / Learning activities and resources: classroom instruction and demonstration, return demonstration, models, solving related problems.
Unit 5: Electrostatics	
Sub-unit 5.2: Electrostatic Field	Hrs. theory 7 Hrs. lab
 Objectives: State and explain Coulomb's law. Explain the properties of lines of force Define electric field and electric flux. Calculate electric field intensity due several point charges. Define electric potential difference, potentia 	3. Electric field and normal electric flux.

	plain series and parallel grouping of pacitors	9.	Simple nur	merical pro	blems
	tion methods: written and viva exams, nance observation.	clas	ssroom inst	ruction and	vities and resources: l demonstration, return olving related problems.
Unit 6:	Magnetism		s. theory	10	Hrs. lab
	it 6.1: Fundamentals of Magnetism	Hrs	s. theory	6	Hrs. lab
Object	ives:	Co	ntent:		
 Ex ma Ex ma Sta De Ca Ca De Ca Tra pro Ex Ex De De Con De Con Con De Con Con Con Con 	plain magnetic field strength, lines of force, agnetic field intensity, and permeability. Ate Coulomb's law for magnetism. Assoribe the properties of a magnet. Iculate magnetic field intensity due to a bar agnet at any point on the equatorial and axial e of a bar magnet. plain Tangent law of magnetism ace the lines of force and describe their operties. An entral point. Magnetism Dit 6.2: Terrestrial Magnetism	2. 3. 4. 5. 6. 7. 8. Hrs Cor 1. 2.	other. Various tyj of poles. Coulomb's Magnetic f (a) end on Lines of fo neutral poi Uniform an Simple num s. theory ntent: Dip, decliric	pes of mag law for mag ield intens position (l rce around nt. nd nonunif merical pro 4 ation, hori ts of earth's of dia, para omain the	ity due to bar magnet at b) broad side on position. a bar magnet and the form magnetic field bblems Hrs. lab Zontal and vertical s magnetic field. a and ferromagnetic ory
Unit 7:	Current Electricity	Hrs	s. theory	20	Hrs. lab
	it 7.1: Electric current		s. theory	7	Hrs. lab
Object			ntent:		
 Sta De Lis con Dia con Fin and Per vol 	scuss current as the rate of flow of charge. ate and verify Ohm's law. fine resistance and resistivity. st the factors that influence resistance of a nductor. stinguish between Ohmic and non-Ohmic nductors. and the equivalent resistance from the series d parallel combination of resistors. rform the conversion of galvanometer into ltmeter and ammeter. Current Electricity	1. 2. 3. 4. 5. 6. 7. 8. 9.	Potential d Ohm's law Expression $1/R = 1/R_1$ parallel con Conversion and voltme Ohmic and curve.	ifference. and its ver $R = R_1 + 1$ $+ 1/R_2 + 1/2$ mbination. n of a galva eter. non-Ohm pes of elecc	$R_2 + R_3 + \dots$ and / $R_3 + \dots$ in series and anometer into ammeter ic conductors from I-V trical circuits.
	it 7.2: Resistance and heat	Hrs	s. theory	5	Hrs. lab
Object			ntent:		
1. Sta	ate and explain Joule's laws of heating. stinguish between potential difference and	1.			g and derivation of the

3.	Relate emf, terminal potential and internal	2. Heat production in resistance wire due to
	resistance.	passage of current.
4.	Derive the equivalent emf from series and	3. Electric power in terms of energy dissipated
	parallel and mixed groupings of cells	in a time in the resistance wire.
5.	Define Joule's conversion factor.	4. Meaning of emf and internal resistance of a
		cell.
		5. Relation $E = V + Ir$.
		6. Purpose of grouping of cells to find maximum
		current and maximum voltage.
		7. Electric power, watt, kilowatt, kilowatt-hour
		and horsepower.
		8. Meaning of Joule's conversion factor.
		9. Simple numerical problems
Un	it 7: Current Electricity	
	b-unit 7.3: Chemical effect of current	Hrs. theory 4 Hrs. lab
	jectives:	Content:
<u> </u>	Explain the term electrolysis, electrolyte,	1. Faraday's laws of electrolysis and the method
	electrodes (cathode and anode) and ions.	of its verification.
2.	Explain electrochemical equivalent of the	 Faraday's constant and electro chemical
<i>2</i> .	elements.	equivalent.
3.	Explain Faraday's laws of electrolysis and	3. Thermocouple, Seebeck and Peltier effect.
] .	experimental verification.	4. Terms, neutral point and temperature of
4.	Define Faraday's constant.	inversion.
5.	Explain the thermocouple principle.	
		5. Concept about thermoelectric series.
6.	Explain Seebeck and Peltier effect.	
Un	it 7: Current Electricity	
	b-unit 7.4: Alternating Current	Hrs. theory 4 Hrs. lab
	jectives:	Content:
1.	Describe alternating current (AC) and its	1. AC and DC.
	interpretation.	2. Importance of AC over DC.
2.	Relate rms and mean value of current and	3. Expressions i_{rms} , v_{rms} and i_{mean} , v_{mean} with peak
	voltage with its peak value.	value.
3.	Appreciate that ac meters measures rms values	4. Introduction of a transformer and energy loss
	only.	mechanisms in transformers.
4.	Explain theintroduction of a transformer and its	5. Faraday's law of electromagnetic induction.
	losses.	6. Simple numerical problems
5.	Describe step up and step down transformers.	1 1 1
6.	Define stabilized voltage.	
7.	State and explain Faraday's laws of	
	electromagnetic induction.	
Un	it 8: Modern Physics	Hrs. theory 30 Hrs. lab
	b-unit 8.1: Electron	Hrs. theory 6 Hrs. lab
	jectives:	Content:
1.	Explain the particle nature of electricity.	1. Particle nature of electricity.
	Discuss the nature, production and properties	2. Production and properties of cathode rays.
	of cathode rays.	3. Moving electrons in electric and magnetic
3.	Derive the motion of electrons in electric and	fields.
	magnetic fields.	4. Specific charge of an electron (introduction)
	magnetie fields.	5. Simple numerical problems
Un	it 8: Modern Physics	
	10 01 11 10 00 11 1 11 11 9109	

Sul	o-unit 8.2: Photoelectricity	Hrs. theory 4 Hrs. lab
Ob	jectives:	Content:
	Define the terms photoelectric effect, photon,	1. Photoelectric effect.
	wave function, threshold frequency and	2. Quantum theory of radiation.
	stopping potential.	3. Einstein's photoelectric equation h $v = \phi + \frac{1}{2}$
2.	Explain photoelectric effect on the basis of the	mv^2 and interpretation
2.	quantum theory of radiation.	4. Workings of photocells
3.	Draw a photoelectric circuit.	5. Light on photographic plate and
<i>4</i> .	State Einstein's photoelectric equation.	photochemical reaction
ч. 5.	Give the application of photoelectric effect	
5.	(photocell).	6. Simple problems using photoelectric equations.
Un	it 8: Modern Physics	equations.
	p-unit 8.3: X-ray	Hrs. theory 5 Hrs. lab
	jectives:	Content:
	Draw well labeled diagram of modern x-ray	
1.	· ·	1. Production, nature and use of x-rays.
2	tube.	2. Property of x-rays.
2.	Explain the production mechanism of x	3. Various uses of x-rays
2	rays(Coolidge X-ray tube)	4. Bragg`s law of X-ray diffraction
3.	Discuss the properties and uses of x-rays	5. Simple numerical problems
4.	Explain Bragg`s law	
	it 8: Modern Physics	
	o-unit 8.4: Radioactivity	Hrs. theory 6 Hrs. lab
	jectives:	Content:
1.	Explain the difference between natural and	1. Radioactivity.
	artificial radioactivity.	2. Properties of α , β and γ radiations.
2.	List the main properties of α , β and γ radiation.	3. Laws of radioactive disintegration.
3.	Explain why these forms of radiation have	4. The constant relationship between half-life
	energy on the order of mega electron voltage.	and decay.
4.	Write down the equations for the laws of	5. Medical uses of radiation and artificial
	radioactivity.	radioactive nuclei.
5.	Write down the formula that shows that the	6. N= N ₀ e ^{-λt} , dN/dt = - λN
	relationship n between half-life and decay are	7. Simple numerical problems.
	constant.	
6.	Graph the decay of radioactivity with time.	
	Explain the principle involved in radio carbon	
	dating.	
Un	it 8: Modern Physics	
	o-unit 8.5: Properties of nucleus	Hrs. theory 5 Hrs. lab
	jectives:	Content:
1.	Describe the constituents of a nucleus.	1. The constituents of nuclei.
2.	Classify different types of nuclei.	2. Isotopes and mass numbers of different
3.	Define unified atomic mass units (amu), mass	elements.
	defect, binding energy and binding energy per	3. Isotope instability.
	nucleons.	4. $E = mc^2$ (only qualitatively).
4.	Calculate the mass defect and binding energy	5. Fission, fusion and energy released from these
ч.	of a nucleus.	nuclear reactions.
5.	Calculate energy equivalence of mass in joules,	 Radiation hazards and safety.
5.	eV, and MeV.	 7. Calculate mass defect, loss of mass due to
6		
6.	Explain Einstein's mass-energy relationship theory.	 radioactive disintegration numerically. 8. Biological effect of nuclear radiations
	LINEOFV.	8. Biological effect of nuclear radiations

7.	Calculate energy released from the decay of	9.	Simple nume	erical problem	ns	
	radioactive isotopes.					
8.	Define fission and fusion and calculate the					
	energy released.					
9.	Discuss health hazards and safety related to radiation.					
10	Explain biological effects of nuclear radiations					
	t 8: Modern Physics					
	-unit 8.6: Physics and Society	Hr	s. theory	4	Hrs. la	ah
	jectives:		ntent:		111 5. 10	ao
<u> </u>	Describe how our environment is being	1.	Deteriorating	g conditions	of the env	vironment
	destroyed due to noise pollution, air pollution		we live in.	2		
	water pollution, radiation pollution	2.	Useful and h	armful aspec	ets of radi	ation.
2.	Discuss the wide spectrum of electromagnetic	3.	Concepts abo			
	radiation from radio waves to cosmic rays.		effect and ac			
3.	Discuss ozone depletion, greenhouse effect,	4.	Environment	tal protection	n strategie	es
	acid rain.			-	C	
4.	Discuss strategies to reduce pollution at local					
	and national levels.					
Pr	acticals	H	rs. theory	Hrs	s. lab	80
Ob	ectives:		ntent:			
1.	Determine the volume of a hollow cylinder and	1.	Application	of theory for	m preced	ing units.
	a solid cylinder using vernier calipers.	2.	Note: Should			
2.	Determine the volume of a steel ball and cross		minimum fif	teen(15) Exp	periments	from above
	section of a glass rod using a micrometer screw		list.			
	gauge.	3.	Marks distrib	oution for fin	al practic	al
3.	Determine thickness of glass plate using		examination			
	spherometer.	4.	(Practical No	ote-3, Oral-3	& Experi	iment -6)
4.	Determine the acceleration due to gravity by					
	using simple pendulum.					
5.	Verify Archimedes' principle and find the					
	specific gravity and density of solids heavier					
	than and insoluble in water					
6.	Determine the specific gravity of solids					
7	dissolved in water.					
7.	Determine the specific gravity and density of					
0	substances lighter than and insoluble in water					
8.	Verify the laws of reflection of light and find the relationship between object distance and					
	the relationship between object distance and					
0	image distance. Determine the refractive index of liquid/glass					
9.	1 0					
10	slab using travelling microscope. Verify laws of refraction and find the refractive					
10.	index.					
11	Determine the upper and lower fixed points of					
11.	a given thermometer and find the correct					
	temperature of tap water.					
12	Find the focal length of a convex lens by the					
12.	double pin method.					
L	double pill memod.					

13. Verify the laws of moments of forces and find	
the weight of a given body.	
14. Determine the latent heat of fusion of ice.	
15. Determine the melting point of wax by cooling curve method.	
16. Determine the magnetic moment and pole-	
strength of a bar magnet by locating the neutral	
points, keeping N-pole pointing south	
17. Verify Ohm's law by using an Ohm meter and	
volt meter.	
18. Demonstrate the variation of lateral	
displacement with an angle of incidence in a rectangular slab.	
19. Determine the refractive index of a prism using	
the I-D curve method.	
20. Determine velocity of sound in air at NTP	
using resonance tube apparatus	
21. Determine angle of dip in the laboratory	
22. Determine frequency of AC source using	
sonometer	

Chemistry

Level : Certificate (Health Science) hrs/week)

Year First

Teaching Hours: 240 (6

Theory Hours:160Practical Hours:80Full Mark:100

Course Description

This course is an introductory course designed for the students specializing Health Science and has two parts: theoretical and practical. The theoretical part consists of different units of general or physical chemistry, inorganic chemistry and organic chemistry - Structure of atom, Chemical bonding, Acids and Bases, Periodic table, Redox reactions, Metals and metallurgy, Principles qualitative analysis, Structure and properties of organic compounds, Alkanes, Alkenes, Alkynes, Aromatic hydrocarbons, Stereoisomerism, Solution, Chemical kinetics, Catalysis, Colloids, Chemical equilibrium, Ionic equilibrium. In practical part, the different experiments to be performed are listed in practical course. The students are required to secure pass marks in theory as well as practical course separately. Emphasis is given to the principles related to chemistry within every day life and to the application of chemistry inhealthscience.

Course Objectives:

The general objectives of this course are as follows:

- To provide students with general knowledge and basic aspects of physical, organic and inorganic chemistry.
- To inculcate the knowledge and skills of chemistry through learning experience and practical activities.
- To provide students with an opportunity to understand enquiry based chemistry with its application various fields.
- To prepare report on practical record file using appropriate methods and approaches.
- To provide students with hands on and mind on experience chemistry processes, skills and tools.
- To interpret the nature and fundamentals of chemistry in health science.
- To assist the students to know about the importance of chemistry and their role in body mechanism.
- To make the students familiar with the sources, effects, chemical present in the atmosphere and control measures of environmental pollution.

Course: Chemistry	Hrs. theory 160 Hrs.lab 80
Unit 1: Physical Chemistry	Hrs.theory 67 Hrs.lab
Sub-unit 1.1: Elements, compounds and chemical change	Theory: 3 hours
Course Objectives:	Contents:
 List the symbols of elements. Identify monovalent, divalent, trivalent elements andradicals. List the information conveyed by symbol and formula. Identify physical and chemical change. 	 Symbols for the atom, molecule, and compound radicaland variable valency. Writing a chemical formula, molecular formula and empirical formulae. Significance of symbol and formula

• Identify the suitable process for separatingconstituents of a mixture: filtration, sublimation, crystallization, distillation.	 Chemicalcompound and itsdifferencesfrom mechanicalmixture. Pure and impure substances. The processes of separating the constituents of a mixture:
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration – Reaction of sodium on water
Sub-unit 1.2: Chemical equations	Theory : 3 hours
Specific Objectives:	Contents:
 Construct a graphical representation of therelationship between amount of reactant and product with time. Demonstrate how to balance a chemical equation. Explain any seven types of reaction with two examples of each. Balance the chemical equation by hit and trial and partial equation method. 	 Chemical formula, Chemical equation, reactant and product. Significance and limitations of chemical equations. Types of chemical reactions (seven-types) withexamples. Balancing a chemical equation by: hit and trialmethod partial equationmethod
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Theoreticalexplanation, Classroom instruction exercises, Demonstration – Reaction of apiece of zinc with excess acid.
Sub-unit 1.3: Periodic table	Theory : 5 hours
Specific Objectives:	Contents:
 Identify the location of S,P,d, and f – block elements. Define atomic radii, electro-negativity IP, EA. Identify alkali and alkaline earth metals, halogens,noble gases, transition metal, radioactive elements and indicate theirlocation. Mention the success and anomalies of Mendeleev's periodic table. 	 Historical development of periodic table. Periodic classification of elements. Location ofs,p,d andf-block elements Mendeleev's periodic table, success and defects of Mendeleev's periodic table Periodocity in properties by: Atomic radii Electronegativity Ionisational potential Electron affinity
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration – Reaction of a piece of zincwith excess acid. Chartdisplay: Long and short form of periodic table.
Sub-unit 1.4: States of matter - Gaseous state	Theory : 5 hours
Specific Objectives:	Contents:
• Compare the volume of gas at different conditions (pressure and temperature).	Differences between solids, liquids and gases.Kinetic theory of gases.

 Compare the rates of diffusion of different gases. Explain the kinetic theory of gases. Explain Dalton's law of partial pressure. Derive Graham's law of diffusion. State the law of stoichiometry. 	 Effect of pressure and temperature on volume of gas. Simple derivation of ideal gas equation (PV=nRT) Dalton's law of partial pressure. Graham's law of diffusion. Law of stoichiometry, Avogardo's hypothesis Simple chemical calculations
Evaluation methods:written exam, oral and written assignments, performance observation in lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration – Reaction of a piece of zincwith excess acid
Sub-unit 1.5: States of matter - Liquid state	Theory : 4 hours
 Specific Objectives: Define solution and its types, solubility and solve problems based on solubility. Define viscosity and surface tension. Describe Raoult's law. List out the colligative properties of solution. Evaluationmethods: written exam, oral and written assignments, performance observationin lab 	 Contents: Solution and its types (Unsaturated, saturated and supersaturated solution). Solubility and related numerical problems. Viscosity and surface tension. Raoult's law Colligative properties of solution Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration – Compare viscosity of glyceroland kerosene.
Sub-unit 1.6: States of matter - Solid State	Theory : 1 hour
Specific Objectives:	Contents:
 Define amorphous and crystalline solids and giveexamples. List the examples of crystallization, molecular crystal, covalent crystal, ionic crystal, water of crystallization 	 Classification of solids. The difference between amorphous and crystalline solids. Molecular crystal, Covalent crystal, Ionic crystal, Water of crystallization
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration – FeCl3 exposed to air, bluevitriol heated
Sub-unit 1.7: Atomic structure	Theory : 5 hours
Specific Objectives:	Contents:
 Define electron, proton & neutron with their chargeand mass. List the postulates of Bohr's atomic model. Explain the Bohr's model of hydrogen atom. Explain Rutherford's nuclear model of atom. 	 Fndamentalparticles of atoms. Bohr'spostulates of atomic model, Bohr's explanation of hydrogen spectrum. Rutherford's nuclear model of atom. Afbau'sprinciple, Hund's rule. Atomic number, mass number, atomic weight and gramatomic weight, Isotopes andisobars.

Define Afbau's principle, Hund's rule, atomic number, mass number, atomic weight, isotopes and isobars. Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration.
Sub-unit 1.8: Electronic theory of valency	Theory : 3 hours
Specific Objectives:	Contents:
 Define electronic theory of valency. List the properties of electrovalent, covalent and co-ordinate covalent compounds. Mention the factors affecting the formation of ionic and covalent bond and also hydrogen bond. Evaluationmethods: written exam, oral and written assignments, performance observationin lab 	 Electronic theory of valency Types of chemical bond Electrovalent Covalent Co-ordinate covalent Factors affecting the formation of ionic and covalent bond; Hydrogen bond. Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation,
Sub-unit 1.9: Oxidation and Reduction	problem solving, demonstration Theory : 5 hours
Specific Objectives:	Contents:
 Describe oxidation and reduction with example. To balance the chemical equation by oxidation number method and ion-electron method. Evaluationmethods: written exam, oral and written assignments, performance observationin lab 	 Classical concept of oxidation and reduction. Electronic concept of oxidation and reduction. Oxidant and reductant and oxidation number Examples of redox reaction. Balancing the chemical equation by oxidation number method and ion electron method. Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation,
Sub-unit 1.10: Electro chemistry	problem solving, demonstration . Theory : 5 hours
Specific Objectives:	Contents:
 Differentiate between Electrolytes and non-electrolytes Strong electrolytes and weak electrolytes. Ions and atoms. Describe the variation of degree of ionization State briefly Faradays' laws of electrolysis. Compare the PHof neutral water above andbelow 250C. Define buffer solution (acidic and basic) Solve numerical problems related with PHofacidic or basic solutions. 	 Electrolytes, Non-electrolytes, strong and weakelectrolytes. Arrhenius theory of ionization. Faradays' laws of electrolysis. Electrolysis of water, Ionic productof water, PH, POH, Buffersolution Importance ofPHand buffer in human body. Simple numerical problems.
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration

Sub-unit 1.11: Acid, base and salt	Theory : 2 hours
Specific Objectives:	Contents:
 Compare general properties of acid, base and salts. Define weak and strong acid and base. Elucidate the Arrhenious and Bronsted-Lowry concept of acids and bases. List the different types of salts. Identify the nature of salt solution. Identify the requirements for the substance to beantacid and antabase. 	 Characteristics of acids, bases and salts. Arrhenious and Bronsted-Lowry concept of acids and bases. Salts and their types. Antacids and antabases and their medical uses.
Evaluationmethods: written exam, oral and	Teaching /Learning activities and resources:
written assignments, performance observationin lab	classroom instruction, theoreticalexplanation, problem solving, demonstration – Reaction between: carbonate andacid, acid and base
Sub-unit 1.12: Solutions- True solution	Theory : 2 hours
Specific Objectives:	Contents:
 Define solution, solubility, solubility product and Henry's law. Define osmosis, reverse osmosis, osmotic pressure, isotonic, hypotonic and hypertonic solutions. Discuss the importance of osmosis phenomenon. Evaluationmethods: written exam, oral and 	 Types of solution, Solubility, Solubility product, Solubility curve, Henry's law. Osmosis,osmotic pressure,isotonic,hypotonic and hypertonic solution. Biological importance of osmosis.
written assignments, performance observationin lab	classroom instruction, theoretical explanation, problem solving, demonstration – Add crystals of KMnO ₄ in water and observe
Sub-unit 1.13: Solution – Colloids	Theory : 3 hours
Specific Objectives:	Contents:
 Identify the particle size in true solution, colloidaland suspension. Compare the lyophilic and lyophobic solutions Withregard to the following characteristics: electrical charge, solution, viscosity, precipitation, Tyndal effect, Brownian movement. List examples of different types of colloidal systems. 	 Comparison between true solution, colloidal solutionand suspension. Difference between lyophilic and lyophobic solutions. Coagulation of solutions by – boiling, electrophoresis, addition of electrolyte. Dialysis, and associated colloids. Application of colloids in the medical field and ineveryday life- precipitation of smoke, kidney dialysis machines. Emulsions, gels and gelation.
Evaluationmethods: written exam, oral and written assignments, performance	Teaching /Learning activities and resources:
observationin lab	classroom instruction, theoreticalexplanation, problem solving, demonstration

Sub-unit 1.14: Mole concept and	Theory : 4 hours		
chemical arithmetic			
Specific Objectives:	Contents:		
 Define mole concept in terms of mass, volume and ion. Relate no of mole with gram molecular weight, number ofparticles and volume occupied (for gas). Identify limiting and excess reagent. Estimate the amount of reactant required and productformed in any reaction. Evaluationmethods: written exam, oral and written assignments, performance observationin lab Sub-unit 1.15: Volumetric analysis 	 Mole concept, Mole in the term of mass, volume and ion Relationshipsbased upon chemical equation- Mass – Mass relationship Mass – volume relationship Volume – volume relationship Limitingreagent. Calculation of related numerical problems. Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration Theory : 5 hours 		
Specific Objectives:	Contents:		
 Define different units of concentration and show their relation. Prepare standard solution of desired concentration andsolve problems on dilution. Solvedifferent numericals regarding acidimetry and alkalimetry. Explain H2 displacement and oxide formation for determining equivalent weight. Find the pH change in acid base titration and choice of indicator. 	 Types of chemical analysis. Equivalent and gram equivalent weight of Element, acid, base, and salt; Determination of equivalent weight by hydrogen displacement method. Titration, acidimetry, alkalimetry, end point, indicator, primaryand secondary standard substance Ways of expressing concentration of solution in terms of Normality, Molarity, molality %by mass, % byvolume, parts permillion (PPm), Normality factor pH change in acid base titration and choice of indicator. Calculations to prepare different concentrations ofsolution. 		
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration		
Sub-unit 1.16: Chemical kinetics	Theory : 7 hours		
Specific Objectives:	Contents:		
 Define and find the differences between molecularity and order of reaction. Define reversible and irreversible reaction. State and explain the Le-Chatelier's principle and its application. State the law of mass action. Explain the effect of pressure, temperature andcatalyst on the equilibrium state. 	 Molecularity of reaction, Difference between order of reaction and molecularity of reaction. Reversible and irreversible reaction. Variation of reactant, product and rate of reactionwith progress of reaction (graphical representation) Law of massaction Le Chateliers' principle andits application 		

 Explain the catalyst and temperaturefasten thereaction rate. Effect of increasing concentration in the case of zero, first and second . 	 Activation energy and activated complex. Zero, first and second order reactions Catalysis:Enzyme catalysis, characteristics of enzyme catalysis, promoter, autocatalysis, negative catalysis, catalytic poisoning
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoretical explanation,problem solving, demonstration
Sub-unit 1.17: Chemical thermochemistry	Theory: 5 hours
Specific Objectives:	Contents:
 Match the systems, surroundings and boundaries with ourbody. Identify whether heat is evolved or absorbed when saltis added to water. Identify that energy is evolved in any combustionprocess. Explain first law of thermodynamics. Elaborate Hess's law of heat summation. 	 Introduction Enthalpy and enthalpy change, exothermic and endothermic reactions, heat of combustion and its application, heat of formation, heat of neutralization and heat of solution, bond energy. First law of thermodynamics Hess'slaw
Evaluationmethods: written exam, oral and written assignments, performance observationin lab	Teaching /Learning activities and resources: classroom instruction, theoreticalexplanation, problem solving, demonstration
Unit 2: Organic Chemistry	Hrs. theory 52 Hrs. lab
Sub-unit 2.1: An introduction to organic chemistry	Theory : 2 hours
Specific Objectives:	Contents:
 List the difference between organic and inorganiccompounds. List the importance of organic compounds in medicinesand drugs with common examples. 	 Introduction Organic chemistry as a separate branch, Reason for large number of organic compounds. Difference between organic and inorganic compounds. Sources of organic compound Importance of organicchemistryin medicalfield Structure and uses of simple drugs: Antipyretics, antiseptics, analgesics, antibiotic, antimalarials, tranquilizers, germicides, and fungicides.
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations
Sub-Unit 2.2: Nomenclature of organic compounds	Theory : 4 hours
Specific Objectives:	Contents:
• Write the reasons for large number of organic compounds.	 Functional group and Homologous series IUPAC rule, IUPAC system of aliphatic compounds.

 Classify the organic compounds into various types. Describe functional group and homologous series with different examples. Applythe IUPAC system for nomenclature. Evaluationmethods: written tests, written assignments, performance observation 	 Nomenclature of compounds containing functional and polyfunctional groups. Teaching /Learning activities and resources: classroom instruction, problem solvingexercises,
	demonstrations
Sub-unit 2.3: Isomerism	Theory : 2 hours
Specific Objectives:	Contents:
 Define the different kinds of isomers. Explain chiral carbon, optically active substance. Define dextro – rotatory and laevo – rotatory with example. 	 Definition of isomerism. Structural isomerism of the types- postitional, functional, and chain metamerism and tautomerism with example.
Evaluationmethods: written tests, written assignments, performance observation	Teaching / Learningactivities and resources: classroom instruction, problem solving exercises,demonstrations
Sub Unit 2.4: Organic reaction	Theory : 4 hours
Specific Objectives:	Contents:
 Identify the nature of reaction. Create concept about writing mechanism of simplereactions. Draw the resonance structure with examples. Show the significance of VSEPR theory. Explain the types of hybridization. 	 Carbocation and carbanion. Inductive effect (+Iand –Ieffect), Significance of inductive effect HomolyticandHeterolyticbond fission Electrophiles and Nucleophiles. Resonance, VSEPR theory The types of organic reactions – Electrophilic and nucleophilic substitution, addition, elimination; Types of hybridization (sp, sp2 and sp3)
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations
Sub-unit 2.5: Hydrocarbons	Hrs.theory 6 Hrs. lab
Lesson B. alkene	Theory : 3 hours
Specific Objectives:	Contents:
 Introduce alkene Describe thelaboratory preparation of ethene. Write down the physical, chemical properties and uses of alkenes. Describe the test of alkene. Explain Markonikov's rule and anti-Markovnikov's rule. 	 Introduction Laboratory preparation of ethene fron ethanol. Physical and chemical properties and uses of alkenes. Markovnikov's rule and anti-Markovnikov's rule.

Evaluation methods:written tests, written	Teaching /Learning activities and resources:
assignments, performance observation	classroom instruction, problem solvingexercises,
	demonstrations
Lesson : C. alkyne	Theory : 3 hours
Specific Objectives:	Contents:
 Describe thelaboratory preparation of ethyne. Explain thechemical properties of alkynes. 	 Introduction Laboratory preparation of ethyne from calcium carbide. Chemical properties –Combustion, hydrogenation, catalytic hydration, withBr2solution, with Na, polymerization Uses of ehtyne
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises,
Sub unit 2 6: Alled balidas	demonstrations
Sub-unit 2.6: Alkyl halides	Hrs.theory Hrs. lab
Lesson:A. chloroform	Theory : 3 hour
Specific Objectives:	Contents:
 Describe the laboratory preparation of chloroform. List the physical and chemical properties of chloroform. List the uses of chloroform. 	 Laboratory preparation of chloroform. Physicaland chemical properties and uses of chloroform.
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations
Sub-unit 2.7: Alcohol	Theory : 4 hours
Specific Objectives:	Contents:
 Classify alcohols. Explain the process of fermentation. Write the physical and chemical properties of ethanol. 	 Introduction Classification alcohol as-monohydric, dihydric, polyhydric, primary, secondary and tertiary. Identification primary, secondary and tertiaryalcohol byoxidation method. Preparation of ethyl alcohol from molasses by fermentation. Physical and chemical properties of ethanol (oxidation with sodium, bleaching powder,oxygen, sulphuric acid, CH3COOH, phosphorus halide)
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises,
	demonstrations
Sub-unit 2.8: Carbonyl compound	Hrs.theory 5 Hrs.lab
Lesson:A.Aldehyde and Ketone	Theory : 5 hours

Specific Objectives:	Contents:		
 Mention the general method of preparation of aldehyde. Describe the physical and chemical properties of aldehyde. List uses of formaldehyde. 	 General methods of preparation of aldehydes and ketone: Physical and chemical properties (NH2OH, NH2CONH2, C6H5NHNH2, NH2NH2, NaHSO3; 2,4-DNP, Formaline, Oxidation of ammonia) Uses 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 2.9: Carboxylic Acid	Theory : 2 hours		
Specific Objectives:	Contents:		
 Describe the preparation of carboxylic acid from alcohol, aldehyde and alkyl benzene. Describe the physical and chemicalproperties of carboxylic acids (solubility, acidic character). Describe the uses of carboxylic acid. Evaluationmethods: written tests, written assignments, performance observation 	 Preparation of carboxylic acidfromalcohol, aldehyde and alkyl benzene. Physicaland Chemicalproperties (acidic character, NaHSOCl2, NH3,C2H5OH, P2O5) Uses Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, 		
	demonstrations		
Sub-unit 2.10: Ether Specific Objectives:	Theory : 2 hours		
 Explain the preparation of ether with their common and IUPAC name. Describe the physical and chemical properties. Write down the uses of ether in medicine and everyday life. 	 Contents: Laboratory preparationfrom ethanol. Physical properties. Chemical properties with- Combustion,hydrolysis, reaction exess HI and PC15. Uses in medicine 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 2.11: Aromatic compounds	Hrs.theory 6 Hrs.lab		
Lesson: A.Introduction	Theory : 4 hours		
Specific Objectives:	Contents:		
 Define aromatic compound and list the characteristics. Identify the name of aromatic compounds and someheterocyclic compounds. Describe the preparation of benzene and its properties. 	 Introduction Aromatic compound. Explain benzene nucleus and side chain. Characteristics of aromatic compound. Preparation of benzene, physical and chemical properties (halogenation, nitration, sulfonation, Freidel Craft's reaction) Uses 		

Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 2.11: Aromatic compounds			
Lesson:B. Nitrobenzene	Theory : 2 hours		
Specific Objectives:	Contents:		
 Drawthe formulaeof aliphatic and aromatic nitrocompounds. Describe the preparation, properties and uses of nitrobenzene. Mention the uses of nitrobenzene. 	 Introduction Laboratory preparation f nitrobenzene. Physical properties Reductionreaction of Nitrobenzene in different medium. Uses in everyday life. 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 2.11: Aromatic compounds			
Lesson: C. Aniline	Theory : 3 hours		
Specific Objectives:	Contents:		
• List the preparation, properties and uses of aniline.	 Introduction Laboratory preparation of pure aniline Physicaland Chemicalproperties- basic nature, alkylation, acylation, sulfonation, halogenation, nitration Uses 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 2.11: Aromatic compounds			
Lesson: D. Phenol	Theory : 2 hours		
Specific Objectives:	Contents:		
 Describe the preparation of phenol . Identify the mono and dihydric phenols. Explain preparation, properties and uses of phenol. Write down the Kolbe's reaction and condensation with formaldehyde. 	 Introduction Preparation of phenol. Physicaland Chemical properties (action with zinc dust, NaOH, NH3, PC15, Kolbe's reaction, Condensation with formaldehyde) 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 2.11: Aromatic compounds			
Lesson:E. Benzoic acid	Theory : 2 hours		

Specific Objectives:	Contents:	
 Describe the laboratory preparation of benzoic acid. Write down the physical, chemical properties and uses of benzoic acid. 	 Laboratory preparation of benzoic acid Physicaland Chemical properties (action with alcohol, PC15, NH3 and soda-lime) Uses 	
Evaluation methods:written tests, written	Teaching /Learning activities and resources:	
assignments, performance observation	classroom instruction, problem solvingexercises, demonstrations	
Sub-unit 2.12: Molecules of life	Hrs.theory Hrs. lab	
Lesson:Carbohydrate, Proteins and enzymes, Vitamins and coenzymes and Lipids	Theory : 5 hours	
Specific Objectives:	Contents:	
 Explain the natural sources of Carbohydrate, Proteinsand enzymes, Vitamins and coenzymes and Lipids and their chemicalcomposition. List the functions and uses of Carbohydrate, Proteinsand enzymes, Vitamins and coenzymes and Lipids. 	 Carbohydrate: Definitionand classification, Structure (Linear Cyclic) of glucose, Functions of Carbohydrates Protein: Amino acidand Peptide bond Essential andnon-essential amino acid Denaturationprotein Functions ofProtein Enzymes (Definition and importance) Lipid: Introductionof lipid, fat and oil and their natural sources. Hydrolysis Functions offat and oil Vitamins andcoenzymes: Introduction Fat soluble and insoluble vitamins 	
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solving exercises, demonstrations	
Unit 3: Environmental Chemistry	Hrs. theory 5 Hrs. lab	
Sub-unit 3.1: Pollution	Hrs.theory 5 Hrs.lab	
Lesson: Airand Water Pollution, Radioactive, Acid rain, Ozone layer depletion and GreenHouse Gas Effect	Theory : 5 hours	
Specific Objectives:	Contents:	
 Define source and adverse effects of pollutants. Describe why is environment getting polluted. Identify the cause of acid rain, water pollution and its effects. 	 The sources and adverse effects due to the followingair pollutants- CO2, SO2, O3, H2S,CO, hydrocarbon, lead, cadmium dust,CFC, oxides ofnitrogen. Air pollution and its effects on: 	

 List the causes of nuclear and pesticide pollution. Evaluationmethods: written tests, written assignments, performance observation 	 human health,materials and climate,Greenhouse effect, Ozone layer depletion Acid rain and its adverse effects. Water pollution and its effects. Nuclear and pesticide pollution. Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations 		
Unit 4: Inorganic Chemistry	Hrs. theory 36 Hrs.lab		
Sub-unit 4.1: Hydrogen	Theory : 3 hours		
Specific Objectives:	Contents:		
 Describe the preparation, properties and uses ofhydrogen. Define isotopes and uses of hydrogen. 	 Laboratory preparation of Hydrogen Physical properties. Preparation of Vanaspati ghee. Nascent hydrogen and molecular hydrogen Reaction of nascent hydrogen with KMnO4, FeCl3, K2Cr2O7; Ortho and Para hydrogen, Isotopes Uses of hydrogen. 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 4.2: Water	Theory : 7 hours		
Specific Objectives:	Contents:		
 Explain the cause of hardness of water. Describe the chlorination of water. List the advantage and disadvantage of hard water. Explain the method of purification of drinking water. Define degree of hardness of water. Define heavy water. Mention the difference between soft and hard water. 	 Introduction of soft and andhard water. The process of removal of hardness- Boiling, Clark's process, using washing soda, permutit process, soda – ash method, deionisation of water. The advantage and disadvantage of hard water. The meaning of drinking water. Method of purification of drinking water by- boiling, candle filtration, chemicaldisinfection, bleachingpowder, Cl2 solution, iodine,KMnO4, ozonisation, using potashalum. The solvent property of water. Difference between soft and hard water. 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 4.3: Carbon andits oxides.	Hrs.theory 2 Hrs.lab		
Lesson: A.carbon monoxide	Theory : 2 hours		
 Specific Objectives: Define allotropes of carbon. Describe the laboratory preparation of carbon monoxide 	 Contents: Introduction Allotropes of carbon Laboratory preparation of carbon monoxide 		

• Describe the physical and chemicalproperties of carbon monoxide.	 Physical properties of CO. Chemical properties in reaction with - O2, Cl2, Ni,NaOH, and haemoglobin. 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 4.4: Ammonia	Theory : 3 hours		
Specific Objectives:	Contents:		
Explain the preparation, properties and uses of ammonia.Write down the uses of ammonia.	 Laboratory preparation of ammonia. Physical and chemical properties of ammonia (action with metals, Nessler's reagent, ammonia as a Lewis base, basic nature) Uses of ammonia 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 4.5: Phosphorous	Theory : 2 hours		
Specific Objectives:	Contents:		
 Write down the toxic nature of white phosphorous. Define phosphorescence. Write down the uses of phosphorus. 	 Introduction Occurrence of phosphorous in animal bones, ATP and ADP. Properties of white phosphorous –reactions with O2, with Cl2, with caustic alkali. 		
Evaluationmethods: written tests, written	Uses of phosphorous. Teaching // coming activities and recoverage		
assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 4.6: 33. Sulphur	Theory : 5 hours		
Specific Objectives:	Contents:		
 Describethepreparation, properties and uses of H2S. Explain the preparation, properties and uses of SO2. 	 Laboratory preparation of SO2 Chemical properties of SO2 (action with lime water, sodium carbonate, oxidizing and reducing properties, bleachingproperties) Laboratory preparation of H2S. Reducing properties of H2S. 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		
Sub-unit 4.7: Halogens	Theory : 5 hours		
Specific Objectives:	Contents:		
Describe the laboratory preparation of chlorine, bromine and iodine.Define halogens.	 Laboratory preparation of Cl2,Br2 and I2 Physical properties of Chlorine, bromine and iodine. Compare the chemical properties of halogens- 		

 Compare the properties of chlorine, bromine and iodine. Mention the uses of Cl2. Evaluationmethods: written tests, written 	 Oxidizing action, bleachingaction, in reaction with H2, with slaked lime, and with organiccompounds. Uses of Cl2 Teaching /Learning activities and resources: 		
assignments, performance observation	classroom instruction, problem solvingexercises, demonstrations		
Sub-Sub 4.8: Metallic compounds	Theory : 7 hours		
Specific Objectives:	Contents:		
 Definethe terms ores, flux, slag, calcination and roasting. Give the metallurgical process of metallurgy. Give the preparation, properties and uses of HgCl2,Hg2Cl2,Plaster of Paris, Epsom salt, Bleaching powder,Cu2O, AgNO3, AuCl3, ZnCl2.2H2O Mention the biological importance of Na and K. Evaluationmethods: written tests, written assignments, performance observation 	 Difference between minerals and ores, flux and slag, calcination and roasting. Preparation, properties and uses of HgCl2,Hg2Cl2,Plaster of Paris, Epsom salt, Bleaching powder,Cu2O, AgNO3, AuCl3, ZnCl2.2H2O Biological importance of Na andK Teaching /Learning activities and resources: classroom instruction, problem solving exercises, demonstrations 		
Sub-unit 4.9: Minerals	Theory : 2 hours		
Specific Objectives:	Contents:		
 Describe the sources andneedsof minerals. Write down the biological importance and effects due to their deficiency. 	 Sources of the following minerals- Na, K, Ca, Mg, Fe, Zn, Ni, Cobalt. Biological importance and effects due to theirdeficiency 		
Evaluationmethods: written tests, written assignments, performance observation	Teaching /Learning activities and resources: classroom instruction, problem solvingexercises, demonstrations		

Recommended Books

Pandit, C.N. Dr.; Chemistry Education; K.P. Publication, 4433738, Dillibazar, Kathmandu.

Mitra, Ladli Mohan, A textbook of Inorganic Chemistry. Ghosh&Co.Current edition.

Tuli, G.D.et al., Intermediate Organic Chemistry. S. Chand&Co. Currentedition.

Jauhar, S.P., Modern ABC's of Chemistry (vol. I&II). Modern Publishers. Current edition.

Reference Books

Jha, J.S.,&Gugliani, S.K.,<u>A Textbook of Chemistry</u>. Seirya Publication.Current edition.

Sthapit, M.& Pradhananga, R.R., Fundamentals of Chemistry (vol. I&II). Taleju

Prakashar.Currentedition.

Pandit, C.N. Dr., Subedi, R.R. and Tiwari, Prakash; <u>A Textbook of Chemistry</u>; K.P. Publication, Dillibazar, Kathmandu.

Course: Chemistry	Hrs. theory 160 Hrs.lab 80
Unit 1: General Chemistry - Practical	Hrs.theory Hrs. lab 30
Sub-unit 1.1: Introduction	Hrs.theory Hrs. lab 8
Specific Objectives:	Contents:
 Follow stated laboratory procedures and guidelines. Describe safety and first aid measures for thechemistry lab. Demonstrate the method for chemistry lab documentation. Evaluationmethods: written and viva 	 Procedural rules and guidelines of the chemistry lab. Proper handling of equipment. Lab safety measures. Documentation procedures for laboratory work.
exams, performance observation in laboratorysettings.	instruction, text book self study,demonstration and return demonstration, laboratory practice, problemsolving
Sub-unit 1.2: Use of the Bunsen burner	Hrs.theory Hrs. lab 6
Specific Objectives:	Contents:
 Identify the names and functions of the parts of aBunsen burner. Describe the correct use of the Bunsen burner and itsflame with: air holes closed with airholes open. Differentiate between the uses of oxidizing andnon-oxidizing flames. 	 The correct operation of the Bunsen burner. Parts of the Bunsen burner. Oxidizing and non-oxidizing flames.
Evaluation methods:written and viva exams, performance observation in laboratory settings.	Teaching /Learning Activities / Resources: classroom instruction, text book self study,demonstration and return demonstration, laboratory practice, problemsolving
Sub-unit 1.3: Simple lab operations	Hrs.theory Hrs. lab 16
Specific Objectives:	Contents:
 Separate sand and common salt in pure and dry states from mixture of sand and common salt. Perform chloride, sulphate and nitrate test by wet way. Separate sand and camphor from a mixture of sand andcamphor. Recover the precipitate obtained in pure 	 The process and methods of filtration. Characteristics of filtrate and residue. Chloride ion, sulphate ion and nitrate ion test. Nature of mixtures and components. Principles and processes of sublimation. Principles and process of precipitation. The distillation process.

Chemistry Practical

 Prepare a sample of bazaar copper sulphate atlaboratory temperature and use the solution to get pure crystals of salt. Obtain sodium chloride by the neutralization of: bench ofhydrochloric acid with a bench of sodium hydroxide. Sodiumcarbonate with hydrochloric acid. Prepare a soluble derivative of barium carbonate andsodium chloride. Evaluationmethods: written and viva exams, performance observation in laboratorysettings. 	instruction, text bool	Activities / Resources k self study,demonstrat ratory practice, problen	tion and return
Unit 2: Inorganic Chemistry - Practical	Hrs.theory	Hrs. lab	18
Sub-unit 2.1: Preparation of gases	Hrs.theory	Hrs. lab	8
Specific Objectives:	Contents:		
 Prepare hydrogen,nitrogen, ammonia andcarbon dioxide gases. Identify the properties of hydrogen,nitrogen, ammonia andcarbon dioxide gases. 	ammonia and caChemicals used	atus and prepare hydrog rbon dioxide gas at lab. in gas experimentation. l andchemical propertie	
Evaluationmethods: written and viva exams, performance observation in laboratorysettings.	instruction, text bool	Activities / Resources k self study,demonstrat atory practice, problem	tion and return
Sub-unit 2.2: Salt analysis	Hrs.theory	Hrs. lab	10
Specific Objectives:	Contents:		
 Performsaltanalysisfor basic andacidradicals by dry and wet methods. Evaluationmethods: written and viva exams, performance observation in laboratorysettings. 	 Procedures for identification of basic andacid radicals in salt. (at least 3 salts) Teaching /Learning Activities / Resources: classroom instruction, text book self study, demonstration and return demonstration, laboratory practice, problemsolving 		
Unit 3: Physical Chemistry-Practical	Hrs. theory	Hrs.lab	16
Sub-unit 3.1: Equivalent weights	Hrs. theory	Hrs.lab	8
Specific Objectives:	Contents:		
 Use a chemical balance to weigh various substances. Determine the equivalent weight of a given metal by thehydrogen displacement from acid method. 	 The operation of a chemical balance scale. The meaning of equivalent weight. Calculation of equivalent weights. Determine the equivalent weight of metal by hydrogen displacement method. 		

Evaluationmethods: written and viva exams, performance observation in laboratory settings.	Teaching /Learning Activities / Resources: classroom instruction, text book self study,demonstration and return demonstration, laboratory practice, problemsolving		on and return
Sub-unit 3.2: Acidimetryand	Hrs.theory	Hrs. lab	8
alkalimetry			
Specific Objectives:	Contents:		
 Standardize the given acid which is approximatelydecinormal. Determine the strength of alkalai with the help of astandard acid supplied. Determine the strength of acid in terms of: normality grams/liter percentage 	 Preparation of solu Calculation of street	titration.	gths. Itions in
Evaluationmethods: written and viva exams, performance observation in laboratorysettings.	Teaching /Learning A instruction, text book s demonstration, laborate	self study, demonstrati	on and return
Unit 4: Organic Chemistry -	Hrs.theory	Hrs. lab	16
Practical			
Sub-unit 4.1: Element detection	Hrs.theory	Hrs. lab	8
Specific Objectives:	Contents:		
• Detect the elements present in given organic compounds.	Detection ofnitrog	en, sulphur, halogens.	
Evaluationmethods: written and viva exams, performance observation in laboratorysettings.	Teaching /Learning A instruction, text book s demonstration, laborate	self study, demonstrati	on and return
Sub-unit 4.2: Identification of	Hrs.theory	Hrs. lab	8
organic compounds			
Specific Objectives:	Contents:		
• Identify given organic compounds systematically.	formaldehyde,oxal	of acetate, formate, late, oxalic acid, glyce ic acid, formicacid. tests.	rol, acetone,
Evaluationmethods: written and viva exams, performance observation in laboratorysettings.	Teaching /Learning A instruction, text book s demonstration, laborate	self study, demonstrati	on and return

Reference Book for Practical

^{1.} Khanal, Tarka; Mishra, Parmatma; Joshi, Krishna Raj; <u>Practical Chemistry for CTEVT Level</u> (<u>PCL</u>); A-Z Publication, Kathmandu.

Zoology

Level: Certificate Year: First Credit Hours: Theory Hours: 120 Practical Hours: 80 Assessment Marks: 100

Course Description

This basic course in zoology discusses the characteristics of unicellular and multicellular structures. The course contains introductory zoology, cell biology, animal diversity, economic zoology, life process of mammals, evolution of organisms, relationships between organism and environment and a brief introduction about snakes found in Nepal. In order to be more relevant to the students of health science, the course involves a detailed study of different kinds of tissues, the life history of relevant parasites, and a detailed study of the anatomy and physiology of mammals.

Practical zoology includes the study of microscopes, a general study of animal kingdom (museum specimens), preparation of temporary slides, dissection of mammals so as to expose different systems and the life cycle of mosquitoes and houseflies.

Course Objectives

At the end of the course, the student will be able to:

Tell the meaning, scope and different branches of zoology.

Explain structure and functions of different kinds of tissues in a body.

Identify diversified forms of animal life.

Explain different kinds of parasites and arthropods related to human welfare.

Describe different systems of mammals.

Describe how organisms of today have been evolved from the ancestral ones.

Describe the importance and strategy of wildlife conservation.

Describe the different applications of biotechnologylogy in human health.

Identify common poisonous and nonpoisonous snakes and their effects.

Handle microscope properly.

Identify different kinds of animals.

Prepare temporary slide mount of the given specimen.

Dissect the mammal so as to expose its different systems.

Describe different stages in the life cycle of mosquitoes and houseflies.

Teaching materials required to full fill above mentioned objectives are : Board, Charts, flex prints, Over head projector, Power point projector and other teaching materials prepared by teachers.

Course: Zoology	Hrs. theory	120 Hrs	s. lab 80
Unit 1: Introduction to Zoology	Hrs. theor	3	Hrs. lab
Sub-unit 1.1: Definition, scope and	Hrs. theory	3	Hrs. lab
branches of zoology			
Objectives:	Content:		
1. State the meaning of Zoology.	1. Meaning of Z	Zoology	
2. Describe the economic, literary and	2. Scope of Zoo	ology	
aesthetic values of Zoology.	3. Different bra	nches of Z	oology related to
	medical scien	nce:	

3. Differentiate the different branches of Zoology.	 On the basis of structure and function - morphology, anatomy, physiology, histology, cytology. On the basis of specific unit or field - toxicology, genetics, embryology, evolution, mycology, microbiology, ecology, parasitology, paleontology, taxonomy. On the basis of specific group - entomology, helminthology, protozoology, bacteriology, virology.
Evaluation methods: oral tests, home assignments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study.
Unit 2: Animal tissues and their types	Hrs. theory 17
Sub-unit 2.1: Epithelial tissue	Hrs. theory 6
 Objectives: Define a tissue. Name different types of tissues (Epithelial tissues, Connective tissues, Muscular tissues, Nervous tissues). Describe structure, function and location (in our body) of each of the following tissue types: Simple epithelium tissue, Squamous epithelium, Cuboidal epithelium, Ciliated cuboidal, Brushbordered cuboidal Columnar epithelium, Ciliated columnar, Brushbordered columnar, Pseudostratified epithelium Compound epithelium tissue :Stratified epithelium, Stratified squamous epithelium (keratinised epithelium and non-keratinised epithelium), Stratified cuboidal epithelium, Stratified cuboidal epithelium, Stratified squamous epithelium, Stratified squamous epithelium (keratinised epithelium, Stratified cuboidal epithelium, Stratified squamous epithelium, Stratified cuboidal epithelium, stratified columnar epithelium, transitional epithelium. 	 Content: Definition of tissue and its types. Functions of epithelial tissues i.e. protection, secretion, excretion, absorption, exchange of materials/gases, sensory. Structure, locations and functions of different types of epithelial tissues.
Evaluation methods: oral tests, home assignments, written examination Unit 2: Animal tissues and their types	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, audiovisuals showing epithelial tissues.
Sub-unit 2.2: Connective tissues	Hrs. theory 6
Objectives: 1. Define connective tissue. 2. Describe briefly the characteristics, structure and functions of different types of cells forming connective tissues (Cell	 Content: 1. Definition of connective tissue and its types. 2. Structural and functional study of different types of connective tissues.

 13. Fluid connective tissue (Haemopoietic tissue): 14. Blood, Lymph 15. Identify composition of blood and blood plasma, functions of blood plasma, structure, usual number and functions of erythrocytes, leucocytes and thrombocytes. 16. Differentiate between blood plasma and serum. Unit 2: Animal tissues and their types Sub-unit 2.3: Muscular tissues Objectives: 	Hrs. theory 3 Content: 1. Definition of muscular tissue and its types. 2. Structure and function of different types of
 Define muscular tissue. Name different types of muscular tissues (striated, unstriated and cardiac). 	muscular tissues.

1. Binomial system of nomenclature adopted by
Content:
ure Hrs. theory 3
classroom instruction, discussion, textbook /reference book self study.
Teaching / Learning activities and resources:
on.
n of
4. Modern trends in taxonomy.
3. (Natural & Artificial).
atural 2. Different systems of classification
higher taxa.
1. Definition of taxonomy, species as a basic unit of classification, systematics, taxon, lower and
Content:
ny Hrs. theory 2
Hrs. theory 5
classroom instruction, discussion, textbook /reference book self study, audiovisuals showing nervous tissues.
Teaching / Learning activities and resources:
fibre.
d or
ixon.
types of nervous tissues.
 Definition of nervous tissue and its types. Structural and functional study of different
Content:
Hrs. theory 2
bes la
muscular tissues.
classroom instruction, discussion, textbook /reference book self study, audiovisuals showing
Teaching / Learning activities and resources:

 Identify the importance of nomenclature. Identify the system adopted by the International Code of Zoological Nomenclature. Write scientific names of commonly found animals. List common names and binomial names of those animals which are used in medical science; identify the useable body parts of each. Describe each of the five kingdoms of classification with examples. Identify the interrelationships among these kingdoms. 	 Selected examples of binomial nomenclature of animals. Five kingdom system of classification. Chief characteristics and examples of five kingdoms.
Evaluation methods: oral tests, home assignments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, visuals showing the five kingdom classification of animals.
Unit 4: Economic Zoology	Hrs. theory 40
Sub-unit 4.1: Hosts and parasites	Hrs. theory 3
Objectives:	Content:
 Define hosts and parasites in general. Define different kinds of parasites - ecto and endo-parasites, temporary and permanent parasites, facultative and obligatory parasites, occasional or accidental and wandering or aberrant parasites. Define different kinds of hosts - intermediate, definitive and paratenic (transport) hosts. Give examples for different kinds of hosts and parasites. Name at least 20 different parasites and their usual hosts. Define mutualism, commensalism and parasitism with examples of each. Identify different attributes of parasites - infectivity, invasiveness, pathogenicity and toxigenicity. Identify the specific and non-specific resistance factors of hosts. Summarize the delicate adjustments between a host and a parasite. 	 Meaning of hosts and parasites Common types of hosts and parasites with examples. Types of relationships between a host and a parasite. Delicate adjustments between hosts and parasites.
Evaluation methods: oral tests, home assignments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook
Unit 4: Economic Zoology	/reference book self study, illustrations, slides.

Sub-unit 4.2: Medically important	Hrs. theory 15	
protozoans		
Objectives: 1. Describe the morphology of trophozoite, pre-cystic and cystic stages of	Content: 1. Systematic position, distribution, habitat, morphology, life cycle, mode of transmission, pathogenia affasts and Proventive measures of	
 <i>Entamoebahistolytica.</i> 2. Define minuta and magna forms, convalescents and carriers. 3. List characteristics of cysts. 4. Identify the usual host and the infective stage of <i>Entamoeba histolytica</i>. 5. Describe the life history of <i>E. histolytica</i> using a labeled diagram. 6. Discuss the relationship between amoebic ulcer and amoebic dysentery. 7. Define ciliated protozoa. 8. Describe the usual habitat and morphology of <i>B. coli</i>. 9. Describe the pathogenic significance of <i>B. coli</i>. 10. Describe control measures of <i>B. coli</i>. 11. Identify usual habitat, life history of <i>Plasmodium vivax</i> using a labeled diagram. 12. Define nutrition in <i>Plasmodium</i>. 13. List control measures of <i>P. vivax</i>. 14. Define flagella and flagellated protozoans. 15. Describe morphology, mode of transmission, pathogenic significance and control measures of <i>Giardia lamblia</i> and <i>Leishmania donovani</i> using a labeled diagram. 16. Differentiate between amastigote and promastigote form of <i>Leishmania donovani</i>. Describe distribution,habitat,morphology,mode of infection, pathogenic significance and control, pathogenic significance and promastigote form of Leishmania donovani. 	 pathogenic effects and Preventive measures of : Entamoeba histolytica, Plasmodium vivax, , Leishmania donovani and Blantidium coli. Systematic position, distribution, habitat, morphology, mode of transmission, pathogenecity and preventive measures of : Entamoeba gingivalis, Giardia lamblia, Trichomonas vaginalis 	
preventive measures of <i>Trichomonas</i> <i>vaginalis</i> . Evaluation methods: oral tests, home	Teaching / Learning activities and resources:	
assignments, written examination	classroom instruction, discussion, textbook /reference book self study charts, slides, diagrams.	
Unit 4: Economic Zoology		
Sub-unit 4.3: Medically important	Hrs. theory 14	
helminthes		
Objectives:	Content:	
1. Systematic position, distribution and habitat, life cycle, mode of transmission,	1. Distribution, habitat, morphology, life cycle, mode of transmission, pathogenic effects and	

 pathogenic significance and prevention of helminth parasites. 2. Describe the mechanism of development of human cystocercosis. Evaluation methods: oral tests, home assignments, written examination 	 Preventive measures of : Taenia solium, Hymenolepis nana, Ascaris lumbricoides, Ancylostoma duodenale, Wuchereria bancrofti. Distribution, habitat, morphology, mode of transmission, pathogenecity and preventive measures of : Taenia saginata, Trichuris trichiura,Echinococcus granulosus, Enterobius vermicularis. Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study charts, slides, diagrams. 	
Unit 4: Economic Zoology		
Sub-unit 4.4: Medically important	Hrs. theory 8	
arthropods		
Objectives:	Content:	
 Describe the distribution, habit and habitat, brief life history, and control measures of: Mangemite (<i>Sarcoptes scabiei</i>) Cockroaches (<i>Periplaneta americana</i>) Houseflies (<i>Muscanebulo</i>) Mosquitoes (<i>Culex, Anopheles</i> and <i>Aedes</i>) Sand flies (<i>Phlebotomusargentipes</i>) Human louse (<i>Pediculushumanus</i>) Bed bug (<i>Cimex</i>) Fleas (<i>Xenopsyllacheopis</i>) List diseases caused or transmitted by each of them. Distinguish between pathogenic and non- pathogenic insects. Distinguish between reservoirs and vectors. Evaluation methods: oral tests, home 	 Introduction, Classification and public health importance of medically important arthropods. Distribution, habit and habitat, morphology, diseases and control measures of : Mangemite (<i>Sarcoptes scabiei</i>),Cockroaches (<i>Periplaneta</i> <i>americana</i>), Houseflies (<i>Musca nebulo</i>), Mosquitoes (<i>Culex, Anopheles</i> and <i>Aedes</i>),Sand flies (<i>Phlebotomusargentipes</i>),Human louse (<i>Pediculus humanus</i>), Bed bug (<i>Cimex</i>), Fleas (<i>Xenopsylla cheopis</i>). General concept of Integrated vector management approaches. 	
assignments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study charts, slides, diagrams.	
Unit 5: Life Process of Mammals	Hrs. theory 34	
Sub-unit 5.1: Digestive system	Hrs. theory 9	
Objectives:	Content:	
 Define food and nutrition. List the basic kinds of nutrients - carbohydrates, proteins, lipids, vitamins, enzymes, minerals. Identify the role of nutrients in the body of organisms. Define digestion and digestive system. List organs involved in digestive system. List parts of alimentary canal. 	 Nutrition in mammals. Structure and functions of parts of alimentary canal of human (Oral cavity, pharynx, esophagus, stomach, small intestine and large intestine). Structure and functions of significant regions of alimentary canal and associated digestive glands. Enzymatic actions of digestive glands for the digestion of carbohydrates, proteins and lipids. 	

Objectives: 1. Define the heart of mammals. 2. Describe external and internal structures of the heart. 3. Describe the course of blood circulation	 Content: Definition and types of circulatios. Structure (external and internal) of the heart of mammals. Course of blood circulation in heart.
	Content:
Sub-unit 5.3: Circulatory system	Hrs. theory 7
Unit 5: Life Process of Mammals	reference book sen study, charts, shues, utagranis.
assignments, written examination	classroom instruction, discussion, textbook /reference book self study, charts, slides, diagrams.
Evaluation methods: oral tests, home	Teaching / Learning activities and resources:
 Transport of oxygen and carbondioxide. Define the terms: Tidal volume, Expiratory reserve volume, Inspiratory reserve volume, Residual volume and Total lung capacity. Bohr effect and Chloride shift. 	Expiratory reserve volume, Inspiratory reserve volume, Residual volume and Total lung capacity. Bohr effect and Chloride shift.
 External respiration (ventilation mechanisms) Internal or cell respiration Transport of exurem and carbondioxide 	 Internal or cell respiration Transport of oxygen and carbondioxide. Respiratory air volumes: Tidal volume, Expiratory reserve volume. Inspiratory reserve
 Describe structure and functions of the respiratory organs and associated structures. Describe mechanisms of: 	 Structure and functions of the respiratory organs of human. Mechanisms of: External respiration (ventilation mechanisms)
1. Define respiration and respiratory system.	1. Definition and types of respiration in animals.
Objectives:	Content:
Sub-unit 5.2: Respiratory system	Hrs. theory 4
Unit 5: Life Process of Mammals	
assignments, written examination	classroom instruction, discussion, textbook /reference book self study charts, slides, diagrams.
Evaluation methods: oral tests, home	Teaching / Learning activities and resources:
mammals.12. Describe the processes of absorption of food products through the absorption sites to the blood circulation.	
 Describe the chemistry of digestion in different organs of alimentary canal. Identify the sites for the absorption of digested foods in the alimentary canal of 	
9. Describe the mechanical and chemical digestion in different organs of alimentary canal.	
 Describe structure and functions of different digestive glands. 	digested food
parts of alimentary canal.	5. Absorption sites and processes of absorption of

 Describe conduction of heart waves (impulses). Identify control of heart working. Define the terms such as Pacemaker, Heart sound, etc. Differentiate arterial blood and venous blood. Differentiate arteries and veins. Define capillaries and capillary network. Describe arterial blood circulation - systemic and pulmonary. Describe venous blood circulation - systematic (including hepatic portal system) and pulmonary. Evaluation methods: oral tests, home assignments, written examination Unit 5: Life Process of Mammals 	(Systemic, pulmonary and coronary). Hepatic portal system. Course of blood circulation in heart. Origin, conduction and regulation of heart beat.
Sub-unit 5.4: Excretory system	Hrs. theory 4
Objectives:	Content:
 Define excretion and excretory system. Name the types of excretory organs in mammals such as skin, lungs, liver and kidney. List excretory functions of skin, lungs, liver and kidney. Describe external and internal structure of a kidney. Describe the structure and functions of different parts of nephron. Describe the process of urine formation in mammals. List composition of urine. Define micturition and its causes. Explain the homeostatic function of the kidney. 	 Meaning of excretion, types of excretory organs and their functions. External as well as internal structure of a kidney. Structure of a nephron. Mechanism of urine formation (glomerular filtration, selective reabsorption, tubular secretion) and functions of different regions of a nephron. Micturition and homeostasis.
Evaluation methods: oral tests, home assignments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, charts, diagrams and visuals showing internal and external structures of the kidney, uriniferous tubules.
Sub-unit 5.5: Reproductive system	Hrs. theory 4
Objectives:	Content: 1. Definition of reproduction and its types -
 Define reproduction and its types - sexual and asexual. Differentiate between sexual and asexual reproduction. 	 Definition of reproduction and its types - sexual and asexual. Spermatogenesis and Oogenesis. Structure and functions of male and female reproductive organs.

3. 4. 5. 6. 7. 8.	Describe structure and functions of primary sex organs or gonads (testes and ovaries). Identify the secondary sex organs of males (prostrate, seminal vesicles, vas deferens and penis) and females (fallopian tubes, uterus, vagina and mammary glands). Describe the structure and function of epididymus and the duct system of male. Give composition of semen. Give short description on spermatogenesis. Describe the structure and functions of the duct system of female (fallopian tubes, uterus and vagina) carrying spermatozoa from vagina to the fallopian tube.	4. Menstruation process.
9.	Give a short description on ovulation and menstruation.	
	aluation methods: oral tests, home signments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, charts, diagrams and visuals.
Un	it 5: Life Process of Mammals	
Su	ıb-unit 5.6: Nervous system	Hrs. theory 6
-	jectives:	Content:
1. 2. 3. 4. 5. 6. 7. 8.	Define nervous system. Identify communication of information With the outside world through eyes, ears, nose, tongue and skin. Within the body through nerve impulses and chemical substances. Summarize functions of nervous system. Name types of nervous system - central, peripheral and autonomous. Describe meninges of brain and subarachnoid space. List functions of cerebrospinal fluid.	 Definition of nervous system. Structure and functions of different types of (central, peripheral and autonomous) nervous systems. Transmission of nerve impulses.

 15. Define reflex action with examples. 16. Describe physiological process of nerve impulse conduction. 17. Define preganglionic fibres, autonomic ganglia and postganglionic fibres. 18. Describe sympathetic and parasympathetic nervous system. Evaluation methods: oral tests, home assignments, written examination Unit 6: Evolution 	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, charts, diagrams and visuals.
	Hrs. theory 8
 Objectives: Brief description of origin of life. Define evolution and organic evolution. Describe historical background of organic evolution. Give examples of organic evolution. Distinguish between progressive and retrogressive evolution. Summarize the evolution of modern man starting from human ancestors <i>Dryopithecus</i>. Describe the evidence of organic evolution: morphological and anatomical, palaeontological, biochemical, genetic and embryological. Describe Lamarck's theory of evolution giving examples cited by him. Identify drawbacks of Darwin's theory. Describe mutation theory of evolution. Describe mutation theory of evolution.	 Content: Brief description about origin of life. Definition and Pattern of organic evolution Morphological and anatomical, palaeontological, biochemical and embryological evidences. Description of : Lamarckism, Darwinism and Neo-Darwinism (modern synthetic theory of evolution) With examples. Summarize the evolution of modern man starting from human ancestors <i>Dryopithecus</i>.
Evaluation methods: oral tests, home assignments, written examination	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, charts, diagrams and visuals of geological time scale showing evolutionary stages.
Unit 7: Wildlife conservation	Hrs. theory 3
 Objectives: Describe wildlife. Differentiate between wild life and domestic life. To know the importance of wildlife conservation. 	 Content: Definition of wildlife and conservation. Importance of wildlife conservation. Categories of wildlife with example. Causes of extinction of wildlife. Brief discussion on protected areas of Nepal.

 Describe different categories of wildlife (Extinct, Endangered, Rare, Intermediate) with examples . Identify different causes of extinction of wildlife. To know about different protected areas of Nepal. Evaluation methods: oral tests, home assignments, written examination 	Teaching / Learning activities and resources: classroom instruction, discussion, textbook /reference book self study, charts, diagrams and visuals.
Unit 8: Application of biology	Hrs. theory 5
 Objectives: To the knowledge on Vaccine in human health. Describe different types of vaccines. To impart the knowledge of tissue and organ transplantation. To give a general concept of Amniocentesis. To give a concept of test-tube baby. 	 Content: Definition, types and application of antibiotics, vaccines in human health. Definition and meaning of organ transplantation. Application of organ transplantation, examples of tissue and organ transplantation. Definition , brief process, advantage and disadvantage of amniocentesis. Definition and brief process of formation of test-tube baby.
Unit 9: Poisonous and nonpoisonous snakes	Hrs. theory 3
Objectives:	Content:
 List physical characteristics and habits of snakes. Identify specific characteristics of poisonous snakes in Nepal. Distinguish between poisonous and non- poisonous snakes. Identify the poisonous snakes commonly found in Nepal and tell their geographical distribution. Distinguish between a poisonous snakebite and a non-poisonous snakebite. Identify the nature and types of snake venum according to their effects in our body. 	 Physical characteristics and habits of snakes. Characteristics of poisonous snakes in Nepal. Distinguish between poisonous and non- poisonous snakes.Common poisonous snakes found in Nepal and their geographical distribution. Identification between a poisonous snakebite and a non-poisonous snakebite. Nature and types of snake venum according to their effects in our body. Snake bite, venum and its effects to our body. First-aid treatment of snake bite.

Co	ourse: Practical Zoology	Hrs. theory	Hrs.	lab	80
Un	it 1: Use of the microscope	Hrs. theory	Hrs.	lab	2
1. 2. 3. 4. 5. 6. 7.	jectives: Name different types of microscopes and their components. Handle a microscope properly. Observe the given slides under the microscope in different levels of magnification. Draw a labeled diagram of a microscope. View given slides under the microscope. Note the characteristic features of the given specimen. Identify the given slide and specimen.	Content: 1. Microscope, function observation technique		ent part	s and
	Classify the specimen and slide properly. aluation methods: practical performance ts, viva.	Teaching / Learning activ classroom instruction, der demonstration.			
kir	it 2: General study of the animal gdom jectives: Study given slides and specimens. Draw diagrams of given specimens. Write down the characteristic features of given specimens and slides. Identify the main features of each slide and specimen. Classify the specimens properly.	Hrs. theoryFigure 1Content:1. Different types of mu2. Protozoa:3. Rhizopoda : - Entamo4. Mastigophora : - Eug Leishmania5. Ciliata : - Paramecium6. Porifera: Sycon7. Coelenterata: Hydra8. Platyhelminthes:9. Cestodes : - Taenia so Echinococcus granul10. Trematoda : - Fascion11. Nemathelminthes: As Trichuris trichiura, E Ancyclostoma duoden bancrofti.12. Annelida: Earthworm13. Arthropoda:14. Crustacea: - Prawn, C 15. Arachnida: - Scorpion16. Insecta: - Anopheles a cycle), Pediculus, Cin 17. Mollusca: Unio, Lima18. Echinodermata: Starf	pebahistolytic lena, Giardia m aginata, Taer osus la hepatica caris lumbric caris lumbric caris lumbric therobious v nale, Wucher n, Leech. Crab. n, Spider and Culex (in mex tx and Pila	ca 1, nia soli coides, ermicu eria	laris,
		 18. Echnodermala: Start 19. Chordata: 20. Pisces: - Scoliodon, L 21. Amphibia: - Frog and 22. Reptilia: - Wall lizard Bungarus and Natrix 	<i>abeo rohita</i> l Toad l, <i>Viper</i> , Torto	oise,	

	23. Aves: - Crow and Pigeon.24. Mammalia: - Bat, Ant eater and Rabbit	
Evaluation methods: practical performance tests, viva.	Teaching / Learning activities and resources: classroom instruction, demonstration, return demonstration.	
Unit 3: Preparation of slides	Hrs. theory Hrs. lab 8	
 Objectives: Demonstrate how to tight a striated muscle specimen and stain the slide. Identify the nucleus of a striated muscle cell. Draw and label a diagram of striated muscle cell. Prepare a temporary slide of <i>Pediculus</i>/ flea . Draw labeled diagrams of preparations. 	 Content: 1. Preparation of temporary mounts of striated muscle, Bed bug, Flea and Louse. 	
Evaluation methods: practical performance tests, viva.	Teaching / Learning activities and resources: classroom instruction, demonstration, return demonstration.	
Unit 4: Dissection of mammals	Hrs. theory Hrs. lab 22	
 Objectives: Name the dissecting instruments and their uses. Dissect selected animal specimens. Dissect the systems of the animals provided. Draw a labeled diagram of each system of the animals provided. Examine the mammalian heart and use of a stethoscope to hear the heart beat. Evaluation methods: practical performance tests, viva. 	 Content: Instruments used for dissections Techniques of dissecting mammals Components of systems of mammals (digestive, arterial, venous, reproductive, brain) through direct observation of the preserved body. Use of stethoscope and measure of human blood pressure. Teaching / Learning activities and resources: classroom instruction, demonstration, return demonstration. 	
Unit 5: Life cycle of Anopheles and Culex mosquitoes and housefly.	Hrs. theory Hrs. lab 5	
Objectives: 1. Describe the different stages of life cycle of mosquitoes and houseflies in given specimens. 2. Identify the characteristics of different stages of life cycles. 3. Draw a labeled diagram of each stage of	 Content: 1. Stages in the life span of <i>Anopheles</i> and <i>Culex</i> mosquitoes and housefly. 2. Characteristics of the stages of each life cycles. 	

Ur	nit 5: Project work	Hrs. theory Hrs. lab 20
Objectives:		Content:
1.	Draw a labeled diagram of alimentary canal, kidney, brain etc.	1. Demonstration of chart of different organ systems of human (Alimentary canal,
2.	Identify the characteristics of different organs of human being.	Respiratory organs, Kidney, Heart and mammalian brain).
3.	Know about different stages in the lifecycle of parasites.	2. Demonstration the chart of lifecycle of <i>Plasmodium, Ascaris, Trichuris</i> and <i>Taenia</i> .
4.	Identify the animals and their characteristic features.	3. Field visit and report preparation.
5.	Prepare the report of field visit.	
Ev	aluation methods : practical performance	Teaching/ Learning activities and resources:
tes	ts, viva.	classroom instruction, demonstration, return
		demonstration, field visit and presentation.

Recommended Texts (Latest eds.)

- 1. Aggarwal, S. 1998. <u>A Textbook of Biology Part II.</u>Vikas Publishing House Pvt. Ltd., New Delhi, India.
- 2. Shukla, G.S. and Upadhyay, V.B. 1993. Economic Zoology. Rastogi Publications, Meerut, India.
- 3. Kotpal, R.L. Modern Textbook of Zoology, Invertebrates. Rastogi Publications, Meerut, India.
- 4. Kotpal, R.L. Modern Textbook of Zoology, Vertebrates. Rastogi Publications, Meerut, India.
- 5. Chatterjee, K.D. <u>Parasitology (Protozology and Helminthology)</u>. Medical Publishers, Calcutta, India.
- 6. Verma, P.S., Practical Zoology (Invertebrate)
- 7. Verma, P.S., Practical Zoology (Chordate)
- 8. Arora, D.R. and Arora B. Medical Parasitology. CBS Publisher and Distributors, New Delhi.
- 9. Lull, R.S. 1926. Organic Evolution. Macmillan, Newyork.

Reference Books

- 1. Paniker, C.K. 1993. <u>Textbook of Medical Parasitology</u>.Jaypee Brothers Medical Publishers (P) Ltd., New Delhi, India.
- Wilson, Kathleen J.W. and Waugh, A. 1998. <u>Anatomy and Physiology.</u> Churchill Living stone, U.K.
- 3. Singh, Dr. V. and Jain, Dr. D.K. 1998. Nootan Biology. Nageen Publication, Meerut, India.
- 4. Vidyarthi, R.D. and Pandey, P.N. 1998. <u>A Textbook of Zoology</u>. S. Chand and Company Ltd., New Delhi, India.
- 5. Gupta and Malik, Practical Zoology (Invertebrate)
- 6. Gupta and Malik, Practical Zoology (Chordate)

Botany

Year First Level Certificate Credit Hours: Theory: 120 Practical: 80 Assessment Marks: 100

Course Description

This course provides basic knowledge about botany, divided into eight units. The first unit gives general information about botany including different life components. The second unit tells about the structure and functions of a cell and its organelles including the cell reproduction. The third unit discusses the diversity of life, and includes basic information about algal plants, fungal plants, bacteria and viruses. Unit four provides information about life processes of plants such as diffusion, osmosis, photosynthesis, respiration and transpiration. Unit five teaches about heredity and variation. This unit also provides the information about genetic diseases/disorders in humans. Unit six provides information about the factors of our environment, their interrelationships, and effects of pollutants to human health. Unit seven includes information about selected medicinally and nutritionally important plants. Unit eight provides information about biotechnology and genetic engineering.

Course Objectives

At the end of the course, the student will be able to:

- Explain the scope of botany and its different branches.
- Explain the life components, cell structures and their functions.
- Explain the different physiological processes in a plant body.
- Explain the role of genes and their transmission to the progeny.
- Describe how environmental factors and pollutants affect our lives.
- Identify different members of plant kingdom based on their general characteristics.
- Describe the life cycle of selected plant species from algae and fungi.
- Identify the economic importance of viruses, bacteria, algae and fungi in the field of medicine.
- Describe the application of biotechnology in the field of medical science.

Evaluation methods: Oral and written tests, home assignments.

Teaching / Learning activities & resources: classroom instruction, illustrations, diagrams, visuals, textbooks, reference books.

Contents

Course: Botany	Hrs. theory 120 Hrs. lab 80
Unit 1: Introduction	Hrs. theory 10
Sub-unit 1.1: Definition, Scope and Different	Hrs. theory 3
Branches of Botany.	
Objectives:	Content:
1. Define Biology and Botany	1. Definition of biology and botany.
2. Explain the scope of pbotany.	2. Objectives and scope of botany.
3. Explain the difference between living	3. Difference between living organisms and
organisms and non-living things.	non-living things.
4. List the importance of plants in every day life.	4. Importance of plants.
Unit 1: Introduction	
Sub-unit 1.2: Different Branches of Botany and	Hrs. theory 3
Their Relationships with Other Science.	
Objectives:	Content:

1. 2. 3. 4. 5. 6.	List the major branches of botany and state the definition of each branch. Discuss the taxonomy related to morphology, anatomy, embryology, cytology, and genetics. Relate the evolution with paleontology. Relate the phytogeography with ecology. Correlate botany with physics/ chemistry and statistics. List the branches of botany based on the organisms.	 Different branches of botany. Correlation between different branches. Correlation between botany and other branches of sciences.
Un	it 1: Introduction	
-	b-unit 1.3: Life Components.	Hrs. theory 4
Ob	jectives:	Content:
	Define the terms cellular pool, macromolecules and micro-molecules. List the basic inorganic molecules of the living system.	 List molecules in living system. i) Water and its properties. ii) Minerals and their functions. iii) Biological Role Of Water
3. 4. 5. 6.	List the basic organic molecules of living system. List the function of carbohydrates, proteins, lipids and nucleic acids. Differentiate the essential amino acids and nonNessential amino acids. List the properties and important of water.	
-	it 2: Cell Biology	Hrs. theory 10
	b-unit 2.1: Introduction to Cell Biology	Hrs. theory 6
Ob	jectives:	Content:
1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11.	Explain about the discovery of cell. Describe the concept of cell theory. Differentiate between unicellular and multicellular organisms. Draw structure of typical plant cell Identify distinctions between a plant cell and an animal cell.	 Discovery of cell and cell theory. Plant cell and animal cell; their differences. Unicellular and multicellular organisms. Meaning of cell organelles and cell inclusions. Structural and functional study (occurrence, shape and size, number and function) of different types of cytoplasmic and nucleoplasmic contents . List the cell organelles and describe their function.

	and ribosome, microtubules and	
	microfilaments.	
13.	Cell wall and cell membrane	
	it 2: Cell Biology	
	b-unit 2.2: Cell Division	Hrs. theory 4
Objectives:		Content:
1. 2. 3. 4. 5. 6. 7. 8.	Define cell cycle, amitosis, mitosis and meiosis. Describe amitosis cell division. Explain the significance of amitosis. Describe the steps of mitotic cell division using a labeled diagram. Explain the significance of mitosis. Describe the steps of meiotic cell division with necessary sketches. Explain why meiosis is called reductional division and important in sexually reproducing organisms. Explain the significance of meiosis.	 Definition of cell cycle. Amitosis, mitosis and meiosis cell divisions. Differences between mitosis and meiosis cell divisions. Meiosis I and meiosis II
9.	Distinguish between mitosis and meiosis.	
Un	it 3: Diversity of life	Hrs. theory 40
Su	b-unit 3.1: Classification of plant kingdom	Hrs. theory 5
Ob	jectives:	Content:
1.	Describe the classification system of plant	1. Classification of plant kingdom (2 Kingdom
	kingdom into different categories.	and 5 Kingdom)
2.	Describe different taxonomic categories such as	2. Binomial nomenclature.
_	species, genus, family, order, class, division.	3. Taxonomic categories.
3.	Tell about how a plant can be placed in	4. Hierarchic system in classification.
TT	hierarchic system in classification.	
	it 3: Diversity of life	П. 4. 5
	b-unit 3.2: General characteristics of different	Hrs. theory 5
	int groups.	Contents
	jectives:	Content: 1. The unicellular plant and multicellular plant.
	List the characteristics of algae. List the characteristics of fungi.	 The unicellular plant and multicellular plant. The characteristics of algae, fungi and
	List the characteristics of Bryophytes.	6 7 6
э. 4.	List the l differences between pteridophytes and	bryophytes.3. Morphological characteristics of
4.	gymnosperms based on morphology.	pteridophytes.
5.	Identify the type of leaves in dicot and monocot	 Morphological characteristics of
5.	plants.	gymnosperms.
6.	Differentiate dicot root from monocot root.	5. General characteristics of angiosperms.
7.	Describe the structural differences between a	6. Differences between monocot and dicot
	monocot plant and a dicot plant.	plants.
	1 1	7. Parts of flowering plants.
Un	it 3: Diversity of life	
	b-unit 3.3: Algae	Hrs. theory 4 Hrs. lab 2
	jectives:	Content:
1.	Define algae.	1. Structure of <u>Spirogyra</u> .
2.	Differentiate chlorophyceae, pheophyceae and	2. Life cycle of <u>Spirogyra</u> .
	rhodophyceae algae.	3. Medicinal values of algae.
3.	Classify Spirogyra.	

1 /1	Describe about vagetative accover and cover	
4.	Describe about vegetative, asexual and sexual	
5	reproduction in <u>Spirogyra</u> .	
	Explain what conjugation is.	
6.	Describe the life cycle of Spirogyra using a	
_	labeled diagram.	
7.	Describe the medicinal values of different	
	kinds of algal plants.	
Un	it 3: Diversity of life	
	b-unit 3.4: Fungi	Hrs. theory 8
	jectives:	Content:
1.	Define fungi.	1. Morphology of a typical fungal plant.
2.	Compare fungal plant with algal plant.	2. Types of reproduction in brief of
3.	Describe the morphology of a typical fungal	reproduction in brief:
	plant.	3. Vegetative
4.	Identify the hypha and mycelium of a fungus.	4. Asexual
5.	Describe briefly the different methods of	5. Sexual
.	asexual reproduction in Fungi.	6. Life cycle of <i>Mucor</i> , <i>Penicillium</i> and
6.	Tell about the stages of sexual reproduction	Aspergillus.
0.	(i.e. plasmogamy, karyogamy and meiosis).	7. Differences between edible and poisonous
7.	Distinguish between edible and poisonous	mushroom.
/ .	mushrooms.	
8.		8. Economic importance of fungi especially in the field of human health and medicine.
0.	List the fungal plants, which are used in	
	antibiotic production.	9. Define lichen, mention their types and
9.	List the fungal plants, which cause diseases in	importance.
10	man.	
10.	Define lichen, mention their types and	
1	T	
	Importance.	
_	it 3: Diversity of life	
Su	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc)	Hrs. theory 9 Hrs. lab 6
Su Ob	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives:	Content:
Su Ob 1.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria.	Content: 1. Structure and types of bacteria.
Su Ob 1. 2.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria.	Content:1. Structure and types of bacteria.2. Differences between gram positive and
Su Ob 1. 2.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria.
Su Ob 1. 2. 3.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria - List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria.
Su Ob 1. 2.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria - List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria.
Su Ob 1. 2. 3. 4.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria - List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria.
Su Ob 1. 2. 3. 4.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate.
Su Ob 1. 2. 3. 4. 5. 6.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria - List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria.
Su Ob 1. 2. 3. 4. 5. 6.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities
Su Ob 1. 2. 3. 4. 5. 6.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria - List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities harmful activities Structure of <u>Nostoc</u>.
Su Ob 1. 2. 3. 4. 5. 6. 7. 8.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities harmful activities Structure of <u>Nostoc</u>. Describe about vegetative and asexual
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities harmful activities Structure of <u>Nostoc</u>. Describe about vegetative and asexual reproduction in <u>Nostoc</u>.
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria. Classify Nostoc.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities Structure of Nostoc. Describe about vegetative and asexual reproduction in Nostoc. Describe the importance of Nostoc in
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define <i>Nostoc</i> . Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities harmful activities Structure of <u>Nostoc</u>. Describe about vegetative and asexual reproduction in <u>Nostoc</u>.
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define <i>Nostoc</i> . Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual reproduction in <u>Nostoc</u> .	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities Structure of Nostoc. Describe about vegetative and asexual reproduction in <i>Nostoc</i>. Describe the importance of Nostoc in
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12.	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual reproduction in <u>Nostoc</u> . Describe the importance of Nostoc in agriculture.	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities Structure of Nostoc. Describe about vegetative and asexual reproduction in Nostoc. Describe the importance of Nostoc in
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12. Un	 it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual reproduction in <u>Nostoc</u>. Describe the importance of Nostoc in agriculture. it 3: Diversity of life 	 Content: Structure and types of bacteria. Differences between gram positive and gram negative bacteria. Factors influencing the growth of bacteria. Koch's postulate. Economic importance of bacteria. beneficial activities Structure of <u>Nostoc</u>. Describe about vegetative and asexual reproduction in <u>Nostoc</u>. Describe the importance of <u>Nostoc</u> in agriculture.
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12. Un Su	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual reproduction in <u>Nostoc</u> . Describe the importance of Nostoc in agriculture. it 3: Diversity of life b-unit 3.6: Virus	Content: 1. Structure and types of bacteria. 2. Differences between gram positive and gram negative bacteria. 3. Factors influencing the growth of bacteria. 4. Koch's postulate. 5. Economic importance of bacteria. 6. beneficial activities 7. harmful activities 8. Structure of Nostoc. 9. Describe about vegetative and asexual reproduction in Nostoc. 10. Describe the importance of Nostoc in agriculture.
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12. Un Su Ob	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual reproduction in <u>Nostoc</u> . Describe the importance of Nostoc in agriculture. it 3: Diversity of life b-unit 3.6: Virus jectives:	Content: 1. Structure and types of bacteria. 2. Differences between gram positive and gram negative bacteria. 3. Factors influencing the growth of bacteria. 4. Koch's postulate. 5. Economic importance of bacteria. 6. beneficial activities 7. harmful activities 8. Structure of Nostoc. 9. Describe about vegetative and asexual reproduction in Nostoc. 10. Describe the importance of Nostoc in agriculture. Hrs. theory 9 Hrs. lab 2 Content:
Su Ob 1. 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 12. Un Su Ob	it 3: Diversity of life b-unit 3.5: Monera (Bacteria & Nostoc) jectives: Bacteria -List the characteristics of bacteria. Describe the structure of bacteria. Differentiate Gram positive bacteria from Gram negative bacteria. Describe the types of bacteria. Define toxin. State Koch's postulate. List the harmful and beneficial aspects and activities of bacteria. Nostoc- Define Nostoc. Differentiate Bacteria and Cyanobacteria. Classify Nostoc. Describe about vegetative and asexual reproduction in <u>Nostoc</u> . Describe the importance of Nostoc in agriculture. it 3: Diversity of life b-unit 3.6: Virus	Content: 1. Structure and types of bacteria. 2. Differences between gram positive and gram negative bacteria. 3. Factors influencing the growth of bacteria. 4. Koch's postulate. 5. Economic importance of bacteria. 6. beneficial activities 7. harmful activities 8. Structure of Nostoc. 9. Describe about vegetative and asexual reproduction in Nostoc. 10. Describe the importance of Nostoc in agriculture.

 Describe the nature and chemical composition of viruses. Describe the structure and function of a bacteriophage. Tell about mode of nutrition in viruses. Mention the types of viruses on the basis of host and nucleic acid - bacterial virus. Compare plant and animal viruses. Differentiate DNA virus from RNA virus. Tell the effect of retrovirus in man. List the agents responsible for transmission of viruses. Tell about the method of multiplication (Lytic and Lysognic cycle) of viruses. List the economic importance of viruses in the field of human health and medicine. 	 multiplication (Lytic and Lysognic cycle) of viruses Transmission of viruses. Economic importance of viruses. 				
Unit 4: Life Process (Physiology)	Hrs. theory 20 Hrs. lab 10				
Sub-unit 4.1: Diffusion	Hrs. theory 3 Hrs. lab 2				
Objectives:	Content:				
1. Define diffusion.	1. Definition and process of diffusion.				
2. Tell about the factors that affect diffusion.	2. Factors affecting diffusion.				
3. List the significance of diffusion on plant life	3. Significance of diffusion.				
and animal body.					
Unit 4: Life Process (Physiology)	Hus theory A				
Sub-unit 4.2: Osmosis	Hrs. theory 4				
Objectives:	Content: 1. Definition of osmosis.				
1. Define osmosis (including endo-and exoosmosis), osmotic pressure and osmotic	 Definition of osmosis. Type of osmosis. 				
potential.	 Type of osmosis. Demonstration of osmosis by potato- 				
2. Tell the meaning of hypertonic and hypotonic	osmoscope & egg membrane.				
solution, isotonic solution	 Plasmolysis and deplasmolysis. 				
3. Tell the meaning of turgid and flaccid cells.	5. Factors which affect osmosis				
4. Describe stomata movement.	6. Significance of osmosis.				
5. Describe osmosis in living cells.					
6. Factors which affect osmosis					
7. List the significance of osmosis.					
8. Relate the osmosis with plasmolysis.					
Unit 4: Life Process (Physiology)					
Sub-unit 4.3: Transpiration	Hrs. theory 4				
Objectives:	Content:				
1. Define transpiration.	1. Definition of transpiration.				
2. Describe types of transpiration.	2. Factors affecting transpiration.				
3. Describe the mechanism of transpiration in	3. Types and significance of transpiration.				
plants.	4. Demonstration of transpiration by:				
4. Describe unequal transpiration in dicot leaf.	a. Beljar method				
5. Mention the factors that affect transpiration.	5. Cobalt chloride paper method.				
6. Describe the role of stomatal transpiration.					
7. List the significance of transpiration.					
Unit 4: Life Process (Physiology)	Ure theory 3				
Sub-unit 4.4: Photosynthesis	Hrs. theory 3				

Ob	jectives:	Co	ntent:
1.	Define photosynthesis.	1.	Definition of photosynthesis.
2.	Identify the sites of photosynthesis.	2.	Identify the sites of photosynthesis.
3.	Discuss the importance of photosynthesis.	3.	Discuss the importance of photosynthesis.
<i>4</i> .	Describe general layout of process of	4.	Describe general layout of process of
	photosynthesis.	''	photosynthesis.
	photosynthesis.	5.	Experiments:
		6.	To demonstrate that CO_2 is necessary for
		0.	photosynthesis
		7.	
		/.	To demonstrate that O ₂ is evolved during
		0	photosynthesis.
		8.	To demonstrate that chlorophyll is
			necessary for photosynthesis
TT		9.	Importance of photosynthesis.
	it 4: Life Process (Physiology) b-unit 4.5: Respiration and Fermentation	U	s theory 6
	jectives:		s. theory 6 ntent:
	5		
1.	Define respiration.	1.	Definition of respiration.
2.	Explain aerobic and anaerobic respiration	2.	Types of respiration.
2	with examples.	3.	Experiments
3.	Identify the sites of respiration.	4.	To demonstrate aerobic respiration
4.	Differentiate anaerobic respiration from	5.	To demonstrate anaerobic respiration.
-	aerobic respiration.	6.	Differences between anaerobic and aerobic
5.	Define fermentation.	-	respiration.
6.	Name the organisms involved in alcoholic	7.	Definition of fermentation.
_	fermentation.	8.	Importance of fermentation in our daily life.
	Correlate fermentation with our daily life. it 5: Genetics	II.	a Ahaanna 11
	b-unit 5.1: Heredity and Variation		s. theory 11 s. theory 2
	jectives:	-	ntent:
	Define heredity and Types of variation	1.	
1. 2.	Tell about the causes of variations	$\begin{vmatrix} 1.\\ 2. \end{vmatrix}$	Definition of heredity and types of variation Difference between heredity and variation
2. 3.		$\begin{vmatrix} 2 \\ 3 \end{vmatrix}$	-
	Tell the terms: alleles, genotype, etc.	3.	Difference between clone and offspring
4.	Differentiate clone from offspring.		
	it 5: Genetics b-unit 5.2: Mendel's Law of Inheritance	Ц.	s. theory 2
	jectives:		ntent:
1.	Mention why Mendel chose pea for his	1.	Description of Mendel's monohybrid cross
1.	experiment.	1.	and dihybrid cross.
2.	Tell an idea of gametogenesis on the basis of	2.	Mendel's law of inheritance
∠.	6 6	∠.	
3.	separation of allelic gene. List the ratio of monohybrid cross and		
5.			
I.I	dihybrid cross.		
	it 5: Genetics h unit 5 3: Nucleia Acid and Constia Disease	11	s theory A
	b-unit 5.3: Nucleic Acid and Genetic Disease		s. theory 4 ntent:
	jectives:		
1.	Name the compounds that build up DNA and	1.	8
_	RNA Differentiate DNA and RNA.	2. 3.	Difference between DNA and RNA.
2.	\mathbf{L}	1 1	Compounds that build up DNA and RNA

3.	List the genetic diseases found in human	4. Chromosomal disorder (Down's Syndrome,				
	being.	Edward's Syndrome, Turner's Syndrome				
4.	Chromosomal disorder (Down's Syndrome,	and Kleinfeilter's Syndrome)				
	Edward's Syndrome, Turner's Syndrome and	5. Gene Disorder (Albinism, Alzheimer				
	Kleinfeilter's Syndrome)	Disease, Daltonism, Haemophilia)				
5.	Gene Disorder (Albinism, Alzheimer					
	Disease, Daltonism, Haemophilia)					
Un	it 5: Genetics					
	b-unit 5.4: Determination of Sex	Hrs. theory 2				
	jectives:	Content:				
1.	Tell about autosome and sex chromosome. 1. Description of autosomes and sex-					
2.	Describe the concept of sex determination in	chromosomes				
2.	mammals, insects, birds and reptiles.	2. Types of sex-determination :				
3.	Explain why the female has no responsibility	 Hypes of sex-determination : Heterogametic males 				
5.						
4	in determining the sex of a child in humans.					
4.	Tell the concept of heterogametic male and	• XX female - XO male				
	heterogametic female.	• Heterogametic females				
		\circ ZO female - ZZ male				
		• ZW female - ZZ male				
Un	it 6: Environmental Biology	Hrs. theory 10 Hrs. lab 6				
	b-unit 6.1: Ecology	Hrs. theory 4 Hrs. lab 6				
	jectives:	Content:				
	Define ecology and ecosystem.	1. Definition of ecology and ecosystem.				
	List the abiotic factors of ecosystem	2. Structural and functional component of				
	List the biotic factors of ecosystem.	ecosystem grassland and pond ecosystem				
4.	Write structural and functional aspects of	3. Describe food chain, food web and				
	grassland and pond ecosystem	ecological pyramid.				
5.	Mention the main source of energy in an	coological pyrainid.				
5.						
I In	ecosystem it 6: Environmental Biology					
	b-unit 6.2: Pollution of Water and Air.	Hrs. theory 3				
	jectives:	Hrs. theory 3 Content:				
	Define pollution.	1. Definition of pollution and pollutants.				
1. 2	List biodegradable pollutants.					
		2. Types of pollutants.				
3.	List non- biodegradable pollutants.	3. Source of water pollution, their effect and				
4.						
1	List sources of water pollutants.	preventive measures.				
5.	Identify the causes of water pollution.	4. Source of air pollutants, their effect on				
1	Identify the causes of water pollution. List the preventive measures to control water	4. Source of air pollutants, their effect on living organisms and preventive measures				
5. 6.	Identify the causes of water pollution. List the preventive measures to control water pollution.	4. Source of air pollutants, their effect on				
5. 6. 7.	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants.	4. Source of air pollutants, their effect on living organisms and preventive measures				
5. 6. 7. 8.	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution.	4. Source of air pollutants, their effect on living organisms and preventive measures				
5. 6. 7. 8.	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control	4. Source of air pollutants, their effect on living organisms and preventive measures				
5. 6. 7. 8. 9.	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control air pollution	4. Source of air pollutants, their effect on living organisms and preventive measures				
5. 6. 7. 8. 9. Un	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control air pollution it 6: Environmental Biology	4. Source of air pollutants, their effect on living organisms and preventive measures of air pollution.				
5. 6. 7. 8. 9. Un Su	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control air pollution it 6: Environmental Biology b-unit 6.3: Ecological Imbalances.	 4. Source of air pollutants, their effect on living organisms and preventive measures of air pollution. Hrs. theory 3 				
5. 6. 7. 8. 9. Un Su Ob	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control air pollution it 6: Environmental Biology b-unit 6.3: Ecological Imbalances. jectives:	 4. Source of air pollutants, their effect on living organisms and preventive measures of air pollution. Hrs. theory 3 Content: 				
5. 6. 7. 8. 9. Un Su	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control air pollution it 6: Environmental Biology b-unit 6.3: Ecological Imbalances. jectives: Explain the theory of the "green house	 4. Source of air pollutants, their effect on living organisms and preventive measures of air pollution. Hrs. theory 3 Content: 1. Description on green house effect, acid rain 				
5. 6. 7. 8. 9. Un Su Ob	Identify the causes of water pollution. List the preventive measures to control water pollution. List the sources of air pollutants. List the effect of air pollution. Mention the preventive measures to control air pollution it 6: Environmental Biology b-unit 6.3: Ecological Imbalances. jectives:	 4. Source of air pollutants, their effect on living organisms and preventive measures of air pollution. Hrs. theory 3 Content: 				

8. 9.	Tell the consequences of green house effect. Discuss the significance of green house effect, and explain why many scientists believe it will create a global crisis. Tell how acid rain is formed List the harmful effects of acid rain List the importance of the ozone layer for living organisms. Tell how some scientists believe the ozone layer is going to deplete. Describe the consequences of depletion of the ozone layer.			
	it 7: Economic Botany	Hrs. theory	10	Hrs. lab 4
	b-unit 7.1: Medicinal plants	Hrs. theory	4	Hrs. lab 4
	jectives:	Content:		
2. 3. 4. 5. 6.	List the habit and distribution of medicinal plants. List the uses of medicinal plants. Identify the parts of the plant which have medicinal value. Name the chemical compounds from particular medicinal plants. Tell what form of plant part is used for the treatment of specific cases.	medicinal medicinal Menth Adhata Adhata Zinger Rauwa Cinnar Datura Papav Santal Cochia Eletter Mellia Ocimu Aloe v Cordya	value and v plants. : a arvensis (od vasica (A officinalis olfia serpen momum zyla a stromoniu er somnifer um album (cum luteum cis cordamo azedarach m sanctum era (Gheuk achita indic ceps sinens	Asuro) (Aduwa) tina (Sarpagandha) enica (Dalchini) um (Dhaturo) rum (Opium) Shreekhanda) (Colchium) onum (Alaichi) (Bakenu) (Tulsi) rumari)
IIn	it 7: Economic Botany	y orenis	naiazirea	(1 difendine)
Su Cr Pla	b-unit 7.2: Nutritional Values of Cereal ops, Fruits, Vegetables and Oil Yielding unts.	Hrs. theory	4	
Ob	jectives:	Content:		
1.	Identify the nutritional value of cereal crops.			s of cereal crops; fruits,
2.	List the nutritional value of vegetables.			elding plants:
3.	Tell the nutritional value of fruits.		ps - rice, w	heat, maize, gram, and
4.	Tell the nutritional value of oil yielding plants.	bean. 3. Fruits- ma	ngo hanan	a, grape, pear and
5.	Compare the nutritional value of rice maize, white, gram and bean.	orange 4. Vegetables	s- cauliflow	ver, cabbage, tomato
6. 7.	Compare the nutritional value of mango, grape, pear banana and orange. Compare the nutritional value of potato, cauliflower, cabbage, tomato and brinjal.	5. Oil yieldin caster		nustard, ground nut,

8.	Compare the nutritional value of mustard,	
	groundnut and caster.	
Un	it 7: Economic Botany	
Su	b-unit 7.3: General Concept on	Hrs. theory 2
Etl	nnobotany.	
Ob	jectives:	Content:
2.	Define the term 'ethnobotany'. Describe the kinds of information included in traditional knowledge. Discuss ways of gathering traditional	 Definition of ethnobotany. Importance of ethnobotany in the field of medicine. A survey questionnaire for data collection, if the plant is used as medicing.
	knowledge. Discuss the value and importance of traditional knowledge.	if the plant is used as medicine.
	List the ways ethnobotany is useful in the field of medicine. Discuss how to gather information about the	
	use of local plants in medicine.	
	it 8: Biotechnology	Hrs. theory 8
	b-unit 8.1: Introduction to Biotechnology	Hrs. theory 5
	jectives:	Content:
2. 3. 4.	Define biotechnology. List the branches of biotechnology. List the scope of biotechnology. Describe the application of biotechnology in medicine, agriculture and fermentation technology.	 Definition and scope of biotechnology. Branches of biotechnology. Application of biotechnology in: Medicine Agriculture Fermentation.
5.	Concept of plant tissue culture.	 Bio fertilizer and organism used as bio fertilizer. Plant tissue culture and its type.
Un	it 8: Biotechnology	
	b-unit 8.2: Genetic Engineering	Hrs. theory 3
	jectives:	Content:
2. 3.	Define genetic engineering Requirement of Genetic engineering Describe the applications of genetic engineering in the field of medicine.	 Definition of genetic engineering and recombinant DNA technology. Steps of genetic engineering. Application of genetic engineering.
4.	Tell about the possible dangers of genetic engineering.	4. Possible dangers of genetic engineering.

Botany Practical

Evaluation methods: performance observation, written exams, viva.

Teaching / Learning activities & resources: Classroom instruction, demonstration,

return demonstration, slide preparation, microscopic observation

Course: Botany Practical		Hrs. Practical :	80		
Un	it 1: Introduction to the compound microscope	Hrs. Practical	6		
Ob	jectives:	Content:			
	 Define compound microscope. Differentiate between simple and compound microscope. Tell the names of lenses used in a compound microscope. List different parts of a compound microscope and their uses. Calculate the magnifying power of a compound microscope in different combinations of objective lens/eye-piece lens. Describe the way of handling a compound microscope. Draw a well labeled diagram of a compound microscope by observation. 	Methods o	nd microscope. f handling of a microscope.		
Un	it 2: Cell biology	Hrs. theory	Hrs. lab		
Su	b-unit 2.1: Temporary slide preparation of	Hrs. theory	Hrs. lab 8		
	int cells.				
	jectives:	Content:			
 2. 3. 4. 5. 	List the apparatus required to prepare temporary slides of plant cells. List the chemicals required to prepare temporary slides of plant cells. List the function of safranine and glycerine. Describe the method of slide preparation from the epidermal layer of onion scale, <i>Tradescantia</i> leaf, <i>Hydrilla</i> leaf, <i>Geranium</i> leaf. Compare the cell structure of Onion scale, <i>Tradescantia</i> leaf, <i>Hydrilla</i> leaf, and <i>Geranium</i> leaf.	of plant cells.	n of temporary slides eellular structure of		
6. 7.	List the characteristics of the cellular structure of each. Describe the method to peel out the epidermal				
8.	layer in each case. Tell why you should use glycerin instead of water when mounting a temporary slide.				

	slides under the microscope.				
Uni	t 2: Cell biology	Hrs. theory	Hrs.	lab	
Sub	o-unit 2.2: Different stages of mitosis and	Hrs. theory	Hrs.	lab	4
mei	iosis divisions from permanent slides.				
Obj	ectives:	Content:			
 2. 3. 4. 5. 6. 7. 8. 9. 	plants. Describe different stages of meiosis by observation of permanent slides under the compound microscope. Tell the time period of meiotic cell division of the plant. Name the type of cell where meiotic cell division occurs. Draw figures of mitosis and meiosis by observing		ent stages of m th the help of p		
	under the microscope.				
Uni	t 3: Biodiversity	Hrs. theory	Hrs. lab		
	o-unit 3.1: Monera	Hrs. theory	Hrs. lab	1	0
	ectives:	Content:			
 2. 3. 4. 5. 6. 7. 8. 9. 10. 11. 	Define bacteria Describe the nature of bacterial cells. List the components of the bacterial cell wall. list the required material and chemicals for Gram staining. List the role of chemicals used in Gram staining. differentiate Gram positive bacteria from Gram negative bacteria. Draw diagrams of bacteria by observing under the microscope. Describe the vegetative structures of <i>Nostoc</i> . Explain the function of heterocyst in <i>Nostoc</i> . Describe the reproductive stages of <i>Nostoc</i> . Draw figures of these as observed under the microscope. Give the systematic position of <i>Nostoc</i> .	cells 2. The metho bacteria. 3. Classificat	and nature of d of Gram stai ion, vegetative luctive stages o	ning o struc	of ture,
Uni	t 3: Biodiversity	Hrs. theory	Hrs. lab		
	unit 2.2. Vagatativa atrustura and	Hrs. theory	Hrs. lab	2	
Sut	o-unit 3.2: Vegetative structure and	iner areary		_	
	roductive stages of Spirogyra			-	

10. Draw figures of each of the plants which are observed during the lab experience.	
 > Gynoecium 10. Draw figures of each of the plants which are 	
> Androecium	
> Corolla	
> Calyx	
9. Describe the parts of a flower:	
8. Differentiate dicot plants from monocot plants.	5. Different parts of a typical flower.
 Differentiate gymnosperms from angiosperms. 	monocot plants.
and angiosperms.	4. The different parts of dicot plants and
cryptograms.6. List the morphological features of gymnosperm	pteridophytes with reference to ferns (<i>Dryopteris</i>).
5. Explain why pteridophytes are also called vascular	3. Morphological features of
sporophyte.	pinus.
4. Discuss the concepts of gametophyte and	gymnosperms with reference to
3. Differentiate bryophytes from pteridophytes.	2. Morphological features of
2. List the characateristics of pteridophytes.	with reference to Marchantia.
1. List the characteristics of bryophytes.	1. The characteristics of bryophytes
Objectives:	Content:
Gymnosperms and Angiosperms	
Sub-unit 3.4: Bryophytes, Pteridophytes,	Hrs. theory Hrs. lab 14
Unit 3: Biodiversity	Hrs. theory Hrs. lab
mushrooms by observing under the microscope.	
 Draw figures of both poisonous and nonpoisonous 	
4. Differentiate between poisonous mushrooms and edible mushrooms.	
1 Differentiate between neisenous much some ond	
Penicillium	mushrooms.
> Aspergillus	2. The morphological features of
Mucor	
stages of:	Penicillium
3. Describe the vegetative structure and reproductive	Aspergillus
clestothecium.	> Mucor
2. Tell the meaning of mycelium, hypha, metuli,	reproductive stages of:
1. Define fungi.	1. The vegetative structure and
Objectives:	Content:
reproduction of selected fungi	
Sub-unit 3.3: Vegetative structure and	Hrs. theory Hrs. lab 8
Unit 3: Biodiversity	Hrs. theory Hrs. lab
5. Give the systematic position of Spirogyra.	
microscope.	
 Describe the reproductive stages of spirogyta. Draw figures of these as observed under the 	
3. Describe the reproductive stages of Spirogyra.	
2. Tell the reasons why Spirogyra is so called.	and reproductive stages of Spirogyra.

Sub-unit 4.1: Demonstration of physiological	Hrs. theory	Hrs. lab	10
experiments Objectives:	Content:		
 Define diffusion, osmosis, photosynthesis, respiration and transpiration. Describe the types of : Osmosis Respiration Transpiration List the required material to demonstrate each experiment. Tell the concept of hypotonic, hypertonic and isotonic solution. Discuss the procedure to demonstrate each physiologic experiment. Describe the result and conclusion of each experiment. List the precautions and possible risks of each experiment. Draw the necessary figures to show the demonstration of each experiment. 	 crystal. 2. The osmosis p membrane me osmoscope. 3. Oxygen is evo photosynthesis 4. Carbon dioxid photosynthesis 5. Carbon dioxid aerobic respira 6. Chlorophyll is photosynthesis 7. Carbon dioxid anaerobic resp 8. Transpiration 1 Cobalt chlorid 	thod and by po lved during s. e is necessary for s (Moll's experin e is evolved durition. essential for s. e gas is evolved	tato or nent) 'ing I during hod, elation
Unit 5: Ecology	Hrs. theory	Hrs. lab	
Sub-unit 5.1: Ecosystem and Adaptation features of	Hrs. theory	Hrs. lab	6
selected plants			
Objectives:	Content:		
 Discuss aquatic ecosystem List the morphological features of xerophytic plants (eg. <i>Pinus</i>). Describe how xerophytic plants compensate for insufficient water supply. List the morphological features of hydrophytic plants (eg. <i>Hydrilla</i>). Tell about the mode of water conduction by hydrophytic plants. 	of aquatic ecos to pond or aqu	t structural com system with refo arium. d hydrophytic p	erence
Unit 6: Genetics	Hrs. theory	Hrs. lab	
Sub-unit 6.1: Structure of DNA	Hrs. theory	Hrs. lab	6
Objectives:	Content:		-
1. Describe the structural components of DNA.		rick model of D	NA
 List the components of a nucleotide. Explain the concept of the base pairing rule. Describe the purine and pyrimidine compounds of a DNA molecule. List the functions of DNA. Draw a figure of the Watson and Crick model of DNA by observation of a model. 			

Sub-unit 6.2: Survey of some human heredity characteristics	Hrs. theory Hrs. lab 6
Objectives:	Content:
 List some hereditary characteristics of humans. Describe the role of dominant or recessive genes to express certain characteristics. Discuss the methods of data collection of hereditary characteristics. Calculate the frequency of each hereditary characteristic in certain populations. 	 The role of dominant or recessive genes to express some hereditary characteristics of humans.

Recommended Texts

- 1. Singh.A.G, Karky. B, Hamal.J.P., Nag. P. **Botany** for CTEVT, Ayam Publication, Kathmandu, Nepal.
- 2. Singh.A.G, Karky. B, Hamal.J.P., Nag. P., Joshi. S. **Practical Botany** for CTEVT, Ayam Publication, Kathmandu, Nepal.
- 3. Dutta, A.C., Classbook of Botany, Oxford University Press, Calcutta.
- 4. Alexopolous, C.J., Introductory Mycology, John Wiley and Sons, New York.
- 5. Pandey, B.P., Economic Botany, S. Chand and Company Ltd., New Delhi.
- 6. Salisbary and Ross, <u>Plant Physiology.</u>
- 7. <u>Medicinal plants of Nepal</u>, HMG of Nepal.
- 8. Gangulee, M.C. and Kar, A.K., College Botany Vol. II New Central Book Agency, Calcutta.
- 9. Kochhar PL Genetics and Evolution, Ratan Prakashar Mandor, Delhi.
- 10. Shah and Sonhas, <u>Cytogenics, Plant Breeding and Evolution</u>, Vikash publishing House Pvt. Ltd. V. P.
- 11. Ranjtkar H.D. <u>Laboratory Manual and Viva-voce for Proficiency Certificate Level</u>, AK Ranjitkar, Kathmandu.
- 12. Pandey, B.P., Modern Practical Botany Vol. I and II, S Chand and Company Pvt. Ltd., New Delhi.

References Books:

- 1. Sharma, O.P. and Agrawal, V.K.S., Cell biology, Genetics, Evolution and Ecology.
- 2. Bhattic, K.N. and Khanna, Modern Approach to Botany, Surya Publication, Jalandhar.
- 3. Saxena A.L. and Sarabhai, R.P., <u>A Textbook of Botany</u>, Batan Prakashan Mandor.
- 4. Bilgrami, K.S., Shrivastava, L.M., and Shremali, J.L., <u>Fundamentals of Botany</u>, Vani Educational Books.
- 5. Dey, N.C., and Dey, T.K., <u>Medical Bacteriology</u>, Messers Allied Agency.
- 6. Sharma, D.P., <u>Hill's Economic Botany</u>, Tata Mc Graw-Hill Publishing Company Limited, New Delhi.
- 7. Winchester, A.M., Biology and Its Relation to Mankind 3rd ed.
- 8. Singh, V., and Sinha, S., Cytogenetics.
- 9. Man Dhar, C. L., Introduction to Plant Virus, S. Chand and Company Ltd., Delhi.

Mathematics & Statistics

Year First Level Certificate Total Hours:200Theory Hours:160Practical Hours:40Assessment Marks:100

Course Description

The course is divided into three parts: (a) Elementary Mathematics, (b) Elementary Statistics and (c) Practical on elementary Statistics and Computer. Part one of this course prepares the student to use mathematical skills necessary for application of medical computations, application of research and statistical interpretations, and for managing the mathematical questions of everyday life. Part two provides a basic overview of the purpose and process of research, a discussion of scientific process, and principles of research methodology in statistics. Part three enables the students to apply statistical methods to the interpretation of data related to public health services using basic computer skills.

Course Objectives

On completion of this course the student will be able to:

- Apply mathematical Skills to solve medical problems and interpret research data.
- Use vital statistic terminology to discuss public health issues.
- Explain the function and value of research.
- Describe the process and methodology of research.
- Apply mathematical formulas to interpret research data.
- Demonstrate the process of report writing.

Recommended Texts

Bajracharya, D.R.& et al., <u>Basic Mathematics</u>, for grade XI and XII National Book Centre, Kathmandu.
Mahajan B.K. <u>Method of Biostaticstics</u>, (16th edition) park's text book of PSM 2003
Pradhan, J. B. & Pantha, B. R. <u>Integrated Mathematics for Health Science</u>. Sukunda Pustak Bhavan, <u>Bhotahity, Kathmandu</u>.
MS-DOS Manual, Microsoft.
MS-Windows Manual, Microsoft.

Evaluation methods: Written assignments to solve related problems, written examination and practical examination for computer

Teaching / **Learning activities and resources:** Charts, models, graph boards, diagrams classroom instruction, teacher led discussion, demonstration of solutions, illustration through practical examples.

Part A: Elementary Mathematics

Course: Mathematics and Statistics	Hrs. theory 160
Unit 1: Elementary Mathematics	Hrs. theory 66
Sub-unit 1.1: Set theory and real number system	Hrs. theory 6 Hrs. lab
Objectives:	Content:
 Define and denote sets. Find subsets of a set and represent the sets in venn diagrams. 	• The concept of sets, specification of sets, representation and types of sets, venn diagrams.

 Find the union, intersection, complement and difference of given sets. Define cardinality of a finite set Solve verbal problems using set operations. Prove algebra of sets Define real numbers, absolute value, open and closed intervals and inequalities. Use the concept of set in selected problems. Sub-unit 1.2: Function and graph Objectives: Define a function Classify functions. Identify the different functions. Define domain and range of relation 	 Proof of the Algebra of sets, De-Morgan's law Problems related to cardinality of sets. Set operation, set of numbers, Cartesian products and relation, domain and range of relation. Real number system and the types of numbers, real numbers line, absolute value, open and closed intervals, inequalities. Hrs. theory 6 Content: Functions and their inverse and related problems. Algebraic only. Domain and range (excluding inverse and related problems.
	composite function)
Sub unit 1 2. Domentation combined and all it is	Exponential and Logarithmic functions Hrs. theory 9
Sub-unit 1.3: Permutation , combination and binomial theorem.	111S. theory 9
Objectives:	Content:
 Concept of Basic principles of counting. Define the permutation {P(n,r)}. Use of different cases of permutation and Problem relating to permutation(simple cases only). Define the combination {C(n,r)} and problem relating to combination (simple cases only) Define binomial expression and Binomial theorem. 	 Introduction of basic counting principle Definition of permutation Formula for finding permutation of n – objects taken r at a time. Application of formula in related problems. Permutation of repeated use of same objects in an arrangement. Meaning of combination. Binomial theorem(without proof) Finding general term , middle term/s, binomial coefficients and their properties.
Sub-unit 1.4: Matrices and determinants	Hrs. theory 9
Objectives:	Content:
 Define the term matrix. Write the rows, columns and order of the matrices. Classify matrices according to their properties. Define the addition and multiplication of matrices (of order m x n, with its different types in 3x3 order). Define a determinant and list the properties of a determinant. Define the inverse of a matrix. 	 Definition of matrix and its notation and order Types of matrices and simple algebra of matrices. Transpose ,Adjoint and inverse of a matrix and related problems. Definition of a determinant. Minors and cofactors Properties of determinants. Application of matrix and determinant to solve linear system of equation (inverse of matrix and Cramer's Rule) Hrs. theory 2

Objectives:	Content:
 Recall the formula of distance between two points and its slope Find the angle between two lines and derive the condition of perpendicularity and parallelism. Find the distance two parallel line. Find the area of triangle. Define quadratic equations and its roots. Define the nature of roots. Sub-unit 1.6: Co-ordinate Geometry (Equation of a pair of lines)	 Formula of distance between two points and its slope Angle between two lines and condition of perpendicularity and parallelism. Distance two parallel line. Area of triangle. Quadratic equations , its roots and nature of roots. Hrs. theory 6
Objectives:	Content:
 Define line pair equation, express two equation of straight lines as a single equation Find the condition required for equation of second degree (ax² + 2hxy + by² + 2gx +2fy + c = 0) to represent a pair of lines and find the separate equations. Prove that the equation ax² + 2hxy + by² = 0 always represents a pair of straight lines passing through the origin. Find the angle between two straight lines represented by the homogeneous equations of second degree. (ax² + 2hxy + by² = 0) 	 Line pair equation, Two equation of straight lines as a single equation Condition required for equation of second degree (ax² + 2hxy + by² + 2gx +2fy + c = 0) to represent a pair of lines and also find the separate equations. Proof that the equation ax² + 2hxy + by² = 0 always represents a pair of straight lines passing through the origin. The angle between two straight lines represented by the homogeneous equations of second degree. (ax² + 2hxy + by² = 0)
Sub-unit 1.7: Limits and limiting values	Hrs. theory 6
Objectives:	Content:
 Define the term <i>limit</i> and <i>limiting</i> value. Evaluate the limiting values of simple algebraic & trigonometric function. 	 <i>Limit</i> and <i>limiting</i> value. Limiting values of simple algebraic & trigonometric function.
• Use the formula • $\lim_{x \to a} \frac{x^n - a^n}{x - a} = na^{n-1}$	• Using the formula $\lim_{x \to a} \frac{x^n - a^n}{x - a} = na^{n-1}$
• $\lim_{\theta \to 0} \frac{\sin \theta}{\theta} = 1$ Define continuity and identify continous and discontinous function	$x \rightarrow a$ $x - a$ $\lim_{\theta \rightarrow 0} \frac{\sin \theta}{\theta} = 1$ Continuity and identification of continous and discontinous function
Sub-unit 1.8: Derivatives and their Applications	Hrs. theory 6
(Maxima and Minima)	
Objectives:	Content:
• Define the term derivatives.	• Definition of the term derivatives.
 Apply definition to get derivatives of the functions xⁿ, (ax + b)ⁿ, sin (ax + b), cos (ax + b), e^x and log_ax. 	Geometrical meaning of derivatives.

 Use the sum, difference, product, quotient and chain rule of derivates to calculate the derivates of algebric function only. Apply derivate to calculate maximum and minimum values of a given algebric function and other related problems. 	 Application of definition to get derivatives of the functions xⁿ (ax + b)ⁿ, sin (ax + b), cos (ax + b), e^x and log_ax. Using the sum, difference, product, quotient and chain rule of derivates to calculate the derivates of algebric function only. Application of derivate : increasing , decreasing and stationary points. Maximum , minimum values of a given algebric function and point of inflection. concave upward and concave downward (algebraic only)
Sub-unit 1.9: Integration	
Objectives:	Content:
 Define integral as anti-derivative, Apply techniques of integration as anti-derivative, substitution method, integration by parts and definite integral. Use definite integral to calculate area enclosed by algebric curve, X-axis and ordinate at x = a to x = b 	 Definition of integral as anti-derivative, Application of techniques of integration as anti-derivative, substitution method, integration by parts and definite integral (algebric only). Using definite integral to calculate area enclosed by algebric curve, X-axis and ordinate at x = a to x = b
Sub-unit 1.10: Probability	Hrs. theory 6
Objectives:	Content:
 Define probability (classical and empirical) Application and use addition and multiplication the law of probability Explain and use binomial probability distribution formula P(r) = c(n, r) p^r q^{n-r.} 	 Definition of probability (classical and empirical) Application and use of addition and multiplication law of probability Explanation and use of binomial probability distribution formula P(r) = c(n, r) p^r q^{n-r.}

Part B: Elementary Statistics

Unit 2: Elementary Statistics	Hrs. theory 46
Sub-unit 2.1: Introduction to Statistics (Revision only)	Hrs. theory 3
Objectives:	Content:
 Define statistics as given by different writers (Prof. Horace Secrist, Prof. Croxton & Crowden and Prof. Ya-Lu-Chan). State the utility, functions and limitations of statistics. Sub-unit 2.2: Collection, Classification and Tabulation diagrams and graphs (Revision only) 	 Definitions by Prof. Horace Secrist, Prof. Croxton & Crowden and Prof. Ya-Lu-Chan). Utility, functions and limitation of statistics. Hrs. theory 3
Objectives:	Content:
 Collect data (primary and secondary) Classify and tabulate data Prepare frequency table (ungrouped and grouped form) 	 Data Collection (primary and secondary) Classification and tabulation of data Preparation of a frequency table (ungrouped and grouped form)

 Represent data on simple, multiple, sub-divided, percentage bar diagram and Pie-diagrams. Represent data on histogram, frequency polygon, frequency curve and Ogive curve Sub-unit 2.3: Central tendency	 Representation of data on simple, multiple, sub-divided, percentage bar diagram and Pie-diagrams. Representation of data on histogram, frequency polygon, frequency curve and Ogive curve Hrs. theory 5
Objectives:	Content:
Define central tendency	Definition of central tendency
• Calculate mean, median, mode, and partition values (Quartiles, Deciles and Percentiles) for ungrouped and grouped data mathematically	• Calculation of mean, median, mode, and partition values (Quartiles, Deciles and Percentiles) for ungrouped and grouped data mathematically
Sub-unit 2.4: Measure of dispersion	Hrs. theory 8
Objectives:	Content:
 Calculate range, quartile deviation and standard deviation for ungrouped and grouped data mathematically Concept of absolute and relative measures of dispersion Compute coefficient of range, quartile deviation, and variation for ungrouped and grouped data mathematically 	 Calculation of range, quartile deviation mean deviation and standard deviation for ungrouped and grouped data mathematically Absolute and relative measures of dispersion Computation of coefficient of range, quartile deviation, mean deviation, and variation for ungrouped and grouped data mathematically
Sub-unit 2.5: Correlation Coefficient	Hrs. theory 9
Objectives:	Content:
 Define the concept of correlation. Define correlation method by drawing Scatter diagram Explain Karl Pearson's coefficient of correlation between two variables. Define Sparman's rank correlation Define Probable error , standard error and test of significant of correlation 	 Concept of correlation. Method of studying correlation by drawing Scatter diagram Calculations of Karl Pearson's coefficient of correlation between two variables. Sparman's rank correlation. Probable error , standard error and test of significant of correlation.
Sub-unit 2.6: Vital statistics	Hrs. theory 10
Objectives:	Content:
 a) Vital Statistics Define the term vital statistics. Describe the utility of vital statistics. Identify the different sources of vital statistics. b) Measure of Fertility Define the meaning of Fertility Describe different measures of fertility Compute different indicators related to fertility 	 a) Vital Statistics Definition of the term vital statistics. Utility of vital statistics. Different sources of vital statistics. b) Measure of Fertility Meaning of Fertility Different measures of fertility Different indicators related to fertility (crude birth rate, specific fertility rate, General Fertility rate, total fertility rate)
c) Measures of mortality	c) Measures of mortality
State the meaning of mortality	Meaning of mortality
 Describe different measures of mortality 	Different measures of mortality

Compute different indicators related to mortality	• Different indicators related to mortality (crude death rate, specific death rate, infant mortality rate, maternal mortality rate and still birth rate)
d) Measures of morbidity (sickness)	d) Measures of morbidity (sickness)
• State the meaning of morbidity	Meaning of morbidity
Describe different measures of morbidity	Different measures of morbidity
• Compute the incidence rate and prevalence rate	• The incidence rate and prevalence rate
Sub-unit 2.7: Research methodology	Hrs. theory 4
Objectives:	Content:
• Define the concept of research.	Definition of research
• Describe the process and methodology of research by	Research methodology.
stepwise scientific method or formula application.	• Steps of research.
• Discuss the importance of interpreting research results	• Scientific method.
	• Statistical tools for measuring reliability of
	results.
	• Interpreting and understanding research data.
	• Applications of research in medical science.
Sub-unit 2.8: Introduction to Report Writing	Hrs. theory 4
Objectives:	Content:
• Explain the concept of report writing.	Purposes and goals of research reports.
• Familiarize with standard research report format.	Significance of research reporting
-	

Unit 3: Basic Computer Skills	Hrs. theory 8 Hrs. lab.
Sub-unit 3.1: Introduction	
Objectives:	Content:
 Describe the functions and uses of computers. Compare and contrast the functions of computer hardware. Describe and demonstrate the functions of computer hardware. Describe and demonstrate the functions of computer memory and storage systems. Demonstrate the procedure for scanning the viruses and removing viruses. Use a virus protection utility to recover damaged files in a diskette or hard disk. 	 Definitions and descriptions of computers and computing activities. Characteristics of various types and generations of computers. Computer hardware: CPU, VDU, Input and Output peripherals. Computer software: systems, applications, and utility software. Memory: RAM, ROM; storage systems (magnetic, optical), storage types (floppy, hard disk, compact disk), and storage capacities. Utilities for virus protection. Operation of virus utilities.

Statistics and Computer

The paper on elementary statistics is designed to supplement the theoretical knowledge. In this subject the students themselves will solve different problems with different types of data and information, which helps them to learn the subject quickly and enjoy the real, need of learning it

and apply their knowledge in real life situations, for an effective health care delivery and administration.

Students will be required to maintain a note book to keep the records of fully practical work duly signed by the instructor which should contain a minimum of practical and this should be submitted on the date of examination.

All the portion of Computer skill should be done in a practical room having individual computer to practice. Here first 1 hour should be devoted for theory description and procedure and second 1 hour should be for practice in the computer. It is better to have statistics practical done in computer as far as practicable.

Statistics:

40 hours

Prepare individual (discrete) and grouped frequency distribution table. Prepare histogram, frequency polygon and curve, and cumulative frequency curve. Draw bar diagram, subdivided, percentage and multiple bar diagram Draw pie diagram. Find mean for individual and grouped series. Find mean by shortcut or graphical method. Find quartiles, deciles and percentiles mathematically and graphically. Find the mean deviation from mean, median and mode. Find standard deviation. Find standard deviation through shortcut method. Find the coefficient of variation. Compute prevalence rate of morbidity. Compute incidence rate of morbidity. Calculate crude and specific death rate. Calculate infant mortality rate. Calculate maternal mortality rate. Calculate fertility of crude, specific and general birth rate Calculate fertility rate of natural increment. 40 hours

Computer:

- Create a directory and file on hard disk under MS-DOS and WINDOW system.
- Open window environment and change the outlook of window.
- Open MS-WORD and create a document explaining statistical methods like function, limitations, graph, diagram and table construction.
- Open MS-EXCEL and create a frequency table and apply formula to calculate mean, median, mode, and standard deviation.
- Create graph and diagrams from MS-EXCEL.
- Open MS-ACCESS and create a database for report writing.
- Operate POWER POINT and prepare presentation.
- Operate VIRUS scanning on hard and removable disks.

(Note: The statistical tools should be introduced from an applied perspective using health related examples. Microsoft excel software will be used throughout the course to aid in statistical analysis)